



1	Database of nitrification and nitrifiers in the global ocean
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35 Abstract

36 As a key biogeochemical pathway in the marine nitrogen cycle, nitrification (ammonia oxidation 37 and nitrite oxidation) converts the most reduced form of nitrogen – ammonium/ammonia (NH4^{+/} NH_3) into the oxidized species nitrite (NO_2^{-1}) and nitrate (NO_3^{-1}). In the ocean, these processes are 38 39 mainly performed by ammonia-oxidizing archaea (AOA) and bacteria (AOB), and nitriteoxidizing bacteria (NOB). By transforming nitrogen speciation and providing substrates for 40 nitrogen removal, nitrification affects microbial community structure, marine productivity 41 42 (including chemoautotrophic carbon fixation) and the production of a powerful greenhouse gas, 43 nitrous oxide (N_2O) . Nitrification is hypothesized to be regulated by temperature, oxygen, light, substrate concentration, substrate flux, pH, and other environmental factors. Although the number 44 45 of field observations from various oceanic regions has increased considerably over the last few 46 decades, a global synthesis is lacking, and understanding how environmental factors control nitrification remains elusive. Therefore, we have compiled a database of nitrification rates and 47 nitrifier abundance in the global ocean from published literature and unpublished datasets. This 48 database includes 2393 and 1006 measurements of ammonia oxidation and nitrite oxidation rates, 49 50 and 2187 and 631 quantifications of ammonia oxidizers and nitrite oxidizers, respectively. This community effort confirms and enhances our understanding of the spatial distribution of 51 52 nitrification and nitrifiers, and their corresponding drivers such as the important role of substrate 53 concentration in controlling nitrification rates and nitrifier abundance. Some conundrums are also 54 revealed including the inconsistent observations of light limitation and high rates of nitrite 55 oxidation reported from anoxic waters. This database can be used to constrain the distribution of marine nitrification, to evaluate and improve biogeochemical models of nitrification, and to 56 quantify the impact of nitrification on ecosystem functions like marine productivity and N₂O 57 production. This database additionally sets a baseline for comparison with future observations and 58 guides future exploration (e.g., measurements in the poorly sampled regions such as the Indian 59 60 Ocean; method comparison/standardization). The database is publicly available at Zenodo 61 repository: https://doi.org/10.5281/zenodo.7942922 (Tang et al., 2023).



63 Introduction

- 64 Nitrification (ammonia oxidation and nitrite oxidation) converts the most reduced form of nitrogen 65 (N) – ammonium/ammonia (NH_4^+/NH_3) into the oxidized compounds nitrite (NO_2^-) and nitrate (NO₃⁻). Ammonia oxidation is conducted by ammonia oxidizing archaea (AOA) and bacteria 66 67 (AOB) with AOA dominating in most marine environments (Francis et al., 2005; Wuchter et al., 68 2006). Marine AOA are often separated into a few major ecotype groups including water column 69 group A, water column group B and Nitrosopumilus-like (Beman et al., 2008; Tolar et al., 2020), 70 with a diverse goup of AOA remaining to be characterized (Alves et al., 2018). Marine nitrite 71 oxidation is carried out by nitrite-oxidizing bacteria (NOB) such as Nitrospina, Nitrospira, 72 Nitrococcus and Nitrobacter, with Nitrospina as the dominant group (Mincer et al., 2007; 73 Pachiadaki et al., 2017). Complete ammonia-oxidizing (comammox) bacteria within the bacterial 74 genus Nitrospira have been identified in freshwater, terrestrial, and coastal environments but not 75 yet been found in the open ocean (Daims et al., 2015; Van Kessel et al., 2015; Xia et al., 2018).
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77 Nitrification and nitrifiers are thought to be regulated by light/solar radiation, oxygen, temperature, 78 substrate concentration, pH, and other environmental factors (Ward, 2008), many of which are 79 experiencing dramatic changes in the ocean. For example, light is generally found to inhibit 80 nitrifiers' growth and nitrification rate (Olson. 1981b; Merbt et al., 2012; Xu et al., 2019). In 81 addition, ocean acidification decreases ammonia oxidation rates (Beman et al., 2011; Breider et 82 al., 2019) partly due to the decreased availability at lower pH of NH₃, which is the actual substrate 83 for ammonia oxidation (Suzuki et al., 1974). In contrast, ocean warming shifts the NH₄⁺/NH₃ equilibrium towards NH₃ by decreasing the pK_a (Emerson et al., 1975) and is observed to enhance 84 85 enzyme activity (Zheng et al., 2017; Zheng et al., 2020), further complicating the effect of climate change on nitrification. 86

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Although nitrification does not directly change the absolute inventory of bioavailable N, it can
control the relative availability of substrates (NH₄⁺, NO₂⁻ and NO₃⁻) for phytoplankton growth.
Since prokaryotic phytoplankton preferentially assimilate NH₄⁺ while eukaryotic phytoplankton
are better able to exploit NO₃⁻ in the sunlit surface ocean (Berthelot et al., 2018; Fawcett et al.,
2011), variations in the relative supply of NH₄⁺ versus NO₃⁻ can influence phytoplankton
community composition and ecosystem functionalities. Because the uptakes of NH₄⁺ and NO₃⁻ are





- 94 often used to differentiate regenerated and new production (Eppley and Peterson. 1979), 95 production of NO₃⁻ by nitrification in the surface ocean may bias the estimate of new production (Yool et al., 2007). NO₂⁻ and NO₃⁻ are also involved in denitrification and anammox, which remove 96 97 bioavailable N from the ocean. Thus, nitrification can indirectly affect the size of the bioavailable N pool, marine productivity and ultimately the atmospheric CO₂ concentration (Falkowski, 1997). 98 99 As a chemoautotrophic process, nitrification in the ocean water column is estimated to supply ~0.13-1.4 Pg C yr⁻¹ of organic matter, which is critical to support the heterotrophic microbial 100 101 community/metabolism in the dark ocean (Bayer et al., 2022; Middelburg, 2011; Pachiadaki et al., 102 2017; Zhang et al., 2020). Nitrification could also contribute to the oxygen consumption and the development of hypoxia or anoxia (Hsiao et al., 2014; Beman et al., 2021). In addition, nitrification 103 104 is the major global ocean source of N_2O , a potent greenhouse gas and dominant ozone-depleting agent, thus connecting the marine N cycle directly to the Earth's climate system (Freing et al., 105 2012; Ji et al., 2018). 106
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Considering the important role of nitrification and nitrifiers in marine N and C cycles and Earth's 108 109 climate, a better understanding of its distribution and regulating factors is highly desirable. Historical observations of nitrification and nitrifiers cover a wide range of environmental gradients 110 111 and biogeography in the ocean, ranging from cross-Atlantic (e.g., Clark et al., 2008; Clark et al., 2022), western Pacific (e.g., Wan et al., 2021; Wan et al., 2018), polar oceans (e.g., Shiozaki et 112 al., 2019; Mdutyana et al., 2020) to oxygen minimum zones (e.g., Peng et al., 2015; Santoro et al., 113 114 2021). This study aims to introduce the newly constructed database of nitrification and nitrifiers 115 in the marine water column and to guide future research efforts in field observations and model development of nitrification. This new global synthesis significantly expands upon what was 116 possible with earlier more limited datasets (Yool et al. 2007; Ward. 2008). Additional reviews on 117 118 marine nitrification and nitrifiers can be found elsewhere (Schleper and Nicol, 2010; Daims et al., 119 2016; Ward, 2011b).





121 Methods

122 Data sources and compilation

123 Nitrification rates including ammonia oxidation and nitrite oxidation, and abundances of ammonia 124 oxidizers and nitrite oxidizers were extracted directly from the literature published between 1984 and 2022 when the data were presented in tables or supplementary materials from publications; 125 otherwise, data were provided by the coauthors. Some previously unpublished data were also 126 included in the database. Table 1 and Table 2 summarize the origin, methods and locations of 127 128 nitrification rate and nitrifier abundance measurements, sorted in alphabetical order by lead author. 129 The metadata format contains geographical sampling information (date, latitude, longitude, and 130 depth) and concurrent measurements of environmental conditions such as light intensity, temperature, salinity, water density, N concentration (NH4⁺, NO₂⁻ and NO₃⁻), pH and oxygen 131 132 concentration if available. In total, there are 2393, 1006, 2187, and 631 measurements of ammonia oxidation rate, nitrite oxidation rate, ammonia oxidizer abundance and nitrite oxidizer abundance, 133 respectively. However, not all measurements of nitrification rates or nitrifier abundance are 134 accompanied by all the environmental factors because such factors were often not reported in the 135 136 literature or recorded during the measurements/sample collections. Rates, nitrifier abundances and environmental parameters below the methodological detection limits are noted as BDL. NM 137 represents parameters that were not measured. Empty/NA means that data are not available or 138 139 reported. The database is deposited into Zenodo repository following the Findable, Accessible, 140 Interoperable and Reusable (FAIR) principles for data management (Wilkinson et al., 2016). We 141 encourage authors and readers to contact us to report an update to or an error in the database.

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Table 1. Summary of the number of observations for nitrification rates in alphabetical order of the
lead author. The method (e.g., substrate tracer addition vs product dilution) and sampling regions

are listed. Methods used for data collection are described in the next section.

References			Nitrification					
							regions	
	Ammonia	Method		Analyte	Nitrite	Method		
	oxidation				oxidation			
Baer et al., 2017	6	¹⁵ NH ₄ ⁺ t	tracer	NO ₂ ⁻ +			Western Coastal	
		addition		NO ₃ -			Arctic	





Beman et al.,	68	¹⁵ NH4 ⁺	tracer	NO ₂ +	64	¹⁵ NO ₂ ⁻ tracer	Eastern Tropical
2012	00	addition		NO ₃ -	0.	addition	North Pacific
Beman et al.,	78	¹⁵ NH4 ⁺	tracer	-	79	¹⁵ NO ₂ ⁻ tracer	Eastern Tropical
2021		addition		NO ₃ -		addition	North Pacific
Bianchi et al.,	21	H ¹⁴ CO ₃ -	tracer	Particulate	21	H ¹⁴ CO ₃ -	Southern Ocean
1997		addition		organic		tracer	
				carbon		addition	
Breider et al.,	10	¹⁵ NH4 ⁺	tracer	NO2 ⁻ +			Western North
2019		addition		NO ₃ -			Pacific
Bristow et al.,	9	¹⁵ NH4 ⁺	tracer	NO ₂ -	9	¹⁵ NO ₂ ⁻ tracer	Gulf of Mexico
2015		addition				addition	
Cavagna et al.,					39	¹⁵ NO ₃ ⁻ tracer	Southern Ocean
2015						dilution	
Clark et al., 2008	32	¹⁵ NO ₂ -	tracer	NO ₂ -	32	¹⁵ NO ₃ ⁻ tracer	Atlantic
		dilution				dilution	
Clark et al., 2011	13	¹⁵ NO ₂ -	tracer	NO ₂ -	13	¹⁵ NO ₃ ⁻ tracer	Eastern North
		dilution				dilution	Atlantic
							(offshore of the
							Iberian
							Peninsula)
Clark et al., 2014	10	¹⁵ NO ₂ -	tracer	NO ₂ -	10	¹⁵ NO ₃ ⁻ tracer	Northwest
		dilution				dilution	European shelf
							sea
Clark et al., 2016	21	¹⁵ NO ₂ -	tracer	NO ₂ -	42	¹⁵ NO ₃ ⁻ tracer	Mauritanian
		dilution				dilution	upwelling
							system
Clark et al., 2022	88	¹⁵ NO ₂ ⁻	tracer	NO ₂ -			Atlantic
		dilution					
Clark et al.,	18	¹⁵ NO ₂ -	tracer	NO ₂ -	18	¹⁵ NO ₃ ⁻ tracer	Eastern North
unpublished 1		dilution				dilution	Atlantic
Clark et al.,	18	¹⁵ NO ₂ ⁻	tracer	NO ₂ -	18	¹⁵ NO ₃ ⁻ tracer	Eastern North
unpublished 2		dilution				dilution	Atlantic
Clark et al.,	21	¹⁵ NO ₂ -	tracer	NO ₂ -	21	¹⁵ NO ₃ ⁻ tracer	Eastern North
unpublished 3		dilution				dilution	Atlantic
	11	¹⁵ NO ₂ -	tracer	NO ₂ -	11	¹⁵ NO ₃ ⁻ tracer	Subpolar North
Clark et al.,	11	1102	tracer	1102		1105 114001	
Clark et al., unpublished 4	11	dilution	tiucei	1102		dilution	Atlantic and





