



SURATLANT: a 1993-2017 surface sampling in the central part of the North Atlantic subpolar gyre

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1 Abstract

This paper presents the SURATLANT dataset (SURveillance ATLANTique), consisting of individual data of temperature, salinity, parameters of the carbonate system, nutrients and water stable isotopes $(\delta^{18}$ O and δ D) collected mostly from ships of opportunity since 1993along transects between Iceland and Newfoundland (doi:10.17882/54517). We discuss how the data are validated, qualified, their accuracy and the overall characteristics of the data set. The data are used to reconstruct seasonal cycles and interannual anomalies, in particular of Sea Surface Salinity (SSS), inorganic nutrients, dissolved inorganic carbon (DIC) and its isotopic composition $\delta^{13}C_{DIC}$, total alkalinity (A_t), and water isotope concentrations. Derived parameters, such as fCO₂ and pH are also estimated. The relation between salinity and A_t is estimated in these data to investigate the possibility to replace missing A_t when estimating other parameters of the carbonate system. We compare the seasonal cycle derived from these data with other climatologies. We find a period of small seasonal change between January and late April, except on the Newfoundland shelf/continental slope, when changes related with springstratification and blooms occur earlier. The data were collected in a period of multi-decennial variability associated with the Atlantic meridional oscillation with warming between 1994and 2004-2007, and the recent cooling having peaked in 2014-2016. We also observe strong salinification in 2004-2009 and fresher waters in 1994-1995 as well as since 2010 south of 54°N and in 2016-2017 north of 54°N. Indication of multi-decadal variability is also suggested by other variables, such as phosphate or DIC, but cannot be well resolved seasonally with the discrete sampling and in the presence of interannual variability. As a whole, over the 24 years ocean fCO₂ trend $(+1.9\mu atmyr^{-1})$ is close to the atmospheric trend and associated with an increase in DIC ($+0.70 \text{ }\mu\text{molkg}^{-1} \text{ }\text{yr}^{-1}$). The data also revealed a "canonical" pH decrease of -0.002 yr⁻¹. There is also a decrease in $\delta^{13}C_{DIC}$ between 2005 and 2017 (in winter, -0.015% vr⁻¹, but larger in summer, -0.042% vr⁻¹), suggesting significant anthropogenic carbon signal at play together with other processes (mixing, biological activity).

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3 **Copyright statement**

4 The author's copyright for partner 6 of this publication is transferred to the National Oceanic and 5 Atmospheric Administration (NOAA) (for DP, KS, FB and GG).

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Data availability

8 (doi:10.17882/54517)

⁷ The data set is freely available and is accessible at <u>http://www.seanoe.org/data/00434/54517/</u>





10

11 **1 Introduction**

12 The North Atlantic subpolar gyre is a major site for formation of intermediate and deep waters, and 13 thus plays a key role in the ocean meridional overturning circulation. The upper ocean circulation 14 brings to its southern and eastern parts relatively warm and salty water of subtropical origin. This water 15 is then cooled by large heat loss to the atmosphere and freshened by local excess precipitation as well 16 as by inputs of fresher water from the Arctic or from continental and ice cap origin (Boyer et al., 2007). 17 Part of this upper water then flows into the Nordic Seas, whereas the other part recirculates 18 cyclonically in the gyre, steered by topography, such as around the Reykjanes Ridge (Fig. 1) and 19 progressively transformed by winter mixing into intermediate waters in the Labrador and Irminger Seas 20 or further entrained in the dense outflows of the Nordic Seas to form Atlantic deep waters (Mercier et 21 al., 2015; Daniault et al., 2016; Rossby et al., 2017).

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23 This region is the only large part of the world ocean which has experienced a surface cooling trend over 24 the last century (Rahmstorf et al., 2015), which some have linked to changes in the meridional 25 overturning circulation and to an observed overall surface freshening (Friedman et al., 2017). It also 26 experiences very large decadal to multi-decadal variability (Yashayaev and Loder, 2016; Reverdin, 27 2010; Frajka-Williams et al., 2017; Robson et al., 2016) associated with Atlantic Multi-decadal 28 variability. This might result from atmospheric variability, as well as by changes in the strength of the 29 Meridional Overturning Circulation (Häkkinen and Rhines, 2004; Häkkinen et al., 2011; Hátún et al., 30 2005; Reverdin, 2010; Chafik et al., 2016). The most recent trend has been a large cooling and 31 freshening since 2005, which reversed a previous warming and freshening since the mid-1990s 32 (Robson et al., 2016). It was associated in 2014-2015 with particularly strong positive NAO 33 atmospheric conditions inducing large vertical mixing and deep convection in the Labrador Sea and the Irminger Sea (Yashayaev and Loder, 2016; Piron et al., 2017; de Jong and de Steur, 2016; Fröb et al., 34 35 2017).

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The North Atlantic contributes substantially to the global oceanic uptake of CO_2 , mainly due to extensive biological activity during summer and considerable heat loss during winter, as well as by the export of surface waters to the deep ocean by the ocean circulation and vertical mixing. As a result, large anthropogenic carbon inventory is evaluated in this region (e.g. Khatiwala et al 2013; Zunino et al., 2014). Takahashi et al. (2009) estimate that the annual mean air-sea CO_2 flux in the North Atlantic, north of 50° N (representing only 5% of the ocean surface), is 0.27 Pg yr⁻¹, i.e. almost 20% of the





43 global flux. Although the mean annual carbon flux is a robust result for the North Atlantic (Takahashi 44 et al. 2002, 2009; Watson et al., 2009; Schuster et al, 2013) there is still disagreement in the magnitude 45 of seasonal, interannual to decadal variability depending on the method used to evaluate air-sea CO_2 46 fluxes (Schuster et al., 2013). Compared to other basins, the air-sea CO_2 fluxes interannual variability 47 in the North Atlantic appears relatively small (Rödenbeck et al., 2015; Landschützer et al., 2016), 48 although during some periods, significant variability has been recognized at regional scale in the 49 NASG (North Atlantic Subarctic Gyre) related to either warming or deep convection (Corbière et al., 50 2007; Metzl et al., 2010; Rödenbeck et al., 2014). The variations of winds would impact on air-sea CO₂ 51 fluxes (Wanninkhof and Trinanes, 2007), but at long-term they are mainly controlled by the rate of 52 change of ocean fCO₂ versus atmospheric concentrations. Based on a synthesis of pCO₂ observations for years 1972-2006, Takahashi et al (2009) evaluate a mean rate of 1.8 µatmyr⁻¹ (+/- 0.4) in the North 53 Atlantic, i.e. close to atmospheric increase, a result revisited and confirmed by McKinley et al. (2011) 54 55 for the period 1981-2009. An analysis of recent data (2005-2014) across the NASG near 59-60°N also suggests surface fCO_2 trends that are near the atmospheric increase (Fröb et al., 2018a). Interestingly, 56 57 this study also illustrates a spatial variety of the mechanisms for these trends. For example, Fröb et al. 58 (2018a) found onlyin the Iceland Basin a large contribution of alkalinity changes related to a salinity 59 decrease.

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The uptake of CO_2 through air-sea gas exchange affects the seawater CO_2 chemistry and leads to ocean acidification (Gattuso and Hansson, 2011). Over the past two decades, the pH in North Atlantic surface waters has declined at a similar rate as global ocean pH (-0.0018 yr⁻¹) (Lauvset et al., 2015; García-Ibáñez et al., 2016). However, in a similar way as for fCO₂, the pH interannual variability could be significant at regional scale. As an extreme case, based on the winter 2001-2008 SURATLANTdata, Metzl et al. (2010) reported a pH rate of -0.0069yr⁻¹ associated with particularly fast rise of oceanic fCO₂ (up to 7.2 µatmyr⁻¹).

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Finally, the large uptake of anthropogenic CO₂ in the ocean leads to a strong change in the isotopic composition of dissolved inorganic carbon (DIC) reducing thus its ${}^{13}C/{}^{12}C$ ratio (noted $\delta^{13}C_{DIC}$ hereafter). The $\delta^{13}C_{DIC}$ decrease in response to the human-induced perturbation is less than -0.007‰yr⁻¹ in polar surface regions (McNeil et al., 2001; Olsen et al., 2006) and reaches-0.025‰yr⁻¹ in subtropical regions (Gruber et al., 1999), observations which has been used to validate oceanic models for these regions (Tagliabue and Bopp, 2008; Sonnerup and Quay, 2012). This decrease is generally masked by the seasonal cycle due to physical and biological mechanisms, which is as large as





1% in the NASG (Gruber et al., 1999; Racapé et al., 2014). Although $\delta^{13}C_{DIC}$ provides additional information allowing us to further understand mechanisms of ocean region CO₂ uptake in key region as

78 NASG, its interannual to decadal variability is still poorly documented.

79

In order to unravel the surface NASG variability, it is necessary to revisit the region studied by (Corbière et al., 2007; Metzl et al., 2010; Racapé et al., 2014) with a more comprehensive data set. The 24years of physical and geochemical data presented here cover the recent multi-decadal sea-saw of warming/cooling and will provide an alternate view on the decadal variability in fCO₂ and pH in the central part of the NASG. The data set encompasses temperature (T), salinity (S), water stable isotopes, dissolved inorganic carbon (DIC), total alkalinity (A_t), nutrients, $\delta^{13}C_{DIC}$. fCO₂ and pH are computed from these data.

