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A high-frequency atmospheric and seawater $p\text{CO}_2$ data set from 14 open ocean sites using a moored autonomous system

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Abstract

In an intensifying effort to track ocean change and distinguish between natural and anthropogenic drivers, sustained ocean time-series measurements are becoming increasingly important. Advancements in the ocean carbon observation network over the last decade, such as the development and deployment of Moored Autonomous $p\text{CO}_2$ (MAPCO₂) systems, have dramatically improved our ability to characterize ocean climate, sea–air gas exchange, and biogeochemical processes. The MAPCO₂ system provides high-resolution data that can measure interannual, seasonal, and sub-seasonal dynamics and constrain the impact of short-term biogeochemical variability on carbon dioxide (CO₂) flux. Overall uncertainty of the MAPCO₂ using in situ calibrations with certified gas standards and post-deployment standard operating procedures is $< 2 \mu\text{atm}$ for seawater partial pressure of CO₂ ($p\text{CO}_2$) and $< 1 \mu\text{atm}$ for air $p\text{CO}_2$. The MAPCO₂ maintains this level of uncertainty for over 400 days of autonomous operation. MAPCO₂ measurements are consistent with ship-board seawater $p\text{CO}_2$ measurements and GLOBALVIEW-CO2 boundary layer atmospheric values. Here we provide an open ocean MAPCO₂ data set including over 100 000 individual air and seawater $p\text{CO}_2$ measurements on 14 surface buoys from 2004 through 2011 and a description of the methods and data quality control involved. The climate quality data provided by the MAPCO₂ has allowed for the establishment of open ocean observatories to track surface ocean $p\text{CO}_2$ changes around the globe. Data are available at doi:10.3334/CDIAC/OTG.TSM_NDP092 and cdiac.ornl.gov/oceans/Moorings/ndp092.

1 Introduction

The global ocean and its interactions with the atmosphere, climate, and marine ecosystem is undergoing a rapid and dramatic transition as it responds to multiple drivers on time scales from days to decades. Sustained observations guide our understanding of this ever-evolving earth system, which in turn, informs the development of solutions for

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human societies to cope with global change. The iconic Mauna Loa atmospheric carbon dioxide (CO_2) time series, or “Keeling Curve”, is an example of how observations gain importance with time, as they provide the basis for understanding future changes to the earth system in the context of current and historical observations (Keeling et al., 1976; Thoning et al., 1989; Hofmann et al., 2009). Similar “ocean observatories” must be sustained in order to track ocean carbon uptake and ocean acidification in the midst of the large natural temporal and spatial variability in the marine environment. These observations will provide a record of past and current behavior of the ocean carbon system and are central to predicting its future.

While high-quality ocean carbon measurements collected on global hydrographic surveys have been carried out approximately once a decade since the 1980s, the scientific community identified that constraining ocean biogeochemical models would require much greater temporal and spatial resolution of field data. Autonomous technology to measure surface ocean carbon was developed to address this need and has undergone rapid advancement in the last three decades (Takahashi, 1961; Weiss et al., 1982; Wanninkhof and Thoning, 1993; Feely et al., 1998; Pierrot et al., 2009). Autonomous underway systems that can measure the partial pressure of CO_2 ($p\text{CO}_2$) on ships were the first major breakthrough in our ability to collect high-frequency observations in the global ocean. These systems are designed to produce climate-quality data sets with measurements accurate to within $1 \mu\text{atm}$ for atmospheric CO_2 and $2 \mu\text{atm}$ for surface seawater $p\text{CO}_2$. This level of accuracy has allowed the scientific community to constrain regional sea–air CO_2 fluxes to 0.2 Pg C yr^{-1} .