Damashek et al.,	15	¹⁵ NH4 ⁺	tracer	NO ₂ ⁻ +			South Atlantic
2018	15	addition	tracer	$NO_2 + NO_3^-$			
	20			-			Bight
Diaz and	20	¹⁵ NH4 ⁺	tracer				Gulf of Lions in
Raimbault, 2000		addition		NO ₃ -			the
							Mediterranean
							Sea
Dore and Karl,	11	NO ₂ ⁻ +	NO ₃ -	NO ₂ +	6	NO ₃ -	Station ALOHA
1996		concentratio	on	NO ₃ -,		concentration	in the North
		change over	r time;	particulate		change over	Pacific
		$\mathrm{H^{14}CO_{3}}^{-}$	tracer	organic		time	
		addition		carbon			
Fernández et al.,	15	¹⁵ NH4 ⁺	tracer	NO2 ⁻ +			Peru upwelling
2009		addition		NO ₃ -			system
Flynn et al., 2021					104	¹⁵ NO ₂ ⁻ tracer	Weddell Sea
						addition	
Frey et al., 2020	21	¹⁵ NH4 ⁺	tracer	NO ₂ -			Eastern Tropical
-		addition					South Pacific
Frey et al., 2022	30	¹⁵ NH4 ⁺	tracer	NO ₂ -			Eastern Tropical
		addition					North Pacific
Ganesh et al.,	5	¹⁵ NH4 ⁺	tracer	NO ₂ -	5	¹⁵ NO ₂ ⁻ tracer	Eastern Tropical
2015		addition				addition	North Pacific
							oxygen
							minimum zone
Kalvelage et al.,	6	¹⁵ NH4 ⁺	tracer	NO ₂ -			Namibian
2011	0	addition	tracer	1102			oxygen
2011		uddition					minimum zone
Kalvelage et al.,	108	¹⁵ NH4 ⁺	tracer	NO ₂ -	110	¹⁵ NO ₂ ⁻ tracer	Eastern Tropical
2013	108	addition	uacei	1102	110	addition	South Pacific
2013		addition				addition	
							oxygen
IZ	0	155 111 +		NO		1510 - 1	minimum zone
Kitzinger et al.,	9	¹⁵ NH ₄ ⁺	tracer	NO ₂ -	9	¹⁵ NO ₂ tracer	Gulf of Mexico
2020		addition				addition	
Lam et al., 2009	14	¹⁵ NH4 ⁺	tracer	NO ₂ -			Eastern Tropical
		addition					South Pacific
Laperriere et al.,	59	¹⁵ NH4 ⁺	tracer	NO ₂ +			Southern





Liu et al., 2018	86	¹⁵ NH4 ⁺	tracer	NO ₂ ⁻ +			South Atlantic
		addition		NO ₃ -			Bight
Liu et al., 2022	10	¹⁵ NH4 ⁺	tracer	NO ₂ ⁻ +			South China Sea
		addition		NO ₃ -			
Mccarthy et al.,	8	$^{15}\mathrm{NH4}^{+}$	tracer	NO ₂ -+			Arabian Sea
1999		addition		NO ₃ -			
Mdutyana et al.,	59	¹⁵ NH4 ⁺	tracer	NO ₂ -	38	¹⁵ NO ₂ ⁻ tracer	Southern Ocean
2020		addition				addition	
Mdutyana et al.,	24	$^{15}\mathrm{NH_{4}^{+}}$	tracer	NO ₂ -			Southern Ocean
2022a		addition					
Mdutyana et al.,					24	¹⁵ NO ₂ ⁻ tracer	Southern Ocean
2022b						addition	
Newell et al.,	8	¹⁵ NH4 ⁺	tracer	NO ₂ -			Sargasso Sea
2013		addition					(western North
							Pacific)
Peng et al., 2015	30	$^{15}\mathrm{NH_{4}^{+}}$	tracer	NO_2^- ,	30	¹⁵ NO ₂ ⁻ tracer	Eastern Tropical
		addition		NO_2^-+		addition	North Pacific
				NO ₃ -			
Peng et al., 2016	47	¹⁵ NH4 ⁺	tracer	NO ₂ -	47	¹⁵ NO ₂ ⁻ tracer	Eastern Tropical
		addition				addition	South Pacific
Peng et al., 2018	28	¹⁵ NH4 ⁺	tracer	NO ₂ -	28	¹⁵ NO ₂ ⁻ tracer	Subarctic North
		addition				addition	Atlantic
Raes et al., 2020	39	¹⁵ NH4 ⁺	tracer	NO ₂ ⁻ +			South Pacific
		addition		NO ₃ -			
Raimbault et al.,	41	¹⁵ NH4 ⁺	tracer	NO ₂ ⁻ +			Equatorial
1999		addition		NO ₃ -			Pacific
Santoro et al.,	11	$^{15}\mathrm{NH_{4}^{+}}$	tracer	NO ₂ -+			Central
2010		addition		NO ₃ -			California
							Current
Santoro et al.,	10	¹⁵ NH4 ⁺	tracer	NO ₂ -,			Central
2013		addition		NO ₂ ⁻ +			California
				NO ₃ -			Current
Santoro et al.,	12	¹⁵ NH4 ⁺	tracer	NO ₂ ⁻ +			Equatorial
2017		addition		NO ₃ -			Pacific
Contons of al		153 77 7 +		NO -1		15110	Eastern Tarriel
Santoro et al.,	57	¹⁵ NH4 ⁺	tracer	NO ₂ -+	57	¹⁵ NO ₂ ⁻ tracer	Eastern Tropical





Sinyanya et al.,				31	¹⁵ NO ₂ ⁻ tracer	Southwest Indian
unpublished					addition	Ocean
Shiozaki et al.,	87	¹⁵ NH ₄ ⁺ tracer	NO ₂ -+			Equatorial
2016		addition	NO ₃ -			Pacific to the
						Arctic Ocean
Shiozaki et al.,	56	¹⁵ NH ₄ ⁺ tracer	NO ₂ +			Arctic Ocean
2019		addition	NO ₃ -			
Shiozaki et al.,	28	¹⁵ NH ₄ ⁺ tracer	NO ₂ +			Arctic Ocean
2021		addition	NO ₃ -			
Smith et al., 2022	11	¹⁵ NH ₄ ⁺ tracer	NO ₂ -			Southern Ocean
		addition				
Sun et al., 2017				9	¹⁵ NO ₂ ⁻ tracer	Eastern Tropical
					addition	North Pacific
Sutka et al., 2004	20	¹⁵ NH ₄ ⁺ tracer	NO ₂ -+			North Pacific
		addition	NO3 ⁻			Subtropical Gyre
						to Eastern
						Tropical North
						Pacific
Tolar et al., 2016	73	¹⁵ NH ₄ ⁺ tracer	NO ₂ +			Antarctic coast
		addition	NO ₃ -			
Tolar et al., 2017	38	¹⁵ NH ₄ ⁺ tracer	NO ₂ +			Georgia coast,
		addition	NO ₃ -			South Atlantic
						Bight, Gulf of
						Alaska,
						Antarctic coast
Tolar et al., 2020	297	¹⁵ NH ₄ ⁺ tracer	NO ₂ +			Monterey Bay
		addition	NO ₃ -			
Wallschuss et al.,	40	¹⁵ NH ₄ ⁺ tracer	NO ₂ ⁻	40	¹⁵ NO ₂ ⁻ tracer	Southeastern
2022		addition			addition	Atlantic
Wan et al., 2018	90	¹⁵ NH ₄ ⁺ tracer	NO ₂ ⁻ +			South China Sea
		addition	NO ₃ -			and Northwest
						Pacific
Wan et al., 2021	17	¹⁵ NH ₄ ⁺ tracer	NO ₂ -	17	¹⁵ NO ₂ ⁻ tracer	North Pacific
		addition			addition	
Wan et al., 2022	85	¹⁵ NH ₄ ⁺ tracer	NO ₂ -			North Pacific
		addition				
	l			1	1	1





Ward et al., 1984	16	¹⁵ NH4 ⁺ addition	tracer	NO ₂ ⁻			Coastal waters off Washington
Ward, 1987	24	¹⁵ NH4 ⁺ addition	tracer	NO ₂ -		¹⁵ NO ₂ ⁻ tracer addition	Southern California Bight
Ward and Zafiriou, 1988	42	¹⁵ NH4 ⁺ addition	tracer	NO ₂ -			Eastern Tropical North Pacific
Ward et al., 1989	47	¹⁵ NH4 ⁺ addition	tracer	NO ₂ -	47	¹⁵ NO ₂ ⁻ tracer addition	Eastern Tropical South Pacific
Ward, 2005	110	¹⁵ NH4 ⁺ addition	tracer	NO ₂ -			Monterey Bay
Xu et al., 2018	78	¹⁵ NH4 ⁺ addition	tracer	NO ₂ -			South China Sea
Zhang et al., 2020	27	¹⁵ NH4 ⁺ addition	tracer	NO ₂ ⁻	27	¹⁵ NO ₂ ⁻ tracer addition	South China Sea and Western Pacific
Total number of observations	2393				1006		

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Table 2. Summary of the number of observations for nitrifier abundance from qPCR assays in
alphabetical order of the lead authors. The top row indicates the gene quantified for each group
(see text for further details). The primers used for individual studies are identified in the database.
AOA: ammonia-oxidizing archaea; AOB: ammonia-oxidizing bacteria; NOB: nitrite-oxidizing
bacteria.