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The data have been binned in time and in latitude bands for later use. The data and the binning method are first presented (section 2). To illustrate the properties of the data set, we will then present the average seasonal cycle (section 3), as well as first order estimates of trends over the whole period (section 4).

92

93 2 Data and Methods

94 **2.1 Data**

95 Data were collected since 1993 along different container vessels operated or leased by EIMSKIP, 96 mostly between Reykjavik (Iceland) and Argentia (southern Newfoundland). The ships most 97 commonly crossed the North Atlantic in nearly 5 days first close to the Reykjanes ridge or to its west reaching the Newfoundland slope southwest of 49°N/49°W (AX02 transect on Fig.1). However 98 99 weather (late autumn to early spring) and ice conditions (January-April) often influence the ship's 100 route, so that it can also cross the central Irminger Sea or intersect meanders of the North Atlantic 101 Current (NAC), further to the southwest. The sampling used here is largely based on the collection of 102 surface samples collected by ship riders sent every three months. In addition, the ships have been 103 equipped a large part of the time with a SBE21 thermosalinograph (TSG)between April 1994 and May 104 2007 and a SBE45 micro TSG with an external temperature sensor SBE38 between February 2011 and 105March 2016, as well as by a fCO₂ equilibration system by NOAA/AOML (R. Wanninkhof, D. Pierrot) in 106 2003-2007 and 2013-2016 (data are in SOCAT database (Bakker et al., 2016) and at 107 http://www.aoml.noaa.gov/ocd/ocdweb/occ.html), both installed on a water circuit pumping water at 108 depths of 4-6m. Expendable bathythermograph (XBT) probes have also regularly been deployed





109 regularly along the AX02 transect (Fig. 1) (profiles from approximately 12000 temperature probes 110 collected between November 2008 and April 2016 at approximately 25 km resolution during these 111 transects are available on NOAA/AOML site http://www.aoml.noaa.gov/phod/hdenxbt/index.php). The 112 validation and correction of the TSG salinity data has been done by one of the authors, mostly based on 113 comparison with the water samples collected from a water intake at the TSG and also using nearby 114 upper level of Argo float data (see also, Alory et al., 2015). Except in April 1994-April 1996T was 115 measured at an intake. In July 1993 to January 1994 and in 2017, T was measured directly in the sea 116 from a bucket. At other times, it is T measured by the TSG that was adjusted to estimate ocean T, by 117 correcting warming in pipes based on comparison with 5-m temperatures from XBTs deployed along 118 the transects. The validated T/S data from the TSG are archived in the SSS repository at LEGOS 119 (www.legos.obs-mip.fr/SSS). In addition, we include data collected during one transect on the R.V. 120 Thalassa during the RREX2017 cruise in July/August 2017 in the same region, for which there were 121 intake temperatures associated with the TSG data.

122

123 The surface samples were usually collected from an intake corresponding to water pumped between 4 124 and 7-m depending on the vessel depth (except in 06/1993 and 01/1994, when water was collected with 125a bucket and its temperature measured by a calibrated thermometer). T is usually reported with the 126samples. Most commonly, it originates from the temperature measured by the TSG (and adjusted as 127 reported above) or from bucket T measurements. During some cruises, T values were complemented 128 using T from XBTs deployed as close to the time and location of those records as possible. In rarer 129 occasions, SSM/I satellite-derived microwave SSTs collocated at the ship time and position were used, 130 and for two transects in 07/2016 and 10/2016, upper-level T from nearby Argo profiles were also used as a proxy. Initially in 1993, the water collection included samples for S, DIC, δ^{18} O of sea water and 131 132inorganic phosphate. These were analysed at LDEO, but due to GR moving to France, the sample analysis was progressively discontinued (inorganic phosphate in late 1994, δ^{18} O of sea water in late 133 1341995, and DIC in February 1997), and then moved to other centres. Salinity sampling was never 135discontinued and S has been analysed since 2000 in Reykjavik (MFRI). DIC and At have been analysed 136at LODYC/LOCEAN in Paris since June 2001, and inorganic nutrients (nitrate+nitrite, phosphate, 137 silicic acid; later reported as NO₃, PO₄, Si) since December 2001 in Reykjavik (MFRI). Water stable isotopes of sea water (δ^{18} O and δ D) have been analysed since late 2011mostly at LOCEAN in Paris 138 139(some of these data are presented in Benetti et al. (2016), but with emphasis on the subset on Newfoundland shelf and slope). Finally, $\delta^{13}C_{DIC}$ has been analysed in 2005/2006 at the Univ. of 140

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141 Washington and since 2010 at LOCEAN (Racapé et al., 2014).Units are standard ones. Conversion was 142 done for nutrients from measurements in μ mol I⁻¹ into μ mol kg⁻¹assuming measurements conducted at 143 25°C. The water stable isotopes (δ^{18} O and δ D) are reported as concentration in VSMOW scale (Benetti 144 et al., 2017), whereas salinity is expressed as a practical salinity (on plots, psu is sometimes mentioned

- 145 to indicate that this is reported in the practical salinity scale).
- 146

147 Details on data collection, validation and accuracy are provided in App. A. Part of the DIC/A_t and 148 nutrient data (1993-2013) have been made available through the PANGEAE database (Reverdin et 149 al.,2007, 2015), whereas most of the water isotope data have been contributed to the GISS database.

150

151 The seawater CO₂ chemistry can be fully described with the measured DIC and A_t, using the 152dissociation constants of Lueker et al. (2000) as implemented in CO2SYS (Lewis and Wallace, 1531998; Pierrot et al., 2006). Estimating fCO₂ or pH from these equations also requires T and S data (see 154 above), as well as nutrients. When nutrient data are missing, we use the climatology derived from all 155the data at the calendar date of the sampling. The error in doing it has little impact on the computation. When no At was measured with DIC such as before 2001, we use parameterized At based on salinity. 156157 The strong correlation between sea surface alkalinity and salinity in the open ocean can be described with an empirical linear relationship (Millero et al., 1998; Friis et al., 2003). In our previous analysis 158159we used a relation based on seasonal SURATLANT data but only for years 2001-2002 (Corbière et al., 160 2007; Metzl et al., 2010)

- 161 At = $45.808 \times S + 713.5$, (r²=0.92, RMSD = $\pm 10.3 \mu molkg^{-1}$).
- 162

163 Here, an updated formula is derived for S > 34 based on all the reported SURATLANT data in 2001-

164 **2016**.

165 $A_t = 46.2543 * S + 687.74; (SSS > 34.0)(r^2 = 0.84, RMSD = \pm 8.4 \mu molkg^{-1})$

166 This relation is close to the one derived by Nondal et al. (2009) for the North Atlantic:

$$A_t = 49.35 * S + 582; (SSS > 34.5)$$

167

Our formula which has a larger 0-crossing explains a large part of the variance in A_t , at least for S larger than 34 psu (Fig. 2). For the lower salinities found on the Newfoundland shelf, different sources of freshwater (from the Arctic or resulting from continental or sea ice melt inputs) contribute to deviations from the relation. The impact on fCO₂ or pH of using this relationship instead of others or





172 measured A_t is discussed in App. B.

173

174Trends in sea water fCO_2 will be compared to trends in atmospheric fugacity. Atmospheric CO_2 mean 175mole fraction data were provided by the Cooperative Global Atmospheric Data Integration Project 176 (2016). Here, the xCO₂ data collected at Mace Head, Ireland (53.3° N) were used, after some editing 177 (mostly when wind from land). The xCO_2 data were converted to fugacities at 100% humidity 178following Weiss and Price [1980]. The xCO₂ trend at Mace Head is1.9 ppm/year in 1993-2016, 179 increasing to 2.1 ppm/year in 2006-2016 which is coherent with global average trends (e.g. Le Quéré et 180 al. 2018). Local values of atmospheric fCO_2 can present significant spatial differences at subpolar 181 latitudes depending on wind regimes, but that is unlikely to have had a strong influence on the long-182 term trends that are indicated here for reference purpose.