While underway $p\text{CO}_2$ observations have greatly enhanced our understanding of the spatial variability in sea–air CO_2 fluxes (Takahashi et al., 2009; Wanninkhof et al., 2013), it has not solved the problem of quantifying temporal variability at a given location. Therefore, further observational improvements in these temporal dynamics are necessary, especially in highly variable regions such as the equatorial Pacific and coastal systems, in order to describe how short-term variability impacts CO_2 flux. Seawater $p\text{CO}_2$ observations that fully capture diurnal variations at a fixed site can be used

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to address this need for temporal resolution and to validate annual, seasonal, and sub-seasonal dynamics in ocean biogeochemical models. The Moored Autonomous $p\text{CO}_2$ (MAPCO₂) system was developed to address this need by autonomously measuring surface ocean $p\text{CO}_2$ and marine boundary layer (MBL) atmospheric CO_2 every 3 h on surface buoys at approximately the same level of accuracy as underway $p\text{CO}_2$ systems. With this recent development of mooring autonomous $p\text{CO}_2$ technology, the combination of all three monitoring approaches (i.e., hydrographic surveys, underway, and buoy measurements) has improved our understanding of the spatial and temporal variability of ocean carbon at the sea surface. For the first time, ocean $p\text{CO}_2$ observations from multiple platforms have been incorporated into the most recent update of the Surface Ocean CO_2 Atlas (SOCAT), a data synthesis effort aimed at bringing together all available CO_2 data in the surface ocean in a common format (Bakker et al., 2014).

Here we describe the methods, data quality control, and data access for an open ocean MAPCO₂ data set collected on 14 surface buoys from 2004 through 2011. These surface ocean $p\text{CO}_2$ observatories are critical for characterizing the natural variability of the ocean carbon cycle, contributing to our understanding of secular trends in ocean chemistry, validating and interpreting modeling results, and developing more sophisticated global carbon models.

2 Methods and data quality control

In 2004, the National Oceanographic and Atmospheric Administration's (NOAA) Pacific Marine Environmental Laboratory (PMEL) began to work with the Monterey Bay Aquarium Research Institute to improve the accuracy, reliability, and ease of use of an early moored $p\text{CO}_2$ system developed for buoys in the equatorial Pacific. Like the well-established underway $p\text{CO}_2$ method (Wanninkhof and Thoning, 1993; Feely et al., 1998; Pierrot et al., 2009), this early moored system described by Friederich et al. (1995) and the MAPCO₂ system described in Sect. 2.1 combines air-water equilibrators with an infrared (IR) analyzer for CO_2 gas detection. In 2009, the MAPCO₂

technology was transferred to Battelle Memorial Institute and is commercially available as the Sealogy[®] $p\text{CO}_2$ monitoring system. This system is now accessible to the larger scientific community and deployed at over 50 locations in open-ocean, coastal, and coral reef environments, including on NOAA's global moored CO_2 network (www.pmel.noaa.gov/co2/story/Buoys+and+Autonomous+Systems) and Australia's Integrated Marine Observing System (imos.org.au).

2.1 Description of MAPCO₂ system

The MAPCO₂ system includes four separate watertight cases that house the electronics, battery, transmitter, and a reference gas cylinder. The reference gases used on all the PMEL systems are traceable to World Meteorological Organization (WMO) standards and are provided by NOAA's Earth System Research Laboratory (ESRL). In the electronics case are the controls for the system, a memory flash card for data storage, a LI-COR LI-820 CO_2 gas analyzer, and a Sensirion SHT71 relative humidity and temperature sensor. The MAPCO₂ also includes an oxygen sensor for system diagnostic information. The LI-820 determines the CO_2 gas concentration by measuring the absorption of IR energy as a sample gas flows through an optical path. The CO_2 concentration is based on the difference ratio in the IR absorption between a reference and a sample optical path. The MAPCO₂ uses temperature and relative humidity to calculate the mole fraction of CO_2 ($x\text{CO}_2$) in air in equilibrium with surface seawater. The LI-820 is calibrated before every measurement using a "zero CO_2 reference" derived by scrubbing CO_2 from air using soda lime and an ESRL standard gas that spans the ocean $p\text{CO}_2$ values where the system is deployed. The system also includes a GPS for accurate position and time, an Iridium satellite communication link, an airblock deployed approximately 1 m above the ocean surface for atmospheric sampling, and an "h" shaped bubble equilibrator assembly described by Friederich et al. (1995). The equilibrator is the only part of the system in seawater and is made of copper-nickel alloy to prevent bio-fouling.