References	amoA-	-based	nxr-	16S i	rRNA-based		Sampling
			based				regions
	AOA	AOB	NOB	Thaumarchaeota	Nitrospira	Nitrospina	
Agogue et al., 2008	55	55		55			North Atlantic
Beman et al., 2012	64	64		64			Eastern Tropical North Pacific
Beman et al., 2013						63	Eastern Tropical North Pacific
Bristow et al., 2016b	27		27				Bay of Bengal oxygen minimum zone





Damashek et al.,	34		34	34	South Atlantic
2018.					Bight
Frey et al., 2020	21				South Pacific
					oxygen
					minimum zone
Frey et al., 2022	30				North Pacific
					oxygen
					minimum zone
Horak et al., 2018	6	6			North Pacific
					Ocean
Kalvelage et al.,	143	89			South Pacific
2013.					oxygen
					minimum zone
Liu et al., 2018.	385	385	385	385	South Atlantic
					Bight
Peng et al., 2013	23				Eastern Tropical
					North Pacific
Peng et a., 2015	19	19			Eastern Tropical
					South Pacific
Peng et a., 2016	19	19			Subarctic North
					Atlantic
Santoro et al., 2010	17	17	17	17	Central
					California
					Current
Santoro et al., 2013	10	10			Central
					California
					Current
Santoro et al., 2017	148				Equatorial
					Pacific
Santoro et al., 2021	78	24	78	78	Eastern Tropical
					South Pacific
Shiozaki et al., 2016	87	87		1	North Pacific
Shiozaki et al., 2019	56	56		1	Arctic Ocean
Shiozaki et al., 2021	28	28			Arctic Ocean
Sintes et al., 2013	115		115		Tropical Atlantic
					and coastal





Sintes et al., 2016	364			364			Atlantic Ocean
Tolar et al., 2016	73	73					Antarctic coast
Tolar et al., 2017	38			38			Georgia coast, South Atlantic Bight, Gulf of Alaska, Antarctic coast
Tolar et al., 2020	297						Monterey Bay
Wuchter et al., 2006	20	20		20			Atlantic Ocean
Zakem et al., 2018	31						North Pacific
Zhang et al., 2020	54	54		54	54	54	South China Sea and Western Pacific
Total points	2187	951	27	1169	54	631	

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We applied Chauvenet's criterion for quality control to flag outliers in nitrification rates and 153 nitrifier abundance (Glover et al., 2011). Chauvenet's criterion is commonly applied to normally 154 distributed datasets to identify outliers whose deviations from the mean have a probability of less 155 than 1/(2n), where n is the number of data points (Buitenhuis et al., 2013). We applied the criterion 156 acknowledging the fact that the data were collected at different environmental conditions. After 157 158 removing measurements of zero and below detection limit, nitrification rates and nitrifier 159 abundances were log10 transformed before further analysis. Nitrification rates and nitrifier abundances reported at 0 or below detection limit are noted separately in the database and 160 161 following analysis. Although we did not find outliers for ammonia oxidation and nitrite oxidation 162 rates, there are some extreme values worth noting. For example, an extremely high ammonia oxidation rate of 4900 nmol L⁻¹ d⁻¹ was observed in the Peruvian oxygen minimum zone (Lam et 163 al., 2009). Low but detectable rates below 0.01 nmol L⁻¹ d⁻¹ were observed in the Eastern Tropical 164 165 North Pacific oxygen minimum zone (Frey et al., 2022), South Atlantic Bight (Liu et al., 2018) and western Pacific (Xu et al., 2018). Some outliers were identified by Chauvenet's criterion for 166 ammonia oxidizers. For instance, an abnormally high abundance of the bacterial *amoA* gene (10^8) 167 copies L⁻¹) was observed in the South Pacific oxygen minimum zone (Kalvelage et al., 2013), 168 169 which was removed from the following analysis. A low abundance of 16S rRNA of Thaumarchaeota (25 copies L^{-1}) was found in the surface water of the western Pacific (Zhang et 170



al., 2020). In addition, the low-ammonia concentration AOA ecotype (or water column group B
AOA) at 2 copies L⁻¹ was reported in the Arctic Ocean (Sintes et al., 2013). Measurements of
nitrification rate and nitrifier abundance of 0 or below detection limit were not included in the
analysis of outlier identification. For example, AOA abundance at 0 or below detection limit
(varies among studies) has been reported in surface waters of South Atlantic Bight (Damashek et
al., 2018), equatorial Pacific (Santoro et al., 2017) and North Pacific (Shiozaki et al., 2016).

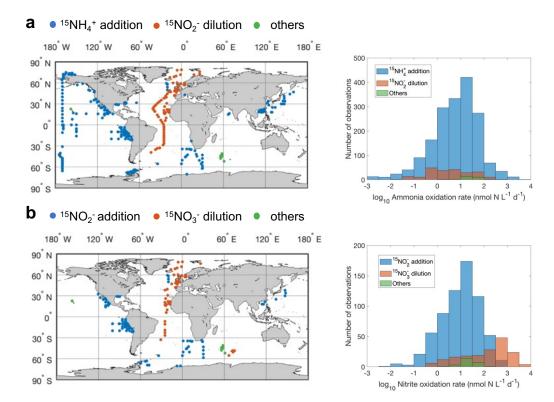
177

178 Methods for measuring ammonia oxidation and nitrite oxidation rates

179 Ammonia oxidation rate is commonly measured by comparing the change in nitrite (NO_2^{-}) and nitrate (NO₃⁻) concentration in controls versus an experimental treatment containing a nitrification 180 inhibitor (e.g., Dore and Karl, 1996), by tracking the oxidation of ¹⁵NH₄⁺ into the NO₂⁻ and NO₃⁻ 181 pool (Olson, 1981a), or by the dilution of ¹⁵NO₂⁻ (Clark et al., 2007). Similarly, nitrite oxidation 182 183 rate can be measured by the change in NO_3^- concentration, by tracking the oxidation of $^{15}NO_2^-$ into the NO₃⁻ pool, or by the dilution of ${}^{15}NO_{3}^{-}$ (Ward et al., 1989). In addition, nitrification has also 184 been estimated from the incorporation of ¹⁴C tracer due to the chemoautotrophic metabolism of 185 nitrifiers (Bianchi et al., 1997). There is a large uncertainty, however, in the conversion factor from 186 187 carbon fixation to nitrification (Bayer et al., 2022). A more detailed description of methods for measuring nitrification can be found in Ward, 2011a. The spatial distribution of different methods 188 189 used to measure nitrification and the frequency distribution of measured rates by different methods are shown in Figure 1. Rates measured with the substrate tracer addition method $(^{15}NH_4^+)$ and 190 $^{15}NO_2$) outnumbered other methods globally but the product dilution method ($^{15}NO_2$ and $^{15}NO_3$) 191 dominated in the Atlantic Ocean. The ammonia oxidation rates measured by different methods 192 have similar median values. However, the median nitrite oxidation rate measured by the ¹⁵NO₃-193 dilution method is significantly higher than the rate measured by the 15NO₂- addition method (200.3) 194 195 vs 7.4 nmol N L⁻¹ d⁻¹). These comparisons, however, are between samples aggregated from measurements taken at different sites. It is thus unclear whether the differences arise from 196 197 differences in the measurement approaches (e.g., in sensitivity) or in the sites where measurements were made. A direct methods comparison is recommended for future exploration. 198 199







200

Figure 1. Distribution of different methods used to measure ammonia oxidation (a) and nitrite
 oxidation (b). Others include ¹⁴C incorporation and concentration change methods.

203

Incubations to measure nitrification rates have been conducted in polycarbonate and glass bottles, 204 205 exetainers and plastic bags. Seawater is directly transferred from the Niskin bottle into the 206 incubation containers to minimize temperature, oxygen and other perturbations. These incubation 207 containers are usually kept in an incubator with light filters to mimic the ambient temperature and 208 light conditions. After incubating for 3 hours to over 24 hours depending on the estimated 209 magnitude of nitrification rates, the incubation is terminated by filtering via GF/F or $0.22 \,\mu m$ filters (e.g., Baer et al., 2017; Wan et al., 2019). The filtrate is then frozen at -20°C or -80°C until further 210 211 analysis on land. The incubation has also been terminated by subsampling and freezing without filtration (e.g., Damashek et al., 2018). Alternatively, the incubation is preserved by adding 212 213 mercury chloride or zinc chloride (Kalvelage et al., 2013; Frey et al., 2020). This method allows 214 gas measurements like N₂O and N₂ production before nitrification analysis.





216 Various approaches have been developed to measure the N isotopes of NO₂⁻ and NO₃⁻. For 217 example, 1) dissolved NO_{2⁻} is extracted by formation of an azo dye. The resulting dye is filtered onto precombusted GF/F or GF/C filters and its ¹⁵N:¹⁴N ratio is analyzed by elemental analyzer 218 isotope ratio mass spectrometry (Ward et al., 1982; Olson, 1981a). NO₃⁻ can be reduced to NO₂⁻ 219 220 by cadmium reduction and then extracted using the azo dye method described above. 2) Dissolved 221 NO_2^- is converted to Sudan-1 and Sudan-1 is collected via solid-phase extraction. The sample is 222 then purified by HPLC and derivatized before analysis by GC/MS (Clark et al., 2007). Similarly, 223 NO_3 can be reduced to NO_2 by cadmium prior to conversion to Sudan-1 for nitrogen isotope analysis. 3) NO_2^{-1} can be converted to N₂ with sulfamic acid and subsequently measured by isotope 224 225 ratio mass spectrometry (Dalsgaard et al., 2012; Bristow et al., 2016). 4) NO2⁻ can also be 226 converted into N₂O by the azide method and subsequently measured by isotope ratio mass spectrometry (Mcilvin and Altabet, 2005). The N isotopes of NO₂⁻ and NO₃⁻ can be measured via 227 the denitrifier method (Sigman et al., 2001; Weigand et al., 2016) where both NO₂⁻ and NO₃⁻ are 228 converted into N₂O. In addition, the δ^{15} N of NO₃⁻ alone can be measured using the denitrifier 229 method after removing NO2⁻ with sulfamic acid (Granger and Sigman, 2009). The azide and 230 231 denitrifier methods require smaller sample volumes and offer a higher sensitivity of nitrogen 232 isotope detection.

233

Many factors may complicate the interpretation of rate measurements, e.g., isotope dilution by 234 regeneration of the ¹⁵N-labeled substrates and stimulation of nitrification by substrate addition 235 (Lipschultz, 2008). For instance, the amount of tracer addition varied substantially from <10 nM 236 237 to 5 μ M, enriching the ambient pool by <10% to over 1000%. The excess addition of substrates will likely enhance the nitrification rate, which will then reflect a potential rate instead of an in-238 situ rate. In addition, the measurement of NO_2^- compared to $NO_2^- + NO_3^-$ could also lead to 239 variations in the estimates of the ammonia oxidation rates. Specifically, ¹⁵NO₂⁻ produced from 240 15 NH₄⁺ may be further oxidized to 15 NO₃⁻, especially when samples are low in NO₂⁻ concentration. 241 Ammonia oxidation rate may be underestimated if only ${}^{15}NO_2$ is measured instead of measuring 242 both ¹⁵NO₂⁻ and ¹⁵NO₃⁻ (Santoro et al., 2013; Peng et al., 2015). Therefore, NO₂⁻ carrier (to increase 243 244 the NO_2^{-} pool and trap the produced ${}^{15}NO_2^{-}$) may be added to the sample before incubation or both NO_2^- and NO_3^- should be measured after incubation when ambient NO_2^- concentration is low. The 245 15 NO₂⁻ isotope dilution method may overestimate ammonia oxidation rates because NO₂⁻ could 246



- also be released from phytoplankton after assimilative nitrate reduction (Lomas and Lipschultz,
 2006). These confounding factors may be difficult to quantify but worth recording and reporting
 in publications for the sake of comparison among studies. In addition, a variety of approaches have
 been applied to calculate nitrification rates. However, some methods correct for the impact of other
 processes such as the uptake of the substrates or products of nitrification on rate estimates (e.g.,
 Lipschultz et al., 1986; Santoro et al., 2010) while others do not (e.g., Dugdale and Goering.,
 1967).
- Nitrification supported by organic N substrates like urea and cyanate has been observed in the Gulf
 of Mexico (Kitzinger et al., 2018), Pacific (Santoro et al., 2017; Wan et al., 2021), off the east
 coast of the United States (Laperriere et al., 2020; Tolar et al., 2017), and in the polar oceans
 (Alonso-Saez et al., 2012; Shiozaki et al., 2021). The number of these observations remains limited
 compared to ammonia oxidation. They can be included in future editions of the database (i.e., not
 included in the current database) and their role in the marine N cycle deserves future investigations.
- 262 Methods for quantifying ammonia oxidizers and nitrite oxidizers