183

184 **2.2 Methods**

185 <u>2.2.1 Binning the time series</u>

186 T and S anomaly time series are presented as Hövmüller diagrams in App. C. The sample data 187 reproduce part of this long term variability, albeit with an insufficient sampling to properly separate the 188 interannual variability from longer term changes. The discrete sampling also results in uncertainties in 189 estimating a seasonal cycle, in particular for parameters such as $\delta^{13}C_{DIC}$ or $\delta^{18}O$ which have been 190 sampled for less time.

191

192 To reduce uncertainties in estimating an average seasonal cycle, we first remove from individual data an expected dependency in S estimated by linear regression for A_t, DIC, δ^{18} O and δ D (as done in Friis 193 194 et al., 2013). For A_t , this is the A_t -S regression mentioned above, and for water stable isotopes, as done 195in (Benetti et al., 2016). For nutrients, we normalize by S/35. We then also remove an average trend for DIC over 1993-2017 and for $\delta^{13}C_{DIC}$ over 2005-2017, as estimated from the whole data set (see section 196 197 4). Then, we bin the data in five 4° boxes from 46-50°N to 62-64°N roughly (see boxes on Fig. 3). The 198 southern box covers the shelf and slope area and incorporates only samples for which S < 34.1. The next 199 box from 50-54°N incorporates only samples with S between 34 and 35 to avoid including shelf water or from the North Atlantic Current (NAC). 54°N is near the shift in S variability observed in the data 200 201 (cf App. C), and thus separates box 2 from 3. We also remove data collected too far west in the 202 western Irminger Sea or central Labrador Sea as well as too far east in the western Iceland Basin 203 (altogether data from 6 transects). Then, in each box, data are binned by month and year. For a given





204 month and box, the individual annual means are averaged for years 2001-2017. The seasonal cycle 205 obtained is then further smoothed with a 1-2-1 smoothing average over successive months, as some of 206 the calendar months were sampled in very few years (worse for the southern shelf box in January-207 May). Rms deviations of samples from this 'smoothed' seasonal cycle are largest during the periods 208 with largest variability. Typically this happens in spring time (May-June) for parameters influenced by 209 phytoplankton blooms. When presenting the average seasonal cycle (renormalized so that it 210 corresponds to S=35, and for DIC and δ^{13} C with the trend added to correspond to 2010), we do not plot 211 the uncertainty resulting from the sampling. This uncertainty which is difficult to estimate properly (as 212 it is due both to interannual variability and variability within the same transect) is however usually 213 much smaller than the seasonal cycle amplitude portrayed.

214

215 <u>2.2.2 Estimating trends</u>.

216 Trends are estimated separately from the seasonal cycle, although the two are intertwined due to the 217 irregular time sampling. Earlier papers (Corbière et al., 2007; Metzl et al., 2010), mostly considered 218 trends in winter. Here, trends are based either on all data or on data collected in one season (without 219 removing the seasonal average). To minimize errors, either all data are considered, or only data in 220 50° N- 63° N. Alternatively, the deviations from the average seasonal cycle are estimated in each box, but we will not discuss them further here. This is done for DIC, A₁, fCO₂, pH, and $\delta^{13}C_{DIC}$ and for 221 222 different periods. Only the most characteristic trend estimates will be presented, for the sake of 223 characterizing the dataset and comparing them with results of other products and analyses.

224

225 **3** Seasonal cycle

In this section, we will present the average seasonal cycle portrayed in the 2001-2017 data. For T (not presented), one finds a seasonal cycle increasing from north to south, with amplitudes close to those portrayed in SST climatologies, such as HadSST3 (Kennedy et al., 2011). The seasonal cycle in S (Fig. 3) is also found as described at the surface in WOD13 (Boyer et al., 2013) with a maximum in February-May (February-April on shelf) followed by a gradual decline until a minimum in September (October, north of 62°N). The amplitude also increases from North to South.

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The DIC seasonal cycle presents (Fig. 3) a maximum in March and a minimum in July (North) transitioning to August (south). The amplitude is maximum on the Newfoundland shelf (73 μ molkg⁻¹), and north of 62°N (70 μ molkg⁻¹) and is smallest for 58-62°N (48 μ molkg⁻¹). There is a slight decrease





from March to April indicating the beginning of carbon consumption during blooms, and a very steep decline in April to May.

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At (Fig. 3) has much smaller seasonal variability than DIC (here expressed as deviation from average 239 relationship A_t-S), and estimated uncertainties of several µmol kg⁻¹ are large compared with the 240 seasonal cycle amplitude (uncertainties are larger on the shelf in January-May). After a minimum in 241 242 February, there is a peak in April-June, with April in the interior and later months north of 62°N and on 243 Newfoundland shelf, followed by a decrease until August north of 54°N (and later further south). The 244 increase could be associated with late winter or early spring blooms reducing NO_3 and PO_4 for the 245 formation of organic carbon, whereas the latter decrease could be associated with calcifying organisms 246 such as coccolithophores, which are known to produce large late spring or early summer blooms in this 247 part of the Atlantic, usually well past the large diatom-dominated blooms (Signorini et al., 2012)

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The $\delta^{13}C_{DIC}$ seasonal cycle (Fig. 3) updates Racapé et al. (2014), and provides further spatial resolution. It mirrors the cycle in DIC, with a maximum $\delta^{13}C_{DIC}$ in July in the north shifting to August further south (and even early September on the shelf) (and increase of 0.7‰ or more compared to late winter). The increase is associated with production of organic matter and the associated fractionation, whereas the later decrease is associated with remineralisation and vertical mixing. Notice also a small spatial gradient in winter with increasing values from north to south (and decreasing salinity).

255

256 The three nutrients present a March maximum (like DIC) associated with maximum entrainment of 257 subsurface water in the mixed layer, a slight decrease to April, and a later decrease until a July 258 minimum for Si and an August minimum for PO₄ and NO₃. This shift in the cycle is also suggesting of 259 a dominance of calcifying organisms in the later portion of the bloom, when Si levels have been 260 strongly depressed. For PO₄ and NO₃there is a north to south decrease of the nutrients (but not so much 261 for Si) in all seasons. The decrease is even stronger for NO_3 on the Newfoundland shelf, a sign of the 262 contribution of fresher water from the Pacific Ocean and western Arctic having experienced 263 denitrification on the shelves. Notice that summer NO_3 levels are very low in this average cycle south 264 of 54°N. The seasonal cycle of the three nutrients corresponds roughly to climatologies that are 265available from WOD13, and also to the data of the time series station in the Irminger Sea close to 266 Iceland (Olafsson et al., 2010).

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 δ^{18} O (and δ D) present no significant seasonal deviations from their average relationship with S north of 58°N. At 54-58°N, there is an early winter minimum and a maximum in April-May, during a period with overall small salinity seasonal variability. South of 50°N, in the shelf region, one finds a late summer maximum as described near the shelf break in Benetti et al. (2016), where it was related to some extent to sea ice melt water. 50-54°N presents an intermediate situation between the seasonal cycles in these two regions with a spring to September maximum

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275 Finally, the fCO₂ and pH seasonal cycles are estimated from monthly mean DIC and A_t in each box 276 (Fig. 4) Here, we do it without normalizing to S=35, in order to compare fCO2, pH, DIC seasonal 277 cycles with the climatology constructed by Takahashi et al (2014). Except for the southern region (46-278 50°N), fCO₂ presents a pronounced maximum in February – March associated with vertical mixing and 279 entrainment of remineralized DIC in the surface layer and a minimum in June associated with the 280 carbon use during the spring bloom. Similarly to DIC described above, the fCO_2 seasonality is most 281 pronounced in the north (amplitude 90 µatm) and in the south (80 µatm from January to May), where 282 the fCO₂ seasonal cycle exhibits secondary maximum (August) and minimum (November), as in the 283 climatology. In the central regions (54-58°N and 58-62°N) the seasonal fCO₂ amplitude is on the order 284 of 40 μ atm and we estimate similar fCO₂ levels for each month in these two boxes, as was also found 285 for DIC and nutrients (Fig 3). The oceanic fCO_2 are near (in December-March in the north) or well below the atmospheric level (Fig. 4a), the maximum ocean CO₂ sink being observed in the southern 286region in May (fCO₂^{ocean}-fCO₂^{atm} = -110 μ atm) a signal also picked up in climatology (Takahashi et al., 287 288 2009, 2014) and regularly observed in recent years during cruises conducted in May (5 cruises in 2004-289 2015, Wanninkhof and Pierrot, in Bakker et al., 2016). The pH seasonal cycle (Fig. 4b) mostly mirrors 290 (with reverse sign) the seasonal cycle in fCO_2 (Fig. 4a), and its amplitude ranges between 0.04 (in the 291 gyre) and more than 0.1 (in the south and north).