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After the equilibrator reading, a MBL air reading is made by drawing air in through the airblock, partially drying it, and passing it through the LI-820. Once the LI-820 path has been flushed, the flow is stopped and a 30 s average reading is collected. All measurements and calibrations are made at atmospheric pressure. The seawater CO_2 measurement occurs approximately 17 min after the start of the measurement cycle followed by the air CO_2 measurement two minutes later. Response time of the MAPCO₂ is dictated by the length of the full 20 min measurement cycle, and in fast mode, the MAPCO₂ system can measure air and seawater CO_2 once every 30 min.

Different types of sensors are used throughout the system for analytical, troubleshooting, and data quality control purposes. Additional parameters measured in each cycle (i.e., zero, span, equilibrator, and air) include temperature, pressure, relative humidity, and oxygen. Other sensors can also be integrated into the MAPCO₂ including CTD (Conductivity, Temperature, and Depth) instruments with auxiliary sensors attached (e.g., dissolved oxygen, fluorescence, turbidity) and pH sensors. The raw data collected by the MAPCO₂ and integrated sensors are stored on a memory flash card, and averaged data are telemetered from the buoy via the Iridium satellite communications system. This communications system also enables the user to control the MAPCO₂ remotely. The user can determine the sampling frequency and other variables, but the MAPCO₂ is nominally designed to make CO_2 measurements every 3 h with daily data transmissions for at least 400 days.

PMEL's MAPCO₂ systems have been deployed on open ocean buoys starting in 2004 with the establishment of NOAA's global moored CO_2 network and the efforts of numerous partners (see Acknowledgements). Figure 3 illustrates the locations, number of measurements, and average $\Delta p\text{CO}_2$ (sea–air) from the 14 surface CO_2 buoys included in this data set. These mooring $\Delta p\text{CO}_2$ observations are consistent with results of a synthesis of underway observations reported by Takahashi et al. (2009). Other than the Bermuda Testbed Mooring (BTM) and Japanese Kuroshio Extension Observatory (JKEO) time series, which have been discontinued, and the Multi-disciplinary Ocean Sensors for Environmental Analyses and Networks (MOSEAN) buoy, which was moved

approximately 20 km to the new Woods Hole Oceanographic Institution (WHOI) Hawaii Ocean Timeseries Station (WHOTS) location, the MAPCO₂ time series shown in Fig. 3 continue to be maintained. Seven of the 14 CO₂ buoys are located in the equatorial Pacific on the Tropical Atmosphere Ocean (TAO) array. Additional open ocean MAPCO₂ sites maintained by PMEL now exist in the North Atlantic, North Indian, and Southern oceans (see www.pmel.noaa.gov/co2/story/Buoys+and+Autonomous+Systems); however, they have been deployed since 2011 and are not included in the finalized data set presented here.

2.2 Data reduction and processing

Prior to deployment, each MAPCO₂ is tested in a seawater tank in the lab. PMEL maintains a MAPCO₂ system and a General Oceanics 8050 underway *p*CO₂ system that are permanently mounted for continuous sampling in the seawater tank. The standard MAPCO₂ is regularly compared to the underway system, which is calibrated every 8 h using four standard reference gases traceable to WMO standards. MAPCO₂ systems are compared to these standard systems in the lab and proceed to deployment if measuring within 3 μmol mol⁻¹ of the standard systems. Laboratory testing of the MAPCO₂ systems suggests instrument precision is < 0.6 μmol mol⁻¹ for *x*CO₂ values between 100 and 600 μmol mol⁻¹.