263 We summarize the primers used to quantify nitrifier abundance based on both functional genes and 16S rRNA genes using quantitative PCR (qPCR) (Table 3). The cell abundance and biomass 264 265 can be subsequently estimated based on the gene abundance, number of genes per cell and specific cell biomass (e.g., Kitzinger et al., 2020; Khachikyan et al., 2019). The oxidation of ammonia to 266 267 hydroxylamine is catalyzed by ammonia monooxygenase, which is partly encoded by the amoA 268 gene. Primers have been designed to quantify both bacterial and archaeal amoA genes (Rotthauwe et al., 1997; Francis et al., 2005; Hornek et al., 2006; Wuchter et al., 2006; Beman et al., 2008; 269 Mosier and Francis, 2011; Sintes et al., 2013). Archaeal ammonia oxidizers are also separated into 270 271 different ecotypes including Water Column ecotypes A and B (WCA and WCB), which preferentially inhabit the surface vs deep ocean, respectively, or high-ammonia concentration vs 272 low-ammonia concentration groups, which dominate in high ammonia vs low ammonia 273 concentration environments, respectively. The nxrB gene, which encodes the beta subunit of nitrite 274 275 oxidoreductase for nitrite oxidation, has been used to quantify Nitrospira (Pester et al., 2014). However, no primers targeting nxr genes are available for other groups of nitrite oxidizers such as 276 Nitrospina, which is the dominant group of nitrite oxidizers in the ocean (Beman et al., 2013; 277





- Pachiadaki et al., 2017). Primers have also been designed to quantify the 16S rRNA gene abundance of *Thaumarchaeota*, *Nitrospira*, and *Nitrospina* (Mincer et al., 2007; Graham et al., 2007). The abundance of nitrifiers can be useful for inferring and interpreting nitrification rates.
 In addition to qPCR, amplicon sequencing and quantitative metagenomics are also useful to determine the abundance of nitrifiers (Tolar et al., 2020; Lin et al., 2019; Satinsky et al., 2013) but these analyses are not included in the database.
- 284

Table 3. qPCR primers commonly used to quantify nitrifier abundance in the ocean.

Target	Name	Primer sequences	References
		(5'-3')	
Gamma-bacterial	amoA-1F	GGGGTTTCTACTGGTGGT	Rotthauwe et
amoA	amoA-2R	CCCCTCKGSAAAGCCTTCTTC	al., 1997
	or		
	amoA-r NEW	CCCCTCBGSAAAVCCTTCTTC	Hornek et al.,
			2006
Water Column ecotype	Arch-amoAFA	ACACCAGTTTGGYTACCWTCDGC	Beman et al.,
A (WCA) archaeal-	Arch-amoAR	GCGGCCATCCATCTGTATGT	2008;
amoA			Francis et al.,
			2005
Water Column ecotype	Arch-amoAFB	CATCCRATGTGGATTCCATCDTG	Beman et al.,
A (WCB) archaeal-	Arch-amoAR	GCGGCCATCCATCTGTATGT	2008;
amoA			Francis et al.,
			2005
Total archaeal-amoA	Arch-amoAF	STAATGGTCTGGCTTAGACG	Francis et al.,
	Arch-amoAR	GCGGCCATCCATCTGTATGT	2005
High-ammonia	Arch-amoA-for	CTGAYTGGGCYTGGACATC	Wuchter et al.,
concentration archaeal-	Arch-amoA-rev	TTCTTCTTTGTTGCCCAGTA	2006
amoA			
Low-ammonia	Arch-amoA-for	CTGAYTGGGCYTGGACATC	Wuchter et al.,
concentration archaeal-	Arch-amoA-rev-New	TTCTTCTTCGTCGCCCAATA	2006
amoA			Sintes et al.,
			2013
Thaumarchaeota 16S	GI_751F	GTCTACCAGAACAYGTTC	Mincer et al.,
rRNA	GI_956R	HGGCGTTGACTCCAATTG	2007





nxr	nxrB169F	TACATGTGGTGGAACA	Pester et al.,
	nxrB638R	CGGTTCTGGTCRATCA	2014
Nitrospira 16S rRNA	Nspra-675f	GCGGTGAAATGCGTAGAKATCG	Graham et al.,
	Nspra-746r	TCAGCGTCAGRWAYGTTCCAGAG	2007
Nitrospina 16S rRNA	NitSSU_130F	GGGTGAGTAACACGTGAATAA	Mincer et al.,
	NitSSU_282R	TCAGGCCGGCTAAMCA	2007

286

287 Results and Discussion

288 Summary of the database

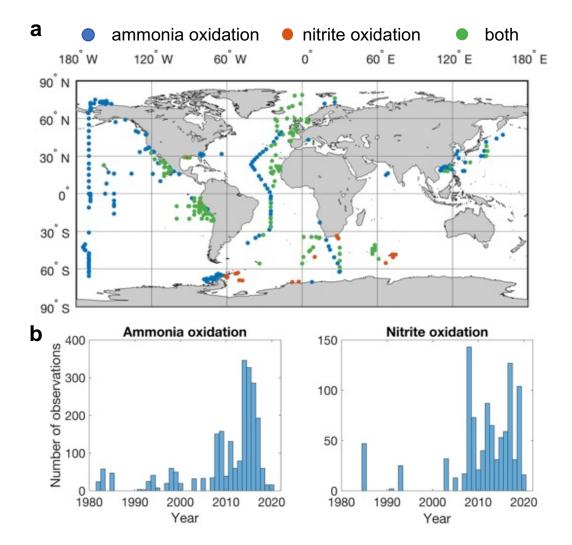






Figure 2. Map showing the distribution of sampling locations for nitrification rate measurements
(a) and the number of observations each year (b). Blue points: only ammonia oxidation is
measured; red points: only nitrite oxidation is measured. Green points: both ammonia oxidation
and nitrite oxidation are measured.

294

295 In total, there are 2393 and 1006 measurements of ammonia oxidation and nitrite oxidation, 296 respectively (Figure 2). Ammonia oxidation and nitrite oxidation have been concurrently measured 297 at 418 locations. The Pacific Ocean has the largest number of nitrification observations followed 298 by the Atlantic Ocean, Southern Ocean and Indian Ocean. Particularly, meridional transects across 299 ocean basins and biomes have been conducted in the North Pacific and Atlantic (Shiozaki et al., 300 2016; Clark et al., 2008; Clark et al., 2022). Observations have recently expanded into oxygen 301 minimum zones (Beman et al., 2012; Beman et al., 2013; Frey et al., 2020; Frey et al., 2022; Peng et al., 2015; Peng et al., 2016; Santoro et al., 2021; Sun et al., 2017) and polar oceans (Cavagna et 302 303 al., 2015; Shiozaki et al., 2019; Smith et al., 2022; Mdutyana et al., 2022a and b; Mdutyana et al., 2020; Flynn et al., 2021). Nitrification rates are more frequently measured after 2010 (Figure 2b). 304 305

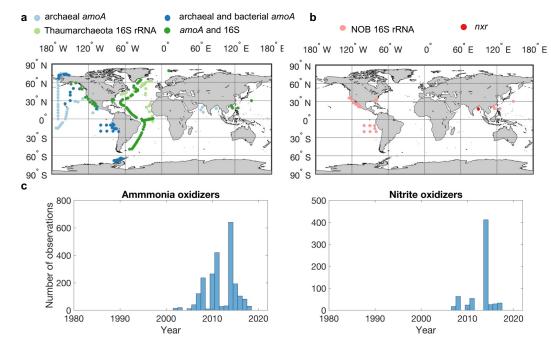






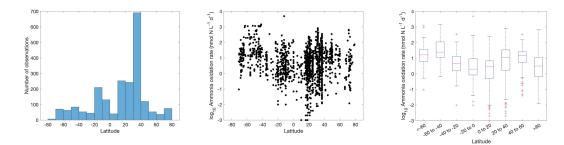
Figure 3. Maps showing the distribution of sampling locations for ammonia oxidizers (a) and nitrite oxidizers (b), and the number of observations each year (c). (a) light blue points: only archaeal *amoA* was quantified. Dark blue points: both archaeal and bacterial *amoA* genes were quantified. Light green points: 16S rRNA gene of *Thaumarchaeota* was quantified; dark green points: both archaeal *amoA* and 16S rRNA gene of *Thaumarchaeota* were quantified. (b) pink points: 16S rRNA of nitrite oxidizers was quantified; red points: *nxr* gene of nitrite oxidizers was quantified.

314

In total, there are 2187 and 631 measurements of ammonia oxidizer and nitrite oxidizer abundance, 315 316 respectively (Figure 3). Most of the nitrifier quantifications have been conducted in the tropical 317 and subtropical oceans (Figure 4a). Data are sparse in the central Pacific, Indian Ocean and 318 Southern Ocean (with the exception of the West Antarctic Peninsula). Both archaeal amoA and 319 16S rRNA genes of Thaumarchaeota were quantified on a transect across the Atlantic (Sintes et 320 al., 2016). There are far fewer observations of nitrite oxidizers compared to ammonia oxidizers. 321 Notably, there are only 27 observations of nxr genes. The number of the quantification of nitrifier 322 abundance starts to accumulate since 2002 (Figure 3c). Most of the observations of nitrite oxidizers 323 originates from one study where samples were collected in 2014 (Liu et al., 2018). Nitrification 324 rate and nitrifier abundance are sometimes determined at the same location, which allows us to 325 assess the relationship between biogeochemical rate and the abundance of functional groups (e.g., 326 Peng et al., 2015; Shiozaki et al., 2019; Santoro et al., 2021).

327

328 Distribution of ammonia oxidation



329

Figure 4. Number of ammonia oxidation observations and ammonia oxidation rates within
latitudinal bands. For the boxplot in this figure and figures throughout the manuscript, the red line
in each box is the median. The bottom and top of each box are the 25th and 75th percentiles of the





- 333 observations, respectively. The error bars represent 1.5 times the interquartile range away from the
- bottom or top of the box, with red + signs showing outliers beyond that range.
- 335

A large number of observations exist for the tropical and temperate oceans (Figure 4), particularly 336 in the 30-40°N band where rates were measured in offshore waters of Georgia and California 337 338 (Tolar et al., 2020; Liu et al., 2018). Ammonia oxidation rates vary from <0.01 to over 1000 nmol N L⁻¹ d⁻¹ with a median value of 7.7±9.8 nmol N L⁻¹ d⁻¹. There is no clear latitudinal trend in the 339 ammonia oxidation rates. In contrast, Clark et al. (2022) found higher ammonia oxidation rates in 340 341 the southern hemisphere along the north-south transect in the Atlantic Ocean. This latitudinal pattern is hypothesized to be explained by the difference in the supply of dissolved organic 342 343 nitrogen (DON) by lateral transport into the gyre interior from the eastern boundary upwelling (Clark et al., 2022). The stimulation of ammonia oxidation rates by a lateral DON supply has also 344 been observed in the Western Pacific (Xu et al., 2018). 345

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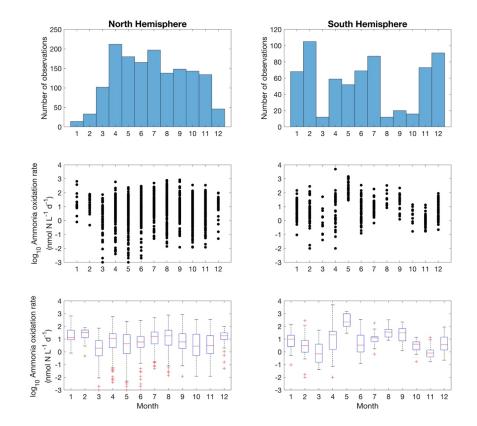


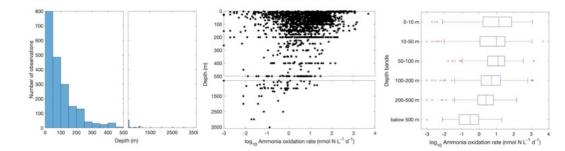




Figure 5. Monthly variation of ammonia oxidation observations and ammonia oxidation rates
divided into observations taken in the Northern Hemisphere (left panels) and Southern Hemisphere
(right panels).