292

Examples of the comparison to the climatological seasonal cycle (Takahashi et al., 2014) are presented on Fig. 5. It shows the fCO₂, A_t, DIC and pH cycles for the two boxes in the central gyre (54-58°N and 58-62°N) where we observed homogeneous properties (Fig. 3). Recall that the climatology (Takahashi et al., 2014) for DIC and pH was calculated from pCO₂ and reconstructed alkalinity, and for reference year 2005, whereas our DIC seasonal cycle constructed with 2001-2017 data is referenced to year 2010 (section 2.2.a), and the corresponding fCO₂ and pH computed from this DIC and A_t. At 56°N (green lines), the fCO₂ (pH) climatology is low (high) compared to SURATLANT, but with rather similar

300 seasonality. At 60°N (blue lines), fCO₂ and pH seasonality are stronger in the climatology with a more

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301 pronounced fCO₂ minimum and pH maximum in June-July, but results are very similar in other seasons 302 (August-May). For both regions, the A_t seasonal amplitude is most pronounced in the SURATLANT 303 data (about 10-15 µmolkg⁻¹, Fig 5b) with largest difference in August. Despite these differences, the 304 DIC cycles derived from independent observations and methods are very similar (Fig 5c), which is 305 typical of all the regions.

306

307 4. Trends

308 Although significant variations have been previously observed and discussed for some periods (e.g., a rapid ocean fCO₂ increase up to $+7 \mu$ atmyr⁻¹ in winter over 2001-2008, Metzl et al., 2010), here we 309 show the trends over the full period (1993-2016) using all data or restricted to the band 50-63°N, and in 310 311 summer (June-September) or winter (January-March). In this band, At does not show any trend (Fig.6), as well as salinity or nutrients (not shown). On the opposite, we find positive trends for DIC ($\sim +0.7$ 312 µmolkg⁻¹yr⁻¹) in both summer and winter (notice no winter data included in 2016 or 2017) (Fig. 7). 313 314 This trend is about half the one reported by Olafsson et al. (2009), based on winter observations in the Iceland Sea for years 1985-2008 (+1.4 µmolkg⁻¹ yr⁻¹), but close to the trend observed over 1990-2015 315 in the Irminger Sea upper ocean waters (+0.63 µmolkg⁻¹ yr⁻¹), a signal mainly explained by 316 anthropogenic carbon uptake (Fröb et al., 2018b). As also suggested by ¹³C_{DIC} data (described below), 317 318 the positive DIC trends derived from SURATLANT data (Fig. 7) are likely due to anthropogenic CO₂. 319 Also, interestingly, in both seasons, the SURATLANT data seem to present no trend before 2005, 320 which corresponds to SST (and AMO) maximum. This lack of DIC trend in the early part of the record 321 was also commented in Metzl et al. (2010). This first part of the record also corresponds usually to a 322 period of decreasing winter winds (decrease in frequency of NAO+ situation), and thus an expected 323 decrease in winter mixed layer depths (and also a decrease in subpolar gyre circulation).

324

For fCO₂ (derived here from At, DIC, SST...; cf 2.1) positive trends are found as expected (Fig.8). In 325 the band 50-63°N, the summer (+2.1 µatm yr⁻¹) and winter (+1.67 µatmyr⁻¹) trends are close to the 326 327 atmospheric increase. This is in the range of the long-term trend (25-30 years) estimated in the North 328 Atlantic by Takahashi et al., (2009) and McKinley et al., (2011), but much lower than values near or above +3 µatm yr⁻¹previously reported for years 1993-2006 (Corbière et al., 2007; Schuster et al., 2009; 329 Metzl et al., 2010) and much larger than the $+1.1 \mu$ atmyr⁻¹trend estimated by Lauvset et al. (2015) for 330 331 years 1991-2011 in the NASG. This highlights that fCO₂ trend is quite sensitive to the period (and data) 332 selected (Fay and McKinley, 2013). Note that if one uses all data (all seasons and regions) the fCO_2 trend is +1.9 µatm yr⁻¹(Fig.9a) so that the delta-fCO₂ (difference between ocean and atmospheric 333





fugacities) present no significant trend (Fig. 9b) suggesting that in this region the air-sea CO₂ fluxes 334 335 driven by delta-fCO₂ (and winds) are relatively constant over time. Although temperature interannual variations (up to +1.5 °C in 2005, or -1.5°C in 2015, cf Fig. B1) could explain rapid fCO₂ changes for 336 some periods (Corbière et al. 2007), over the longer term the fCO_2 trends presented here are mainly 337 explained by DIC (At being relatively constant). The same is true for pH (Fig.10) and its negative trend 338 of -0.002 yr^{-1} mirrors the fCO₂ trend. Similarly to fCO₂ trend, this pH trend for the NASG is close to the 339 mean global ocean estimate of -0.0018yr⁻¹(Lauvset et al., 2015) and comparable to other trends 340 evaluated in the North Atlantic polar waters, ranging between -0.0017 yr^{-1} and -0.0024 yr^{-1} depending 341 342 on the periods, seasons and regions (Lauvset and Gruber, 2014; Lauvset et al., 2015; Olafsson et al., 343 2009).

344

For $\delta^{13}C_{DIC}$, we find significant trends in all seasons for the 2005-2017 period for which samples are 345 available. The winter trend (-0.015‰yr⁻¹) is smaller than the summer trend (-0.042‰ yr⁻¹) (Fig. 11). 346 347 Both are small compared to the seasonal cycle, and have large uncertainties, due to the small number of 348 years sampled. Notice for example that the summer season is during a time of large seasonal change 349 (Fig. 3). The trends seem of a similar magnitude since 2010, although in summer it is sensitive to the 350 large positive deviations of August 2010 data (see Racapé et al., 2014), with lower trends in 2010-2017 when not including it (-0.0040 versus -0.042‰ yr⁻¹). However, the trend estimates are sensitive to 351 uncertainties in the different corrections that we apply to the data before and after 2010 (see App. A3). 352 353 Altogether, the surface winter trend deduced from the adjusted data set is lower than the expected Suess effect in the atmosphere $(-0.025\% \text{yr}^{-1})$, based on current change rates in atmospheric pCO₂, such 354 as from the Alert station). It is also comparable with model estimates in the NASG for other periods 355 from Sonnerup and Quay (2012; -0.12‰ decade⁻¹over the period 1970-95) and from Tagliabue and 356 Bopp (2008; 0.10‰ decade⁻¹ between 1970 and 2005). 357

358

361

359 **Data availability**

- 360 The dataset is available at http://www.seanoe.org/data/00434/54517/
- 362 **5. Conclusions**

The SURATLANT data set in 1993-2017 is based mostly on discrete collection of surface samples (currently 2832 data points during 76 transects, but not always for all parameters). The collection methods and the parameters analysed have varied through the sampling, due to funding as well as logistical and analytical issues. We have documented data issues and have thus edited the data set. The





accuracy of the data is usually well documented (cf Annex A). Of course, we cannot guarantee that there are no remaining issue due for example to contamination of the water samples from pipes or water collection on board a ship, or due to storage in bottles before analysis in a laboratory. We derive a new A_t -S relationship adapted to the dataset that is used to estimate fCO₂ and pH, as well as other parameters of the carbonate systems, when A_t was not measured such as in 1993-1997. Thus estimated fCO₂ and pH are also provided with the dataset.

373

The sampling is found to be sufficient to document the average seasonal cycle of most parameters 374 375 analysed in this region providing a coherent data-set for processes analyses and/or biogeochemical 376 ocean models validation. It is also sufficient to document the long-term trends (1993-2017) in different 377 seasons. These trend estimates are provided to illustrate properties of the data set, and are found to be 378 in the bulk range of other studies. However, clearly, because of both interannual and decadal variability 379 (for example indication of evolution that is different in the warming period until 2005 than afterwards), 380 it is rather difficult to compare with other analyses, without further information on the variability, 381 either from other datasets or model simulations. Furthermore, the possibility of large spatial variability 382 in these signals could complicate the comparison. For example, Fröb et al. (2018a) document a large 383 spatial change in the recent decrease of SSS across the Reykjanes Ridge that implies (based on fCO₂ 384 observations) different decadal trends in DIC between the two regions.