351

More ammonia oxidation measurements were conducted in summer in both hemispheres (Figure 352 353 5) which is likely due to the more challenging weather conditions in winter for field explorations. The northern hemisphere has more observations compared to the southern hemisphere. Although 354 355 no clear seasonal pattern is apparent for ammonia oxidation rates at a global scale, seasonal variation in ammonia oxidation has been seen at time-series stations near and offshore of 356 California (Ward, 2005; Tolar et al., 2020; Laperriere et al., 2020). In addition, ammonia oxidation 357 358 showed a substantial seasonal pattern in the polar ocean with higher rates observed in the NH4⁺enriched dark winter season (Baer et al., 2017; Mdutyana et al., 2020; Mdutyana et al., 2022b). 359 360



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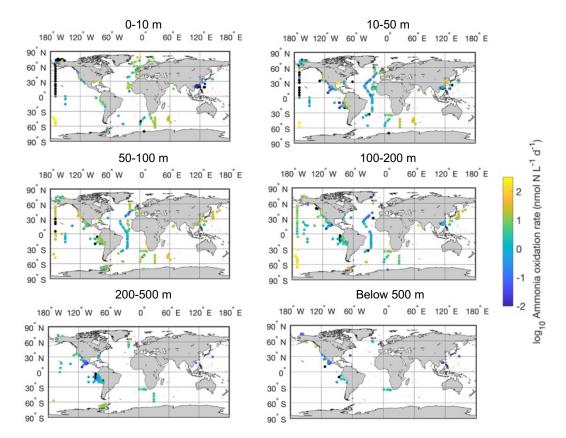
Figure 6. Vertical variation of ammonia oxidation observations and ammonia oxidation rates.

364 Most of the ammonia oxidation rate measurements were made shallower than 500 m, accounting 365 for ~96% of the total measurements (Figure 6). Ammonia oxidation rates often reach a maximum 366 near the base of the euphotic zone or in the 50-100 m layer before decreasing with depth below the euphotic zone. Although nitrification is thought to be inhibited by light, high ammonia 367 oxidation rates >100 nmol N L⁻¹ d⁻¹ have been observed within the euphotic zone (Raes et al., 368 2020; Bianchi et al., 1997), suggesting complex regulation of nitrification in the surface ocean. 369 This complicates the interpretation of the source of NO₃⁻ in the euphotic zone and further the NO₃⁻ 370 -supported new production (Diaz and Raimbault, 2000; Yool et al., 2007; Grundle et al., 2013; 371 372 Mdutyana et al. 2020).





373



374

Figure 7. Spatial distribution of ammonia oxidation rates in six depth intervals in the global ocean.Locations with rates below the detection limit are shown in the black circles.

377

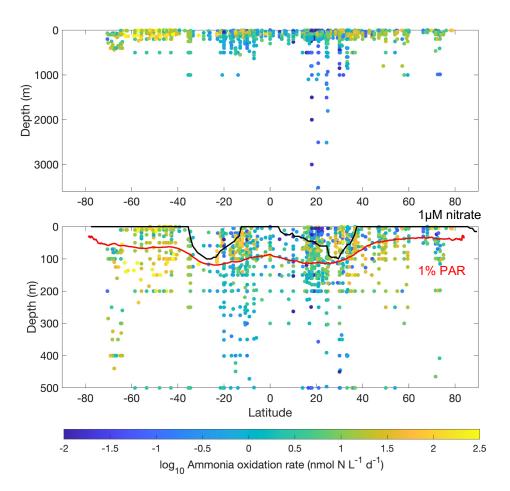
There is a large spatial and vertical variation in ammonia oxidation rates (Figure 7). Some hotspots 378 with rates >100 nmol N L⁻¹ d⁻¹ include the subpolar North Atlantic (Clark et al., unpublished), 379 380 Southern Ocean (Mdutyana et al., 2020), and coastal waters off California and Georgia (Tolar et 381 al., 2020; Liu et al., 2018). Particularly, there are extremely high ammonia oxidation rates >1000 nmol N L⁻¹ d⁻¹ observed in the surface Pacific Southern Ocean (Raes et al., 2020), deserving further 382 studies to confirm this pattern. In contrast, some low rates <0.01 nmol N L⁻¹ d⁻¹ or rates below the 383 384 detection limit are found in the surface sunlit North Pacific, which is likely caused by the light inhibition on nitrifiers, and nitrifiers' competition with phytoplankton for NH_4^+ in well-lit areas 385 386 (Smith et al., 2014). For example, peak ammonia oxidation rates are often found in regions/depths





387 where NO_{3}^{-} is present or light levels are low such that competition of nitrifiers with phytoplankton 388 for NH_{4}^{+} diminishes (Figure 8; Wan et al., 2021). Additionally, low rates are found in oxygen-389 depleted waters of the eastern tropical Pacific where ammonia oxidation is likely limited by 390 oxygen availability (Peng et al., 2016)

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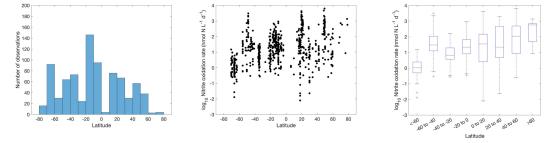
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Figure 8. Latitudinal and vertical distribution of ammonia oxidation rates. The lower panel shows
data from the top 500 m. The climatological depths of the euphotic zone (1% PAR) obtained from
MODIS satellite observations and 1 µM nitrate obtained from World Ocean Atlas 2018 (García et
al., 2019) are shown by the red line and black lines, respectively.





398 Distribution of nitrite oxidation



400 Figure 9. Number of nitrite oxidation observations and nitrite oxidation rates within latitudinal401 bands.

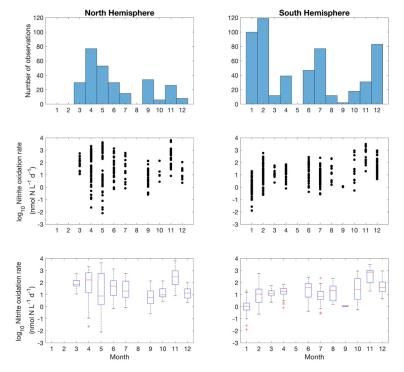
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399

403 Similar to ammonia oxidation, the majority of the nitrite oxidation observations were conducted in the tropical and subtropical oceans (Figure 9), particularly in the eastern tropical Pacific oxygen 404 405 minimum zones (Ward et al., 1989; Peng et al., 2015; Kalvelage et al., 2013; Santoro et al., 2021). Recent observations extended into the Southern Ocean (Cavagna et al., 2015; Mdutyana et al., 406 2020; Mdutyana et al., 2022a; Flynn et al., 2021). The rates vary from 0.01 to >1000 nmol N L⁻¹ 407 d⁻¹ with a median value at 15.9±10.7 nmol N L⁻¹ d⁻¹. Nitrite oxidation rates seem to increase from 408 the southern hemisphere to northern hemisphere. The lowest median rates were found in the 409 410 Southern Ocean south of 60°S, which is hypothesized to be regulated by low iron availability 411 (Mdutyana et al., 2022a). Overall, more measurements of nitrite oxidation over a large spatial scale 412 are desired to resolve the latitudinal distribution of nitrite oxidation rates. 413





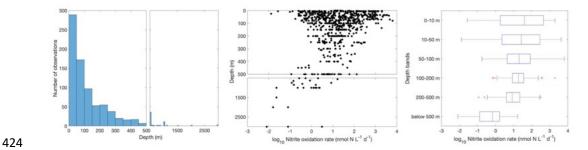


414

415 Figure 10. Monthly variation of nitrite oxidation observations and nitrite oxidation rates.

416

417 Nitrite oxidation measurements are limited in winter in the northern hemisphere (Figure 10). No 418 clear seasonal pattern is found for nitrite oxidation rates at a global scale, except for some of the 419 lowest rates detected in January in the Southern Ocean (austral summer). In addition to iron 420 limitation, light inhibition and competition with phytoplankton for nitrite during the growing 421 season may be important factors driving these low rates. Unlike ammonia oxidation, there is no 422 time-series study of nitrite oxidation to show its seasonal variations.







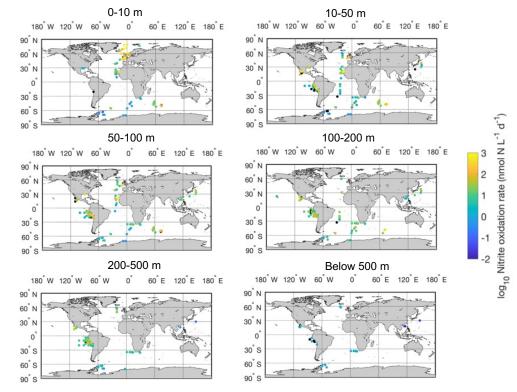
425 Figure 11. Vertical variation of nitrite oxidation observations and nitrite oxidation rates.

426

427 Most of the nitrite oxidation rates were also measured at depths shallower than 500 m, accounting 428 for $\sim 94\%$ of the total measurements (Figure 11). There is a large spatial variation in the nitrite 429 oxidation observations and rates (Figure 12). Observations are lacking in the central Pacific Ocean 430 and Indian Ocean outside of the oxygen minimum zones. Nitrite oxidation rates decrease with 431 depth. Globally, the highest median nitrite oxidation rates were found in the surface water (0-10 m layer), which is mainly attributed to the high surface rates observed over the United Kingdom 432 shelves, subpolar North Atlantic and Mauritanian upwelling system (Figure 12; Clark et al., 433 434 unpublished; Clark et al., 2016). After removing these high surface nitrite oxidation rates, the depth 435 profiles of nitrite oxidation often show a subsurface maximum that is slightly deeper than the 436 subsurface maximum of ammonia oxidation (Figure 13). This difference may be related to the higher sensitivity of nitrite oxidizers/nitrite oxidation to light (Wan et al., 2021; Olson, 1981b). 437 438 Interestingly, some deep peaks of nitrite oxidation rates have been found in the oxygen-depleted waters in the oxygen minimum zones (Peng et al., 2015; Babbin et al., 2020; Ward et al., 1989; 439 440 Beman et al., 2013). These high rates stand out in depths below the 1 μ M nitrate threshold and above the 1% PAR level between 20°N and 20°S (Figure 14). Many hypotheses (Sun et al., 2023) 441 442 have been proposed to explain the observed "anaerobic" nitrite oxidation, including alternative 443 oxidants like iodate (Babbin et al., 2017), distinct nitrite oxidizers that are only present in the 444 OMZs and adapted to the low oxygen conditions (Sun et al., 2021), nitrite dismutation $(2H^+ +$ $5NO_2^- \rightarrow N_2 + 3NO_3^- + H_2O$; van de Leemput et al., 2011; Babbin et al., 2020; Tracey et al., 2022), 445 and oxygen intrusions (Buchanan et al., 2023). Whether nitrite oxidation is truly anaerobic and 446 447 how nitrite oxidation is sustained in oxygen depleted waters remain to be determined. 448







449

450 Figure 12. Spatial distribution of nitrite oxidation rates in six depth intervals in the global ocean.

451 Locations with rates below the detection limit are shown in the black circles.

452

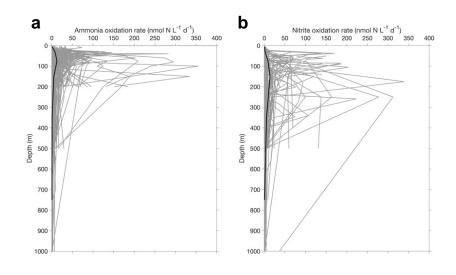


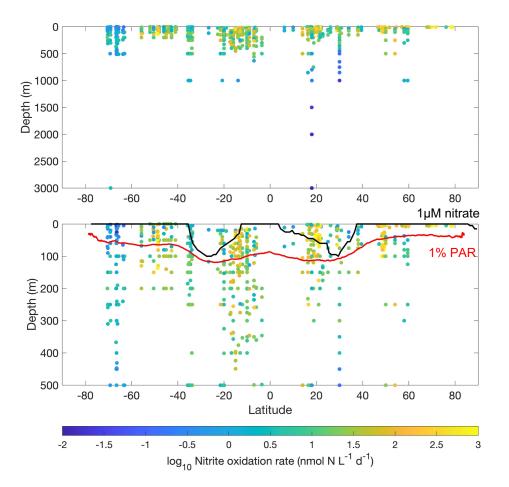




Figure 13. Depth profiles of ammonia oxidation (a) and nitrite oxidation (b) in the top 1000 m. Only depth profiles with five or more measurements are included in this figure. The median profiles of ammonia oxidation and nitrite oxidation are shown in thick black lines, showing the

457 maximum of nitrite oxidation deeper than the maximum of ammonia oxidation.

458



459

Figure 14. Latitudinal and vertical distribution of nitrite oxidation rates. The lower panel shows data from the top 500 m. The climatological depth of the euphotic zone (1% PAR) and 1 μ M nitrate are shown by the red and black lines respectively.





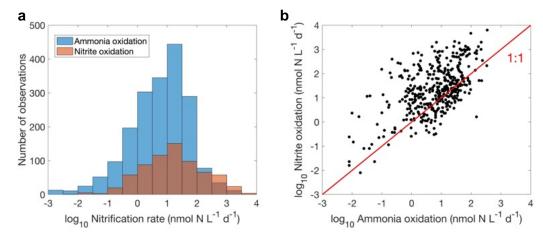


Figure 15. Comparison between ammonia oxidation and nitrite oxidation. (a) Histogram of both rates globally. (b) Relationship between ammonia oxidation and nitrite oxidation measured at the same locations and time (y=0.62*x+0.82, r=0.54, p<0.01).

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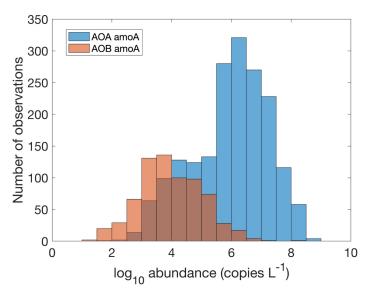
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Overall, there are fewer nitrite oxidation rate measurements compared to ammonia oxidation 469 470 measurements (Figure 15a). Ammonia oxidation and nitrite oxidation are generally of similar 471 magnitude (Figure 15b), leading to the low concentration of NO₂⁻ in most of the ocean. However, 472 ammonia oxidation and nitrite oxidation could be decoupled. For example, higher ammonia 473 oxidation rates than nitrite oxidation rates (Lomas and Lipschultz, 2006) and competition between 474 ammonia oxidation and phytoplankton ammonium assimilation (Zakem et al. 2018) may both 475 partly explain the presence of the primary nitrite maximum The median nitrite oxidation rate is higher than the median ammonia oxidation rate (15.9 vs 7.7 nmol N L⁻¹ d⁻¹), which may be related 476 477 to nitrite production pathways from urea and cyanate oxidation in addition to ammonia oxidation 478 (Wan et al., 2022; Kitzinger et al., 2018). Consistently, when comparing ammonia oxidation and 479 nitrite oxidation rates measured at the same locations and same time, nitrite oxidation rates are 480 mostly higher (Figure 15b). Mechanisms driving the decoupling of ammonia oxidation and nitrite 481 oxidation deserve further investigations. 482





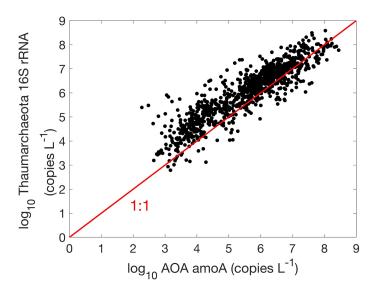
483 Distribution of ammonia oxidizers



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Figure 16. Comparison between the gene abundance of AOA *amoA* and AOB *amoA*.

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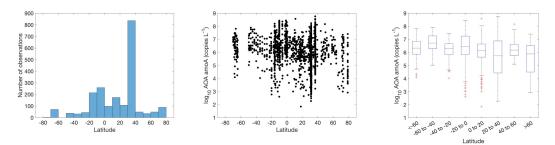
Figure 17. Comparison between archaeal *amoA* and *Thaumarchaeota* 16S rRNA gene abundances
(y=0.78*x+1.66, r=0.9, p<0.01).





491 There are 1893, 892, 1073 measurements of the abundance of archaeal amoA gene, bacterial amoA 492 and 16S rRNA of Thaumarchaeota, respectively. In addition, 1204 and 1101 measurements were separately conducted for water column ecotype A (WCA) amoA and water column ecotype B 493 (WCB) *amoA*. The AOA *amoA* abundance with median of 1.34×10^6 copies L⁻¹ is substantially 494 higher than AOB amoA gene abundance with median of 7.96 x 10³ copies L⁻¹ (Figure 16), 495 496 confirming the dominance of archaeal ammonia oxidizers in the ocean. We also found that 497 Thaumarchaeota 16S rRNA gene abundance positively correlates with but slightly outnumbers the *amoA* gene abundance (Figure 17). This may suggest that not all the *Thaumarchaeota* contain 498 the *amoA* genes to oxidize NH_4^+ or some organisms containing *amoA* genes (such as the 499 Nitrosopumilus-like group) may have been missed due to primer bias (Sintes et al., 2016; Hiraoka 500 501 et al., preprint), Since archaeal *amoA* genes have the largest number of observations and better 502 represent ammonia oxidation capability, we will use it to show the spatial and vertical distribution 503 of ammonia oxidizer abundance.

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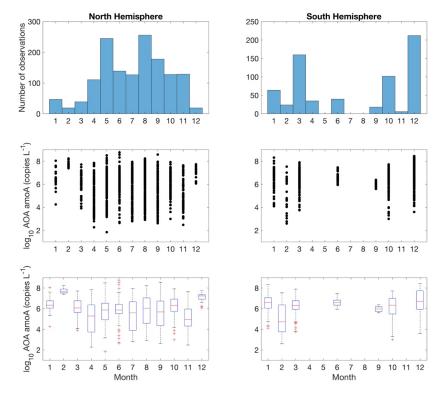
Figure 18. Number of observations of ammonia oxidizers and the abundance of ammonia oxidizerswithin latitudinal bands.

508

The eastern Pacific Ocean and Atlantic Ocean have the majority of the observations for ammonia oxidizers, particularly in the 30-40°N band where ammonia oxidizers were measured in the coastal waters off California and Georgia (Liu et al., 2018; Tolar et al., 2020). In contrast, observations in the Indian Ocean and Southern Ocean are scarce. The AOA *amoA* gene abundance varies from a few copies per liter in the surface ocean to over 10⁸ copies L⁻¹ in the subsurface of equatorial Atlantic. There is no clear latitudinal trend in the abundance of ammonia oxidizers.







516

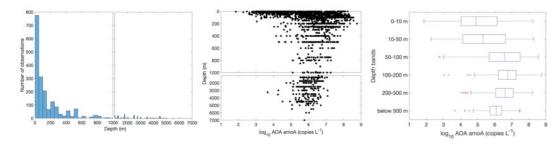
517 Figure 19. Monthly variation of the observation and abundance of ammonia oxidizers.

518

There are substantially more observations in the northern hemisphere than the southern 519 520 hemisphere. We do not see a clear seasonal trend in nitrifier abundance due to the large monthly 521 variation. A time-series study in the Monterey Bay shows that seasonality can be observed for the 522 top 200 m while the overall community of ammonia oxidizers was stable at 500 m (Tolar et al., 523 2020). In addition, mid-summer peaks in Thaumarchaeota abundance have been observed at the coast off Georgia (Hollibaugh et al., 2013). More time-series studies with high-frequency sampling 524 525 would be useful for characterizing the response of the nitrifier community to seasonal changes in environmental drivers. 526



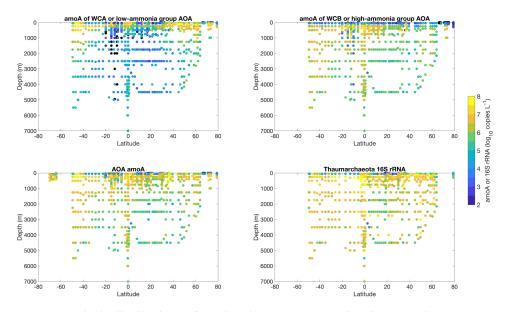




529 Figure 20. Vertical distribution of archaeal *amoA* observations and archaeal *amoA* gene 530 abundance.

531

528



532

Figure 21. Vertical distribution of archaeal *amoA* gene abundance and 16S rRNA of *Thaumarchaeota* across the latitudinal gradient. WCA and high-ammonia concentration groups
are shown together while WCB and low-ammonia concentration groups are shown together.

536

537 Most of the abundance measurements of ammonia oxidizers were made in the top 1000 m (Figure 538 20). Median ammonia oxidizer abundance increases from $\sim 10^5$ copies L⁻¹ in the 0-10 m depth layer 539 to $\sim 10^7$ copies L⁻¹ in the 100-200 m layer, then decreases with depth and remains relatively 540 constant at $\sim 10^6$ copies L⁻¹ in the deep ocean below 500 m depth. The archaeal *amoA* is sometimes 541 quantified separately for two ecotypes including water column groups A and B. Water column





- group A dominates the upper 200 meter while water column group B is more abundant in the
 mesopelagic and bathypelagic deep ocean below 500 m, likely reflecting their different affinities
 for NH₄⁺ (Beman et al., 2008; Sintes et al., 2016). The vertical distribution of ammonia oxidizers
- 545 is similar to the vertical distribution of ammonia oxidation rates (Figure 13).

546

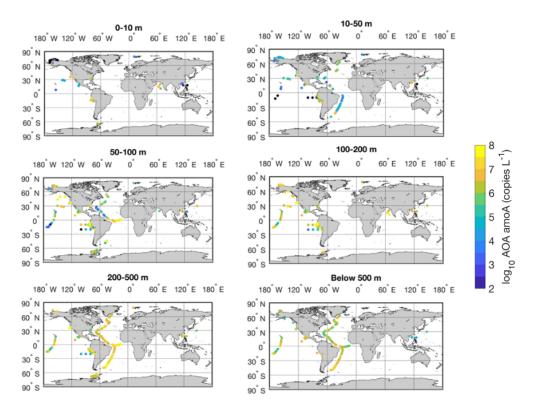




Figure 22. Spatial distribution of *amoA* gene in six depth intervals in the global ocean. Locations
with abundance below the detection limit are shown in the black circles.