385

386 In addition to this data set of discrete samples, other measurements have been made on the same ships 387 of opportunity throughout most of the 24 years of operation by continuous measurements. These 388 include near surface temperature and salinity from TSGs, and also at times pCO₂ measured with 389 equilibrator systems. Notice also that the information on instantaneous mixed layer depth and 390 stratification information was provided in a large part of the time by XBT profiles. These different data 391 sets (operated by NOAA/AOML and LOCEAN) will have to be incorporated to provide further 392 analysis of the discrete water samples. Validated versions of these complementary datasets can be 393 downloaded from the different sites mentioned in the paper. The investigation should also be based on 394 the compilations of stations data in GLODAP (Olsen et al., 2016) or the very rich SOCAT (Bakker et 395 al., 2016) data base. There is also a large array of complementary observations, such as from the Argo 396 and the bio-Argo (Organelli et al., 2017) data sets. For example, these data suggest blooms in mid-397 winter that could have a depletion impact on net production and export of nutrient and carbon from the surface layer already in March (Lacour et al., 2017). Notice however that the bio-Argo floats have 398 399 mostly sampled the rim of the subpolar gyre and provide only indirect evidence on near-surface carbon





- 400 and nutrient that they did not measure. Thus combining the different in situ cruise data sets with the
- 401 Argo data will provide other challenges.

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406 Appendix A: Data methods and uncertainties

407 We will review the different chemical parameters analysed, and discuss the uncertainties, as well as 408 anomalies observed.

409

410 A.1 Inorganic macro-nutrient (nitrate, phosphate, silicic acid) concentrations:

411 Samples for macro-nutrient concentrations were collected starting in late 2001 in pre-cleaned 250 ml 412 low density polyethylene bottles that were frozen on board (for two crossings, the samples were 413 probably not correctly frozen, and are discarded). For spring and summer samples, filtering was done 414 before measuring the macro-nutrient concentrations of nitrate (including nitrite), phosphate and silicate. 415 They were measured usually within 3 months of collection with standard colorimetric methods at the 416 Marine Research Institute (Reykjavik, Iceland). The analytical procedure and the quality control for 417 the nutrient analyses have been described in detail in Olafsson et al. (2010) where the long term accuracy has been estimated as $\pm 0.2 \text{ }\mu\text{mol }\text{kg}^{-1}$ for nitrate (includes also nitrite) and silicate, and ± 0.03 418 umol kg⁻¹ for phosphate. Additional uncertainties could result from contamination during collection, or 419 420 from poor conservation of the nutrients in the frozen samples. In particular this can result in occasional 421 elevated levels in phosphate, which are discarded when too far from neighboring samples with similar 422 T, S and DIC, or from the expected largest values. Values of phosphate were deemed too low in 423 January 2017, when it was found that most of the water had gone through some storage resulting into a 424 too large contribution of particulate phosphate to total inorganic phosphate.

425

Samples for phosphate in 1993-1994 were poisoned and analyzed at LDEO (Columbia Univ., New
York) shortly after the return of the water samples. Samples during RREX2017 (July/August 2017)
were pasteurized, kept cool and analyzed 5 months after the cruise at UMS IMAGO of IRD in Brest.

- 429 Samples from these different laboratories have not been inter-compared, and are reported as such.
- 430

431 A.2 Dissolved inorganic carbon and total alkalinity

Since June 2001, water samples collected on the ship are then shipped back to LOCEAN (Paris) where they are stored at 4°C and analyzed within three months of collection. No filtration is done, thus there might be a small amount of particulate inorganic carbon measured as well. DIC is determined at the same time as total alkalinity (A_t) by potentiometric titration derived from the method developed by Edmond (1970) using a closed cell and calibrated Certified Reference Material (CRM) provided by A.

437 Dickson (Scripps Institution of Oceanography, San Diego, USA). Analytical accuracy of the DIC and





At concentrations is $\pm 3 \mu mol \text{ kg}^{-1}$ (further details in Corbière et al. (2007)). Most of the bottles used at 438 439 LOCEAN since 2003 were 500 ml round bottles with screw caps (before that and for a few isolated 440 samples since 2003, the 500 ml bottles had a ground glass stopper and Apiezon grease was used). For 441 some of these bottles, we found that samples presented systematic biases in 2010-2015, either in DIC 442 or At. This led to the replacement of some bottles in July 2015, and after that, most bottles are from a 443 newer set. Since 2005, poison (saturated (or half saturated) mercuric chloride solution) volumes were 444 kept to 0.3 ml, and no correction was applied for dilution. In earlier years, poison solution volume has evolved, and data reported are usually corrected for the dilution effect. For the period 06/1993-02/1997, 445 446 the samples were stored in bottles with ground glass stopper with use of Apiezon grease, and DIC was determined by a coulometric method at LDEO (Chipman et al., 1993). For some samples in 2005-2006 447 associated with a $\delta^{13}C_{DIC}$ value, DIC was also estimated manometrically during acid CO₂ extraction 448 procedure with somewhat lesser accuracy ($\pm 5 \mu$ mol kg⁻¹). We used these DIC values when there was 449 450 no DIC estimated at LOCEAN.

451

452 We first edited the data to remove suspicious values. In particular, in some instances, errors seemed to 453 have resulted from poor sample conservation in the bottles before analysis. For example, we often noticed erroneously large DIC and At values from bottles 8 (103-mid 2015) and 19 (08/2010 to mid-454 455 2015). There are also a few transects when samples had to be discounted. In one case (April 2007) this 456 happened because the samples were stored in a very hot space before shipment to France. For April 457 2015 DIC values were too high for unknown reasons (but possibly, the bottles had aged), and for 458 January 2017, both DIC and At were discounted as the origin of the water collected was suspicious (problem also encountered for salinity samples and with the phosphate, with an anomalous contribution 459of particulate phosphate). In April 2015, we adjusted the DIC values by -10.3 µmol kg⁻¹, based on 460 comparisons of estimated fCO₂ with those directly measured that are in the AOML and SOCAT data 461 462 base.

463

The LOCEAN DIC values have been compared for some crossings with DIC samples collected at the same time and analyzed in other laboratories(in 2005, 2006, 2010, 2015, 2016 and 2017, that altogether involved four laboratories). These comparisons summarized in Table A1 (average and rms standard deviation)reveal for individual transects LOCEAN DIC often lower by 5 μ mol kg⁻¹ or more, but this is far from being systematic. Very often, dilution by the mercury chloride solution is taken into account by other groups, but not at LOCEAN, which could contribute to up to 2 μ mol kg⁻¹ difference (LOCEAN DIC lower).There can also be issues of poor conservation of some of the other water





471 samples and uncertainties in their analysis, so this is not an absolute validation of the LOCEAN values. 472 Usually, the comparisons for A_t (in 2010, 2015 and 2016) suggest smaller average differences. For A_t 473 there were other comparisons of samples collected in the same region and analyzed both at LOCEAN 474 and ICM/CSIC during the Ovide cruises (every 2 years between 2002 and 2016) but with a different set 475 of LOCEAN bottles than for SURATLANT. They suggest a similarly close agreement between A_t analyzed in the two laboratories (for example, average difference of -1.4 (σ =3.4 n=57) µmol kg⁻¹ for 476 477 the 2014 cruise). A recent international inter-comparison on two shared water batches (spring 2017) suggests that the LOCEAN analysis presents a small negative bias both for A_t and DIC, but not in a 478 479 very similar range of values to the ones observed during SURATLANT.

480

481 Whether these results are relevant for the whole dataset will need to be further ascertained. We can nonetheless expect that the DIC and At reported in the SURATLANT data set have uncertainties 482 always smaller than 10 µmol kg⁻¹, that they can present biases at time, but that they are probably of a 483 higher accuracy most of the time. To provide further validation, we take advantage that during some 484 485 transects, sea surface underway fCO2 measurements were also conducted following Pierrot et al., 486 (2009) and data available at AOML (http://www.aoml.noaa.gov/ocd/ocdweb/occ.html) and regularly 487 submitted and qualified in SOCAT (Bakker et al, 2016). We have collocated these data (within 2 488 minutes time of discrete sampling) and compared them with fCO2 calculated from DIC, At pairs. For 489 this comparison we only use samples when DIC and At were measured (not At derived from salinity if At is missing). A total of 172 points have been collocated for different years (2004-2007 and 2014-490 2015), for almost all seasons (months: Jan, Feb, Apr, Jun, Jul, Oct, Nov and Dec) and those data 491 492 represent a large fCO2 range (225-420 µatm, Figure A1a). We find both positive and negative 493 differences (Figure A1b) that are not associated to years, seasons, SST, DIC or At concentrations. The mean difference (fCO2calc-fCO2mes) of -3.6 (+/- 12.4) µatm is thus attributed to methods 494 495 uncertainties (including sampling time, measurements errors and data processing). These new results 496 (mean and deviation) are in the same range as obtained in previous comparisons (Metzl et al., 2010) but for different, fewer data and different constants (n=54, mean difference= -2.3 (+/- 11.1) µatm). We thus 497 conclude that fCO2 (and pH) calculated here with discrete DIC, At data are suitable to interpret both 498 499 seasonality and trends.