550

There is a large spatial variation in the abundance of ammonia oxidizers (Figure 22). High abundances are found in the tropical Atlantic and eastern tropical Pacific where upwelling drives high rates of marine primary production. In contrast, some of the lowest abundances of ammonia oxidizers are found in the South China Sea and oligotrophic subtropical Pacific. Therefore, the distribution of marine productivity and organic matter production and export may play an important role in regulating the distribution of ammonia oxidizers because ammonia oxidizers rely on the supply of NH_4^+ , which is generated by of organic matter decomposition.





558

559 Distribution of nitrite oxidizer abundance

560 There are only seven studies available reporting the abundance of nitrite oxidizers in the ocean.

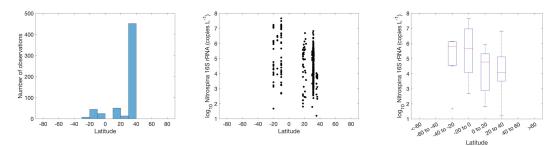
561 One study used the nxr marker gene and the other six studies used 16S rRNA gene of either

562 *Nitrospina* or *Nitrospira*. Since *Nitrospina* is the dominant nitrite oxidizer in the ocean (Beman et

al., 2013; Pachiadaki et al., 2017) and accounts for most of the observations, we use it to show the

564 distribution of nitrite oxidizers.

565

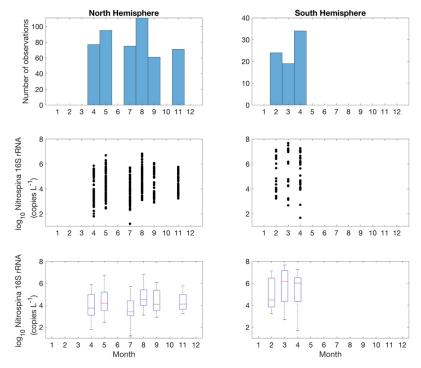


566

567 Figure 23. Number of observations and abundance of *Nitrospina* within latitudinal bands.







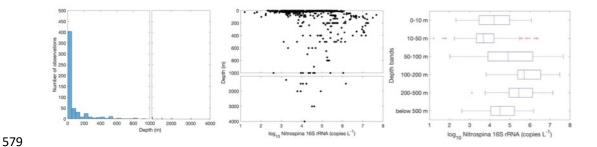
569

570 Figure 24. Monthly variation of observations and abundance of Nitrospina.

571

Quantification of nitrite oxidizers using the 16S rRNA gene is limited to a few locations between 40°N to 40°S including the coastal waters off California and Georgia, the eastern tropical South Pacific, Bay of Bengal, and western Pacific (Figure 23). The number of observations is dominated by one study conducted near the coast of Georgia (Liu et al., 2018). The highest abundance of 4.68 $\times 10^7$ copies L⁻¹ was found in the eastern tropical South Pacific. No clear latitudinal or seasonal trend can be determined based on the limited number of observations (Figures 23-24).

578







580 Figure 25. Vertical distribution of *Nitrospina* observations and gene abundance.

581

The vertical distribution of nitrite oxidizers resembles the vertical distribution of ammonia oxidizers: increases from $\sim 10^4$ copies L⁻¹ in the surface 0-10 m depth layer to a maximum of $\sim 10^6$ copies L⁻¹ in the 100-200 m layer, then decreases to $\sim 10^{4.5}$ copies L⁻¹ in the deep ocean below 500 m (Figures 25-26). However, data below 500 m are insufficient to describe the distribution of nitrite oxidizers in the deep ocean. The vertical distribution of nitrite oxidizers qualitatively matches the vertical distribution of nitrite oxidation rates (Figure 13).

588

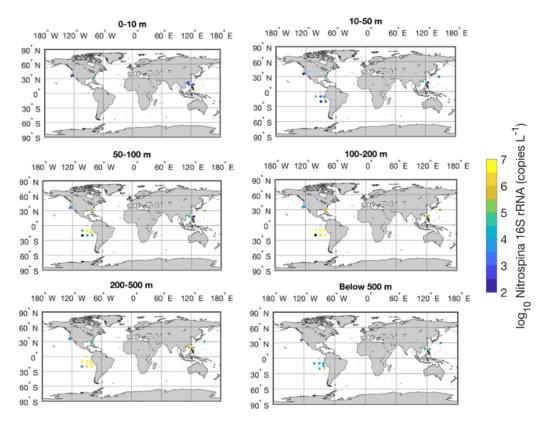




Figure 26. Spatial distribution of *Nitrospina* in six depth intervals in the global ocean. Locationswith abundances below the detection limit are shown in the black circles.





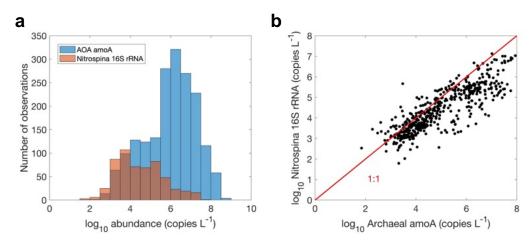


Figure 27. Comparison between the abundance of the archaeal *amoA* gene and *Nitrospina* 16S
rRNA gene (y=0.68*x+1, r=0.85, p<0.01).

596

593

When comparing ammonia oxidizers with nitrite oxidizers, median abundance of ammonia 597 oxidizers of 1.34×10^6 copies L⁻¹ is approximately two orders of magnitude higher than the median 598 599 nitrite oxidizer of 2.14 x 10⁴ copies L⁻¹. The difference in their abundance has been predicted by the relative biomass yields and cell quotas (Zakem et al., 2018; Zakem et al., 2022) and 600 601 alternatively is explained by the difference in the mortality/loss rates between AOA and Nitrospina 602 (Kitzinger et al., 2020). In addition, there is a positive relationship between the abundance of 603 ammonia oxidizers and nitrite oxidizers (Figure 27) as previously shown in observations from the 604 Pacific (Santoro et al., 2019), indicating their coexistence under most conditions.

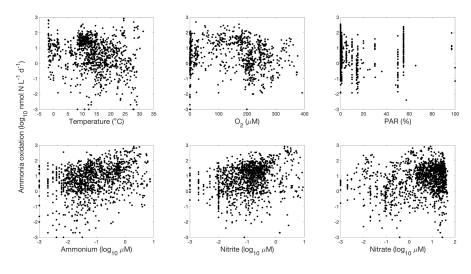
605

606 Environmental controls on nitrification rates and the abundance of nitrifiers

We compared the measured nitrification rates and nitrifier abundance with concurrently measured or available environmental factors including temperature, oxygen, light, and N concentration (NH_4^+, NO_2^-, NO_3^-) to assess the environmental controls on nitrification and nitrifiers (Figures 28-31). We acknowledge that nitrification rates and nitrifier abundance are regulated by multiple environmental factors, which may not be revealed by the simple correlation analysis with individual factors. The new database will facilitate more sophisticated future analyses.









615 Figure 28. Relationship between ammonia oxidation rates and environmental factors.

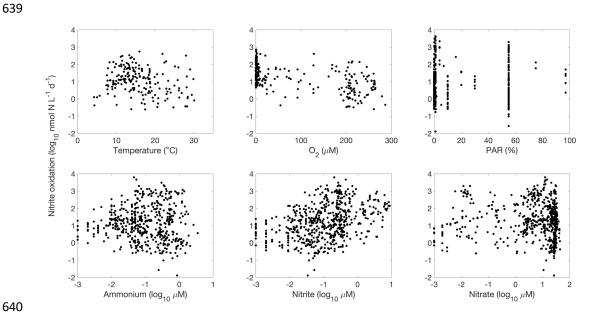
616

Ammonia oxidation rates did not show a clear pattern with temperature (r=-0.22, p<0.01; Figure 617 618 28). Some of the high rates are found below 0°C, and at around 10°C and 25°C. Temperature 619 manipulation experiments showed varying temperature sensitivity of ammonia oxidation in 620 different regions or among natural assemblages (Baer et al., 2014; Horak et al., 2018; Zheng et al., 621 2020). The highest ammonia oxidation rates were found in the oxygen range between 100 and 200 μ M (p>0.01). But ammonia oxidation has also been detected in low oxygen waters (e.g., <10 μ M) 622 in the oxygen minimum zones (Bristow et al., 2016a; Peng et al., 2015), reflecting the high affinity 623 624 of ammonia oxidizers for oxygen. Oxygen production by ammonia-oxidizing archaea may support their presence and activity in the oxygen minimum zones (Kraft et al., 2022). Ammonia oxidation 625 generally decreases at relatively high light intensity (PAR% relative to surface PAR) due to light 626 inhibition and substrate competition with phytoplankton (but the negative slope is not significant, 627 p > 0.01). Nevertheless, high ammonia oxidation rates have been measured in the euphotic zone 628 629 at 55% PAR in the Atlantic Ocean (Clark et al., 2008; Clark et al., unpublished). Ammonia oxidation increases with N nutrient concentration (p < 0.01). NH₄⁺ is the substrate while NO₂⁻ is the 630 631 product of ammonia oxidation. The Michaelis-Menten-like kinetics of ammonia oxidation rate 632 have been observed in various ocean regions (Frey et al., 2022; Newell et al., 2013; Horak et al., 633 2013; Xu et al., 2019; Zhang et al., 2020; Mdutyana et al., 2022a and b). High concentrations of





 NH_4^+ and NO_2^- likely reflect intense recycling of organic matter and remineralization. The presence of high NO_3^- concentration may relieve the competition between ammonia oxidizers and phytoplankton for NH_4^+ , therefore leading to high ammonia oxidation rates (Wan et al., 2018). In addition, recent studies have shown that AOA have a high requirement for iron and copper, which may affect the distribution of nitrification in the ocean (Shafiee et al., 2019; Shafiee et al., 2021).



641 Figure 29. Relationships between nitrite oxidation rates and environmental factors.

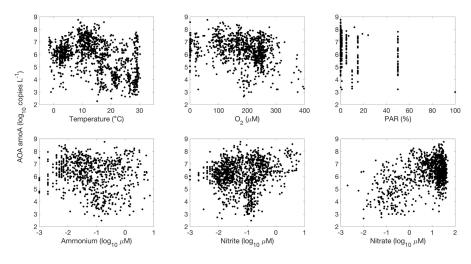
642

643 High nitrite oxidation rates are found between 10°C and 20°C (Figure 28). Surprisingly, some of 644 the highest nitrite oxidation rates were measured in the oxygen minimum zones even with oxygen 645 levels below detection limits (Ward et al., 1989; Sun et al., 2017; Sun et al., 2021). Nitrite oxidation in anoxic waters has been observed to be inhibited (Sun et al., 2017) or stimulated (Bristow et al., 646 647 2016a) by the addition of oxygen. The mechanisms for apparently anaerobic nitrite oxidation 648 remain to be determined (Sun et al., 2023). Similar to ammonia oxidation, nitrite oxidation is often 649 reported to be inhibited by high light levels, but the relationship is not statistically significant 650 across the database (p>0.01; Figure 29) partly due to the presence of high nitrite oxidation rates in 651 the euphotic zone (e.g., Clark et al., 2016). High nitrite oxidation rates are often observed in regions 652 with high NO_2^- concentration (r=0.23, p<0.01). For example, the highest nitrite oxidation rates were observed at NO_2^- concentrations near 0.5 μ M (Figure 29). 653





654



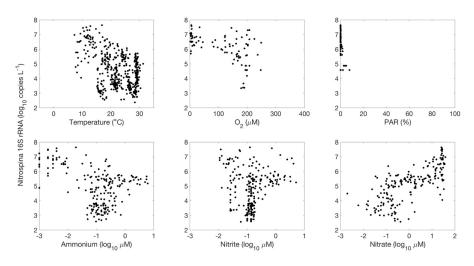
655

Figure 30. Relationship between archaeal *amoA* gene abundance and environmental factors.