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- 501

502 A.3 δ¹³C_{DIC}:

503 Over the period 2005-06, acid CO₂ extraction was done for $\delta^{13}C_{DIC}$ measurements from helium





504 stripping technique. This analytical method has been described previously by Quay and Stutsman (2003). These measurements have an accuracy of $\pm 0.02\%$ for $\delta^{13}C_{DIC}$ based on a helium stripping 505 technique adapted from the one used by Kroopnick (1974) and ±5µmol kg⁻¹ for DIC based on a 506 comparison to coulometric DIC values and to Certified Reference Material (CRM) provided by A. 507 508 Dickson (Scripps Institution of Oceanography, San Diego, USA). However, the DIC of these samples were higher by an average 5 µmolkg⁻¹than the LOCEAN DIC. If this was caused by a small 509 510 remineralization of dissolved organic DIC, and based on the relationship described in Racapé et al (2014), this could be associated with a negative bias of -0.05% in $\delta^{13}C_{DIC}$. Thus, we chose to correct 511 512these pre-2006 data by 0.05%.

513

514 Over the period 2010-17 during most of the SURATLANT cruises, $\delta^{13}C_{DIC}$ were measured by mass 515 spectrometry via an acid CO₂ extraction method in a vacuum system developed by Kroopnick (1974) 516 whereas further details on the sampling methods and analytical techniques are provided in Racapé et al. 517 (2014) for $\delta^{13}C_{DIC}$. Water was collected in small glass bottles poisoned by at least 1 ml of saturated 518 solution of mercuric chloride, and stored in the dark when possible at 4°C (at least, after return to the 519 lab, when it was stored for up to a year before analysis). Data of some crossings were dismissed, 520 probably because poisoning had been insufficient, and for one crossing the mass spectrometer did not 521 function properly. Other isolated samples were dismissed either because not enough gas was collected 522 after acidification or due to possible leaks on the mass spectrometer gas lines. These $\delta^{13}C_{DIC}$ values are expected to have a precision of $\pm 0.01\%$ (Vangriesheim et al., 2009) and a reproducibility of $\pm 0.02\%$. 523 Issues on the accuracy of other $\delta^{13}C_{DIC}$ samples from LOCEAN have been raised, and LOCEAN 524 525 participated to an inter-laboratory comparison run by C. Normandeau (Dalhousie University) with deep NASG water samples conditioned by Dahlousie Univ. The results suggest that recent LOCEAN 526 527 samples have a slightly poorer reproducibility ($\pm 0.04\%$) than earlier ones. These comparisons suggest 528 an average bias of LOCEAN measurements of -0.13‰. This is less than the -0.25‰ (0.20 to 0.30‰) 529 bias corrected in the GLODAP database to LOCEAN samples collected during North Atlantic cruises 530 in 2002 and 2006and analyzed with the same method and standards as here. This adjustment was based 531 on the intercomparison of different cruises (Becker et al., 2016), which could nonetheless include a part 532 due to anthropogenic signal. We chose to adjustall the LOCEAN values (samples collected in 2010-5332017) by +0.13‰.

534

535 A.4 Water isotopologs:

536 Since 12/2011, the oxygen isotopic composition of discrete sea water samples has been usually





537 analyzed with a PICARRO CRDS (cavity ring-down spectrometer; model L2130-I Isotopic H₂O) at 538 LOCEAN-IPSL (Paris, France). The internal references which have been used to calibrate the data in 539 the V-SMOW scale, have been previously calibrated using IAEA references and vary from -6.61% to 2.24 ‰ for δ^{18} O and from -44.3 ‰ to 3.31 ‰ for δ D.All reference waters are stored in steel bottles 540 541 with a slight overpressure of dry nitrogen to avoid evaporation processes and exchanges with ambient 542 air humidity. Based on repeated analyses of an internal laboratory standard over several months, the accuracy of the measurements is usually better than ± 0.05 % for δ^{18} O and ± 0.50 % for δ D. All sea 543 544 water samples before 04/2016 have been distilled to avoid salt accumulation in the vaporizer and its 545 potential effect on the measurements (e.g., Skrzypek and Ford, 2014). Between 04/2016 and 07/2016, 546 as well as since May 2017, the analysis has been often done without distillation and with a salt trap 547 mesh. Samples in 09-10/2016, 01/2017, as well as a few samples in 05/2017 and 08/2017 samples were 548 instead analyzed on a GV Instruments Isoprime dual inlet IRMS coupled with Aquaprep sample 549 preparation system (at IES, Reykjavik, Iceland). The different methods are fully described in Benetti et al. (2017). All samples in 1993-1995 were analyzed on a similar Isoprime dual inlet IRMS system but at 550 LDEO. Most of the samples done on IRMS were only run for δ^{18} O. A recent study (Benetti et al., 2017) 551 552 suggests that different corrections have to be applied on the data depending on the method used to 553 report the data in 'absolute concentration scale'. We follow their recommendations and adjusted data to 'absolute concentration scale', except for the following cases. For the δ^{18} O values of samples in 1993-554 555 1995, for which the information on the internal standard used has been lost (although it probably was 556 deep Atlantic Ocean water), we assume that they are already reported in concentration scale, and apply 557 no correction. For the 12/2011 section, samples were analyzed either by the PICARRO CRDS (but without distillation), or at LOCEAN with a GV Instruments Isoprime dual inlet IRMS coupled with 558 Aquaprep sample preparation system (with δ^{18} O adjusted to the PICARRO measurements), and the 559 accuracy of these samples is not as high (probably closer to ± 0.10 % for δ^{18} O and ± 1.00 % for δ D). 560 561 Finally, for data of a 2017 run with a different salt trap mesh, we found a slight additional bias that was 562 also corrected. 563

- - - -





565 Appendix B: At-S relationship and resulting uncertainties in estimating fCO₂ and pH

566 As explained in 2.1, we constructed anA_t -S linear relationship by least square fitting on the SURATLANT data (2001-2016) for S larger than 34, mixing all seasons. The R² correlation coefficient 567is 0.83 and rms deviations of 8.4 μ mol kg⁻¹ (for S larger than 34), which is larger than the uncertainty 568 on the measurements. We also suggested when describing the seasonal cycle of A_t in section 3 that part 569 570 of the scatter could be due to seasonal variability, and certainly there is a range of variation expected 571 for this relationship. Here, we will document what effect the choice made has when estimating fCO_2 or 572 pH from DIC, SSS, SST, nutrient data, when A_t was not measured such as in 1993-1997. In particular, 573 it is interesting to estimate how it affects trends, as done in section 4. 574

- First, we compare the computed versus measured A_t (Fig. B1). They present differences that tend to be correlated over a year or more, such as the lower computed values in 2001-early 2005, 2011-2012 or part of 2015. On the other hand there is no significant trend in the difference between the two during the measurement period from 2001 to 2017. Not surprisingly, the same can be said for computed fCO₂ or pH (Fig. B1). Therefore the conclusions on the long-term trend for fCO₂ and pH are valid using A_t , either from measurements or A_t/S relation, but for short-term and processes analyses measured A_t should be used.
- 582

We then compare the alkalinity estimated with the SURATLANT relationship with the one derived with the relationship by Nondal et al (2009) which has a steeper slope. We also show the earlier fit of Corbière et al (2007), which result in larger A_t, and a fit on all the SURATLANT data, including the shelf data for S< 34 (Fig. B2). The Nondal et al (2009) relation underestimates A_t for low S, but is rather well within the data spread near S=34 – 35 psu. The Corbière et al. (2007) relationship overestimates A_t at all salinities, and will not be considered later on.

- 589
- 590 We then compare fCO_2 and pH computed with the two best estimated $A_t(S)$ which seem to cover the
- 591 possible range of relationships. When applied on all the individual data, the difference (fCO₂ using the
- 592 SURATLANT relationship minus fCO₂ using the Nondal et al (2009) relationship) appears as a spread
- 593 for S near 35, as well as for the low S on the shelves (Fig. B3). For all samples for S>34, the mean
- 594 difference is 3.8 (+/- 1.75) µatm, i.e. lower than uncertainty associated to fCO2 calculations. As overall
- changes in S are not that large over the 24 years (cf App. C) these two At/S relations, originally based
- 596 on different data-sets, lead to same conclusion for the fCO₂ trend (1.7-1.8 μ atm yr⁻¹) that is close to the





597 trend in the atmosphere. The impact on the trend in pH is also not significant.

598

599





601 Appendix C: T and S time series in the subpolar gyre

- 602 We also estimated monthly binned temperature and salinity time series (smoothed 1-2-1 over 603 successive months) along a standard ship route since mid-1993 (B-AX02 transect between Iceland and 604 southern Newfoundland) which is close to where most samples were collected (Fig. 1). Most data 605 originate from ship-mounted thermosalinographs with additional data from XBTs, CTDs, instrumented 606 drifters and floats (Argo floats or the earlier Palace floats). The data qualification, processing and how 607 the gridded time series are produced, is reported in Reverdin et al. (2018). To summarize the main 608 steps: an average seasonal cycle at 1° resolution is first removed from individual data, and anomalies 609 are then grouped in bins along the ship track on a monthly time scale. Gaps in the time series are filled by first linearly interpolating from neighbouring spatial bins, and then in time from neighbouring time 610 611 steps (with a further 1-2-1 running average on the monthly anomaly time series). Time series along the 612 AX02 transect start in July 1993 with monthly few short gaps (the largest gaps are found in winter over 613 the Newfoundland shelf and slope). These time series are used to check whether the discrete sampling 614 presented in this paper has accounted for the interannual/decadal variability portrayed in this time 615 series. They are also helpful in estimating the domains over which the hydrographic variability presents 616 some coherence.
- 617

618 Hovmüller diagrams of T and S along AX02 are presented on Fig.C1 (left panels). Along this transect, 619 bins correspond usually to 1° latitude range, with two wider bins on the shelf between southern 620 Newfoundland and the shelf break, and two bins that correspond to the Newfoundland shelf break and 621 slope. The T and S AX02 time series usually present large correlation between successive seasons 622 (correlation coefficient higher than 0.6), except for the two time series on the Newfoundland shelf. The 623 S variability along AX02 is very coherent in latitude from the close vicinity of Iceland to 54°N, and 624 after a strong increase in 1996 present oscillations at 4-10 years period, before a recent decrease in 625 2016-2017. The transition at 54°N corresponds to North Atlantic Current frontal position further east 626 and where the line stops paralleling the western flank of the Reykjanes Ridge. Further south over the 627 deep ocean (until 49°N/49°W), variability in S is much larger, but not similarly phased to what is 628 observed further north, with some suggestion of a lead-time of one to two years. Variability is different 629 on the Newfoundland shelf and less coherent between successive seasons. The data sampling there is 630 not always sufficient to be correctly portrayed in this analysis, in particular due to occasional winter or early spring ice cover, in particular in 1994-1995 and in 2014-2016. Still, it seems also to indicate 631 632 negative low frequency anomalies until 2000 and between 2010 and 2015, and more positive in 633 between, as is found further offshore.





634

T is not well correlated to S, in particular at seasonal to interannual periods, but the time series are too short to identify whether correlation increases at lower frequencies. There is a large meridional coherency in the signals at least north of 50°N. This clearly resembles the subpolar gyre scale AMO index or average temperature, such as presented for example in Robson et al. (2016). There is the clear swing from negative T-anomalies before 1996, and again in 2000, to maximum positive anomalies in 2004-2007 or 2010, followed by more negative anomalies (as seen in Robson et al., 2016) that have amplified in 2014-2016.

642

643 We compared the winter S anomalies from the binned analysis (B-AX02) with the deviations from the 644 average seasonal cycle from the discrete water samples of the SURATLANT data set. The comparison 645 is very encouraging, as illustrated by the salinity plots at 60°N (Fig. C2). Clearly the SURATLANT 646 reduced sampling is able to capture the largest signals in salinity (and in temperature) and thus in 647 surface water masses, that the ocean witnessed. This holds also to a good extent in other seasons, and 648 less so on the Newfoundland shelf, where variability tends to be more high frequency. Notice though 649 that there are inter-annual differences. For example, SURATLANT would describe (at this latitude), 650 early 1997 as anomalously fresh, whereas B-AX02 shows near-normal salinity conditions.





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Author contribution 654

655 GR has remained the prime coordinator of the project since it was initiated in 1993 and prepared the 656 manuscript with contributions from all co-authors. NM has been co-Pi on most of the proposals since 657 2000 and has contributed to the collection, qualification and validation of large parts of the data set. SO 658 has been in charge of the nutrients since 2001 as well as intercomparison of DIC/At samples, and is the main Icelandic PI associated to the project. VR has been in charge of the validation of the $\delta^{13}C_{DIC}$ data. 659 TT has contributed to the initiation of the project in 1993, and has produced the DIC and PO₄ data 660 661 before 1997. MB has been in charge of the validation of the water stable isotope data. HV has been 662 associated in Iceland to the initiation of the project and has supervised the analysis of salinity samples since 2000. ABC has performed the analysis of nutrient data since 2010. MD has analyzed the salinity 663 664 samples since 2001. JF has performed and qualified the DIC/A_t analyses since 2014. AN has analyzed many $\delta^{13}C_{DIC}$ and water stable isotope samples. 665

667 Acknowledgments

668 The authors declare that they have no conflict of interest. Access on the merchant vessels run or leased 669

by EIMSKIP has been the core for this long effort to maintain surface sampling between Iceland and 670 Newfoundland. The nearly one hundred volunteer ship riders and their enthusiasm have been key to the

success of this monitoring. The project was initiated when one of the authors, GR, was at LDEO, with 671

672 initial support from this institution. NOAA/AOML and NOAA/CPO Ocean Observing and Monitoring

673 Division have contributed by maintaining the thermosalinographs and providing XBTs on the different

674 ships that have operated between Iceland and Newfoundland. The French effort was supported by

675 various agencies throughout the years, and in particular INSU (for SNO SSS and by the LEFE/CYBER

676 grant CO2SINK since 2016). Support by National Power Company of Iceland Landsvirkjun is

acknowledged. The $\delta^{13}C_{DIC}$ sampling was initiated by Paul Quay (Univ. of Washington, Seattle) in 677

2005-2006. SNAPO-CO2 is acknowledged for analyzing DIC/At at LODYC/LOCEAN since 2001, and 678 the isotopic platform of OSU Ecce Terra for analyzing $\delta^{13}C_{DIC}$ as well as water stable isotopes at

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1022 1023	Figure and table captions Table A1 : Comparisons of DIC and At samples shared between LOCEAN and other institutes (1, 2, 3, 4). Institute 1 was a manametric method for massing DIC institutes 2, 2 and 4 was a potentiametric
1024 1025 1026	4). Institute 1 uses a manometric method for measuring DiC, institutes 2, 5 and 4 use a potentiometric method for DIC, whereas institutes 3 and 4 use a potentiometric method for A_t . The different columns are for institute number, months and year of sampling, number of samples, average and rms diff
1027 1028 1020	(LOCEAN-other) first for DIC, then for A_t . Figure 1 : monitoring of the surface subpolar sure SUP ATLANT is along ship track AY2. The red
1023 1030 1031 1032 1033	currents indicate a schematic view of surface circulation in the subpolar gyre originating from the Gulf Stream and North Atlantic Current. The blue lines indicate the path of the freshest waters of polar or continental origin, and the purple arrows illustrate the deep export circulation.
1034 1035 1036 1037 1038	Figure 2 : Left panel spatial distribution of the A _t samples collected in 2001-2016 and used to estimate a regression to salinity (for S > 34 psu) (color corresponds to A _t value). Right panel: A _t -S dispersion diagram. The red line corresponds to A _t =46.2543*S+687.741 best fit (with rms = 8.4 µmol kg ⁻¹ , R ² =0.84 and n=1332 for S> 34)
1039 1040 1041 1042 1043 1044	Figure 3 : Average seasonal cycle in 2001-2017. For DIC, A_t , $\delta^{18}O$, PO_4 , NO_3 , Si, it is normalized for S variations and reported at S=35. The colour of the curves correspond to the geographical boxes presented in upper right corner (yellow, red, green, blue, black from south to north; the black trajectory correspond to January 2017 sampling). Yellow curves (south) for S and $\delta^{18}O$ and DIC have been shifted. A map with the location of the boxes is also provided.
1045 1046 1047 1048	Figure 4 : Seasonal cycles of fCO2 (a) and pH (b) in each box. On (a) the purple dashed-line is the mean monthly atmospheric fCO2 derived from CO2 concentrations at Mace-Head station for year 2010.
1049 1050 1051	Figure 5 : Seasonal cycles of fCO_2 (a), A_t (b), DIC (c) and pH (d) derived from SURATLANT data and the climatology (Takahashi et al., 2014) for the central region (boxes 54-58°N and 58-62°N).
1052 1053 1054 1055 1056	Figure 6 : Time-series of A_t observed in the latitudinal band 50-63°N. For 1993-1997 A_t was derived from Salinity. The brown line depicts the long-term trend (+0.02 µmol kg ⁻¹ yr ⁻¹), i.e. no trend detected for A_t . Low A_t values (< 2250 µmol kg ⁻¹) observed near 50°N. The color corresponds to calendar month (right scale).
1057 1058 1059 1060	Figure 7 : Time-series of DIC observed in the latitudinal band 50-63°N for winter (January-March, left) and summer (June-September, right). The dashed lines depict the long-term trend (+0.72 μ mol kg ⁻¹ yr ⁻¹ in winter, +0.67 μ mol kg ⁻¹ yr ⁻¹ in summer,). The color corresponds to calendar month (right scale).
1061 1062 1063 1064	Figure 8 : Time-series of fCO ₂ observed in the latitudinal band 50-63°N for winter (January-March, left) and summer (June-September, right). The dashed lines depict the long-term trend (+1.67 μ atm yr ⁻¹ in winter and +2.1 μ atm yr ⁻¹ in summer). The color corresponds to calendar month (right scale).
1065 1066 1067 1068	Figure 9 : Time-series of fCO ₂ (left) and delta-fCO ₂ (right) for all SURATLANT data. The brown lines depict the long-term trends (+1.9 μ atm yr ⁻¹ on left panel, -0.04 μ atm yr ⁻¹ on right panel). The color corresponds to calendar month (right scale).
1069	Figure 10: Time-series of pH SURATLANT data. The brown line depicts the long-term trend (-0.002