We use *amoA* gene abundance to represent the abundance of ammonia oxidizers with the caveat that the number of gene copies may not equal the cell numbers. Ammonia oxidizers are adapted to a wide range of environmental conditions (Figure 30). Their abundance reaches a maximum at around 10°C. Ammonia oxidizers are also present in low oxygen waters and the euphotic zone with slightly lower abundance. Interestingly, ammonia oxidizers show relatively constant abundance across the NH₄⁺ concentration gradient while ammonia oxidation rates are low under low NH₄⁺ concentration (e.g., <0.01 μ M).







666

Figure 31. Relationships between *Nitrospina* 16S rRNA gene abundance and environmentalfactors.

669

670 It is difficult to evaluate the relationship between nitrite oxidizers and environmental factors due 671 to the limited number of observations (Figure 31). Nevertheless, one interesting pattern is the 672 presence of high *Nitrospina* abundance in oxygen depleted waters. The nitrite oxidizers present in 673 the oxygen depleted waters are distinct from those found in oxygenated waters or currently 674 cultivated strains (Sun et al., 2019; Sun et al., 2021).

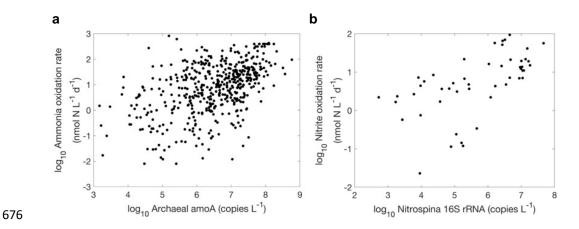






Figure 32. Relationships between nitrifier abundance and nitrification rate. (a) ammonia oxidation vs AOA *amoA* gene abundance (y=0.46*x-2.08, r=0.48, p<0.01); (b) nitrite oxidation vs

679 *Nitrospina* 16S rRNA gene abundance (y=0.35*x-1.27, r=0.59, p<0.01).

680

681 There is a positive correlation between AOA amoA gene abundance and ammonia oxidation rates 682 (r=0.44, p<0.01), and between *Nitrospina* 16S rRNA abundance and nitrite oxidation rate (r=0.58, p < 0.01) (Figure 32) even though the correlation is weak. This lack of a strong relationship has also 683 684 been found in regional studies (Tolar et al., 2020), which may be caused by the perturbation of the 685 microbial community during rate measurement incubations. Furthermore, the addition of nitrogen 686 substrate during rate measurement incubations may stimulate the growth of nitrifiers and the 687 subsequently measured nitrification rate. Overall, using functional gene abundances to predict their 688 functional activity needs to be conducted with caution since the presence of genes only reflects the 689 functional potentials.

690

691 Applications of the database and suggestions for future research

692 This database will be useful for a broad scientific community who are interested in marine 693 biogeochemistry and marine microbial ecology. Potential applications include but are not limited 694 to: 1) Compare future measurements of nitrification rates or nitrifier abundance in a specific region 695 to previous measurements conducted in the same regions, and contextualize new measurements at 696 the global scale. 2) Meta-analysis of environmental controls on the distribution of nitrification and 697 nitrifier abundance at regional and global scales. The simple correlation analyses provided here 698 only considered individual environmental drivers separately while other drivers are changing simultaneously. Analysis with environmental assemblages could complement laboratory culture 699 700 experiments. 3) Help to validate and improve model parametrization of nitrification and nitrifiers. 701 For example, ammonia oxidation has been modeled as a function of substrate concentration and 702 specific ammonia oxidation rate (Yool et al., 2007). However, nitrification has been found to be 703 regulated by temperature, oxygen, light and many other environmental factors that are not considered in that model. A better representation of nitrification in ocean biogeochemical models 704 705 could help to constrain the estimates of marine new production, N₂O production and many other 706 key processes. 4) Guide future nitrification studies, e.g., conducting measurements in poorly 707 sampled regions or seasons.





708

709 Based on the historical measurements of nitrification and nitrifiers, we provide recommendations

- 710 for future research below.
- 711

712 1. Method standardization is useful for comparison among studies. Nitrification has been mostly measured by ¹⁵N substrate tracer addition and product dilution methods. The ¹⁵N product dilution 713 method tends to yield higher rates relative to the ¹⁵N substrate tracer addition method (Figure 1). 714 This is perhaps to be expected since the ¹⁵N product dilution method measures all the NO₂-715 production pathways including ammonia oxidation (and organic N oxidation) and nitrate reduction 716 instead of only NO2⁻ production from ammonia oxidation as is measured by the ¹⁵NH4⁺ tracer 717 addition method. Comparison between different methods should be conducted to resolve the 718 719 difference or even correct some of the previous measurements.

720

Additionally, the amount of tracer added should be recorded and reported because the increased substrate concentration may enhance nitrification rate. Therefore, the measured rates should be interpreted as potential rates rather than in-situ rates when the amount of tracer addition is large compared to the ambient substrate concentration. If possible, substrate kinetic experiments should be conducted for in-situ rate calibration (e.g., Wan et al., 2018; Mdutyana et al., 2022a and b).

726

The measured product of ammonia oxidation should also be reported (e.g., either only NO_2^- or NO₂⁻⁺ NO_3^-). When ambient NO_2^- concentration is low, the ¹⁵NO₂⁻ produced from ¹⁵NH₄⁺ tracer may be further oxidized to ¹⁵NO₃⁻. Thus, nitrification may be underestimated if only NO_2^- is measured. Alternatively, NO_2^- carrier may be added into the incubation to 'trap' the produced ¹⁵NO₂⁻. In addition to only measuring ammonia oxidation, more observations of nitrite oxidation are desirable to evaluate mechanisms controlling the coupling or decoupling of the two steps of nitrification.

734

Furthermore, measurements with at least three time points are preferred during the incubation time
courses in order to examine whether the rate has changed during the incubation period. Depending
on the incubation period, nitrification rates are reported as either nmol N L⁻¹ d⁻¹ or nmol N L⁻¹ h⁻¹.
A conversion factor (e.g., 12 or 24 hours) is required to obtain the same unit. The choice of the





conversion factor may be critical if there is a diel cycle of nitrification rate, e.g., in the euphotic zone where light/solar radiation varies diurnally (Wan et al., 2021). Therefore, incubation conducted under both light and dark conditions may be preferable to obtain the daily nitrification rates. The detection limit of rate measurements should also be estimated and reported (Santoro et al., 2013) instead of presenting rates that are below detection limit as zero.

744

745 For in-situ rate measurements, incubations should mimic the in-situ environmental conditions as 746 closely as possible, e.g., using light filters to simulate in-situ light/solar radiation intensity and 747 quality; using a temperature-controlled incubator to simulate the in-situ temperature. Particularly 748 for samples collected in the oxygen minimum zones, oxygen concentration in the incubation 749 containers should be measured or monitored throughout the incubation because oxygen 750 contamination is common during the sampling process (Garcia-Robledo et al., 2021). Samples 751 collected from the anoxic layer of the oxygen minimum zones need to be purged with helium or 752 nitrogen gas to remove any oxygen contamination before incubation.

753

754 2. Various primers have been designed to target ammonia oxidizers. However, current primers 755 miss the *Nitrosopumilus*-like *amoA* (Tolar et al., 2013; Hiraoka et al., preprint) and such group 756 accounts for a large fraction of the AOA based on 16S rRNA sequencing (Tolar et al., 2020). New 757 primers or techniques need to be developed to cover the diverse groups of ammonia oxidizers. In 758 addition, the quantification of nitrite oxidizers is limited. Developing primers for nxr genes may 759 be useful to untangle the relative contribution of different nitrite oxidizers particularly for the 760 unique ones found in the oxygen minimum zones. The report of qPCR assay should follow the 761 MIQE guidelines (Bustin et al., 2009) including the amplification conditions, amplification 762 efficiency, detection limit and other parameters. Alternatively, the abundance of nitrifiers may be 763 determined with quantitative metagenomics (Lin et al., 2019; Satinsky et al., 2013). In comparison 764 to the gene presence, gene expression and protein synthesis may be better linked to the activity of 765 nitrifiers (Tolar et al., 2016; Frey et al., 2022; Saito et al., 2020), deserving more observations.

766

767 3. Future observations should target regions that have been poorly sampled and regions that are
768 experiencing or expected to experience dramatic changes. For example, the Indian Ocean has the
769 fewest number of observations of nitrification and nitrifiers. With regards to change, oxygen





- minimum zones are projected to change under future climate (Breitburg et al., 2018; Busecke et
 al., 2022). Polar oceans (Arctic Ocean and Southern Ocean) are experiencing warming, ice melt
 (which affects light/solar radiation availability) and ocean acidification (Meredith et al., 2019).
- 773 Upward nutrient supply into the subtropical gyres may be affected due to enhanced stratification
- (Li et al., 2020). How nitrification will respond to these changes deserves further exploration.
- 775

Time-series studies, observations across a large-scale transect, and observations at a mesoscale or submesoscale would be desirable for investigating the temporal and spatial variation of nitrification rates and nitrifier abundances. When possible, both nitrification rates and nitrifier abundance should be measured at the same locations. While this approach incurs logistical and financial complications in requiring collaborations among laboratories with different expertise, the benefit to comprehensive process description is manifold.

782

783 4. Incubation conditions (mentioned in point 2) and ambient environmental conditions associated with rate measurements or gene quantification should be recorded and reported (e.g., temperature, 784 785 light, substrate concentration, oxygen). This information would be helpful for comparison among 786 different studies and future meta-analyses of environmental controls on nitrification and nitrifiers. 787 For example, light/solar radiation should be reported as both absolute light/solar radiation intensity 788 and relative light/solar radiation intensity to the surface ocean. Analysis of trace metals like iron 789 and copper concentration will be useful to assess their impact on nitrification. Standard notation 790 should be used to denote measurements below detection limit or measurements not conducted, 791 e.g., BDL for below detection limit, NM for not measured, empty/NA for data not available. A 792 data compilation template is provided for anyone who is interested in contributing to the database 793 with new datasets or datasets currently not included in the database. We encourage the scientific 794 community to contact us with suggestions to improve the database and to contribute to the database 795 with new datasets or datasets currently not included in the database.

796

797 Data availability

798 Data described in this manuscript can be accessed at Zenodo repository under data doi:

799 <u>https://doi.org/10.5281/zenodo.7942922</u> (Tang et al., 2023).





800

801 **Conclusions**

802 We present a newly compiled database of nitrification rate and nitrifier abundance measurements 803 in the global ocean. This database sheds light on the spatial and temporal pattern of nitrification 804 and nitrifiers even though the spatial and temporal coverages remain limited. In recent years, 805 observations have expanded into oxygen minimum zones and polar oceans while the Indian Ocean and Pacific Basin remain poorly sampled, especially with regard to nitrite oxidation and nitrite 806 807 oxidizers. This database can be applied to assess the environmental controls on nitrification at 808 regional and global scales, to validate and develop biogeochemical models, to guide future 809 observational efforts, and to better constrain the distribution of nitrification and assess its impact 810 on the marine ecosystem and climate. This database has been deposited into the Zenodo repository 811 and can be updated with new datasets.

812

813 Author contributions

Weiyi Tang and Bess Ward designed the study with input from Fabien Paulot and Charles Stock.
Weiyi Tang compiled the database with data contribution from coauthors, and Weiyi Tang
analyzed the database. Weiyi Tang and Bess Ward wrote the manuscript with contribution from
coauthors.

818

819 **Competing interests**

820 No competing interest is declared.

821

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