1070 1071	yr^{-1}). Note high pH (> 8.2) observed in coastal regions (north or south). The color corresponds to calendar month (right scale).
1072	eulendul month (right souro).
1073	Figure 11 : trend for $\delta^{13}C_{DIC}$, left panel, data distribution (2005-2017); right, time series with trend for
1074	summer season (red dashed line: trend -0.042 yr^{-1}), trend for winter season(blue dashed line, trend, -
1075	0.015 yr^{-1}), all seasons (brown dashed line, trend -0.022 yr ⁻¹) (notice that data have been adjusted
1076	by+0.05 in 2005-2006 and by +0.13 since 2010). The color on left panel corresponds to $\delta^{13}C_{DIC}$ value
1077	and on right panel corresponds to the calendar month (right scale).
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1079	Figure A1: (a) fCO2 calculated versus fCO2 measured (µatm) for 172 colocated samples. Dashed line:
1080	fCO2-SUR= 1.05 fCO2-AOML (r2=0.9) (b) fCO2 differences versus fCO2 measured for same samples
1081	(µatm)
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1083	Figure B1: Comparison of using calculated A _t (with SURATLANT relationship) with using measured
1084	At. Left panel, difference in At (calculated – measured), middle panel difference in fCO2 and right
1085	panel, difference in pH.
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1087	Figure B2 : scatter diagram of SURATLANT At versus S. Different linear fits are also presented. The
1088	blue line corresponds to the relationship adopted in this study and the red line to the Nondal et al. $(2000) = 1 + \frac{1}{1000} = 1$
1089	(2009) relationship.
1090	Figure P3: Difference in fCO2 estimated when using A derived from S with SUDATI ANT
1091	Figure D5. Difference in ICO2 estimated when using A_t derived from S with Nordal et al. (2000) relationship. Plot of
1092	diff($f(\Omega)$) as a function of salinity for all SURATIANT samples
1094	un(1002) as a function of saminty for an SORATEARY samples.
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1096	Figure C1 : Hövmüller diagram along AX2 (left, Newfoundland; right, Iceland) of S' (left, psu) and T'
1097	(right, °C) deviations from an average seasonal cycle in July 1993-June 2017.
1098	
1099	Figure C2: January-April salinity deviations from the seasonal cycle near 60°N. In red, from the
1100	monthly analysis (see Fig. C1), and in blue from the discrete salinity samples (for those, the analysed
1101	seasonal cycle is presented on Fig. 3).
1102	





- 1103 Table A1: Comparisons of DIC and At samples shared between LOCEAN and other institutes (1, 2, 3,
- 4). Institute 1 uses a manometric method for measuring DIC, institutes 2, 3 and 4 use a potentiometric
- 1105 method for DIC, whereas institutes 3 and 4 use a potentiometric method for A_t . The different columns
- are for institute number, months and year of sampling, number of samples, average and rms diff
- 1107 (LOCEAN-other) first for DIC, then for A_t .
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institute	Month/year	N samples	Average	Sigma	N samples	Average	sigma
			diff			Diff	
1	01/2005-	115	-4.1	6.8			
	11/2006						
2	6-8/2010	15	-7.4	3.5			
3	8/2010	9	-5.4	3.1	10	-1.5	4.1
4	8/2010	9	-5.9	4.0	10	1.4	4.8
3	1/2015	8	-6.8	3.6	8	0.0	7.3
2	4-10/2016	8	-1.2	5.3			





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1131	Figure 1: Monitoring of the surface subpolar gyre. SURATLANT is along ship track AX02. The red
1132	currents indicate a schematic view of surface circulation in the subpolar gyre originating from the Gulf
1133	Stream and North Atlantic Current. The blue lines indicate the path of the freshest waters of polar or
1134	continental origin, and the purple and grey arrows illustrate the deep export circulation.







- 1162 and n=1332 for S> 34)
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Figure 3: Average seasonal cycle in 2001-2017. For DIC, A_t , $\delta^{18}O$, PO_4 , NO_3 , Si, it is normalized for S variations and reported at S=35. The colour of the curves correspond to geographical boxes presented in upper right corner (yellow, red, green, blue, black from south to north; the black trajectory correspond to January 2017 sampling). Yellow curves (south) for S and $\delta^{18}O$ and DIC have been shifted. A map with the location of the boxes is also provided.





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1256	Figure 4 : Seasonal cycles of fCO ₂ (a) and pH (b) in each box. On (a) the purple dashed-line is the

Figure 4: Seasonal cycles of fCO₂ (a) and pH (b) in each box. On (a) the purple dashed-line is the
 mean monthly atmospheric fCO₂ derived from CO₂ concentrations at Mace-Head station for year 2010.

















1337Figure 6: Time-series of At observed in the latitudinal band 50-63°N. For 1993-1997 At was derived1338from Salinity. The brown line depicts the long-term trend (+0.02µmol kg⁻¹ yr⁻¹), i.e. no trend detected1339for At. Low At values (< 2250 µmol kg⁻¹) observed near 50°N. The color corresponds to calendar month1340(right scale).







13631364Figure 7: Time-series of DIC observed in the latitudinal band 50-63°N for winter (January-March, left)1365and summer (June-September, right). The dashed lines depict the long-term trend (+0.72 μ mol kg⁻¹ yr⁻¹1366in winter, +0.67 μ mol kg⁻¹ yr⁻¹ in summer). The color corresponds to calendar month (right scale).







1392Figure 8: Time-series of fCO_2 observed in the latitudinal band 50-63°N for winter (January-March,1393left) and summer (June-September, right). The dashed lines depict the long-term trend (+1.67 µatm yr⁻¹1394¹ in winter and +2.1 µatm yr⁻¹ in summer). The color corresponds to calendar month (right scale).

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1420Figure 9: Time-series of fCO2 (left) and delta-fCO2 (right) for all SURATLANT data. The brown lines1421depict the long-term trends (+1.9 μ atm yr⁻¹ on left panel, -0.04 μ atm yr⁻¹ on right panel). The color1422corresponds to calendar month (right scale).







1459Figure 10 : Time-series of pH SURATLANT data. The brown line depicts the long-term trend (-0.0021460 yr^{-1}). Note high pH (> 8.2) observed in coastal regions (north or south). The color corresponds to1461calendar month (right scale).







Figure 11: trend for $\delta^{13}C_{DIC}$. left panel, data distribution (2005-2017); right, time series with trend for summer season (red dashed line: trend -0.042 yr⁻¹), trend for winter season(blue dashed line, trend, -0.015 yr⁻¹), all seasons (brown dashed line, trend -0.022 yr⁻¹)(notice that data have been adjusted by+0.05 in 2005-2006 and by +0.13 since 2010). The color on left panel corresponds to $\delta^{13}C_{DIC}$ value and on right panel corresponds to the calendar month (right scale).





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Figure A1: (a) fCO₂ calculated versus fCO₂ measured (μatm) for 172 colocated samples. Dashed line:
 fCO₂-SUR= 1.05 fCO₂-AOML (r2=0.9) (b) fCO₂ differences versus fCO₂ measured for same samples
 (μatm)

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Figure B1: Comparison of using calculated At (with SURATLANT relationship) with using measured
 At. Left panel, difference in At (calculated – measured), middle panel difference in fCO₂ and right
 panel, difference in pH.







Figure B2: scatter diagram of SURATLANT A_t versus S. Different linear fits are also presented. The
 blue line corresponds to the relationship adopted in this study and the red line to the Nondal et al.
 (2009) relationship.











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1620 (right, °C) deviations from an average seasonal cycle in July 1993-December 2017.







- 16231995 2000 2005 2010 20151624Figure C2: January-April salinity deviations from the seasonal cycle near 59°N. In red, from the
- monthly analysis (see Fig. C1), and in blue from the discrete salinity samples (for those, the analysedseasonal cycle is presented on Fig. 3).
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