When the MAPCO₂ is recovered from the field, the system is compared against four ESRL standards to verify accuracy, and the raw data stored on the internal memory flash card are downloaded to a local database. Averaged raw data are then used for final processing of each data set. Final data quality control (QC) includes implementing a post calibration curve consistent with underway methods (Feely et al., 1998). Averaged *x*CO₂(wet) seawater and air measurements (defined in Table 1) are corrected based on the linear regression between raw averaged measurements and calibration coefficients. This process also corrects for the rare miscalibrations that occur during the deployment. Since the system is calibrated before each measurement, detector

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drift, if any, is negligible. This is confirmed by a mean difference between corrected and original raw data of $-0.02 \mu\text{mol mol}^{-1}$.

Data are quality controlled and flagged according to the SOCAT guidelines (Pfeil et al., 2013). For $p\text{CO}_2$ mooring purposes, we use three quality flags (QF): a flag value of 2 represents an acceptable measurement, 3 is a questionable measurement, and 4 is a bad measurement. A measurement can be questionable for a variety of reasons often revealed by MAPCO₂ system diagnostic information (e.g., low equilibrator pressure causing incomplete seawater equilibration), and the reasoning for each flag is included in the metadata QC log so the end user can decide whether or not to use questionable data. Prior to a data QC software update in June 2013, $x\text{CO}_2$ values flagged as bad (QF = 4) were still included in the published data sets, but after the software update, bad values are replaced with -999 . Other parameters published in the data sets that do not have an associated flag, such as sea surface temperature (SST) and sea surface salinity (SSS), are given a value of -999 or -9.999 when the measurement is missing or bad. As a final check of the data QC process, air $x\text{CO}_2(\text{dry})$ data are compared to MBL data from the GLOBALVIEW-CO₂ product (GLOBALVIEW-CO₂, 2013). In cases where air $x\text{CO}_2$ values from an entire MAPCO₂ deployment are offset from the GLOBALVIEW-CO₂ MBL time series and the surrounding MAPCO₂ deployments, a correction (typically $\leq 3 \mu\text{mol mol}^{-1}$) is applied. This correction is noted in the metadata and can be removed by the data user if desired.

Post-QC calculation of $p\text{CO}_2$ and $f\text{CO}_2$ (fugacity of CO₂) are made according to recommendations of the underway $p\text{CO}_2$ community (Pierrot et al., 2009). First, $x\text{CO}_2$ in dry air is calculated by

$$x\text{CO}_2(\text{dry}) = x\text{CO}_2(\text{wet}) \times \frac{P_{\text{Licor}}}{P_{\text{Licor}} - VP_{\text{Licor}}} \quad (1)$$

where $x\text{CO}_2(\text{wet})$ is the LI-820 measured concentration ($\mu\text{mol mol}^{-1}$), P_{Licor} is the pressure of the air and seawater samples measured in the LI-820 (kPa) and considered atmospheric pressure, and VP_{Licor} is the vapor pressure in the LI-820 (kPa). Since the

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$x\text{CO}_2(\text{wet})$ measurement is at equilibrium with sea surface humidity, relative humidity (RH) measurements of the air samples exiting the LI-820 are used to calculate VP_{Licor} in Eq. (1) using the following as defined by LI-COR for the IR analyzers:

$$\text{VP}_{\text{sat}} = (0.61121)(1.004)e^{\left(\frac{17.502 \times T_{\text{RH}}}{240.97 + T_{\text{RH}}}\right)} \quad (2)$$

$$\text{VP}_{\text{Licor}} = (\text{RH}_{\text{sample}} - \text{RH}_{\text{span}}) \times \frac{\text{VP}_{\text{sat}}}{100} \quad (3)$$

where VP_{sat} is the saturation vapor pressure of the RH sensor cell (kPa), T_{RH} is the temperature of the RH sensor ($^{\circ}\text{C}$), $\text{RH}_{\text{sample}}$ is the RH of the air sample (%), and RH_{span} is the RH of the span (%). We use an enhancement factor for moist air of 1.004 for 20°C and 1000 mbar in Eq. (2) as defined by Buck (1981).

Since the MAPCO_2 equilibration occurs directly in the ocean, it does not require the warming correction necessary for underway $p\text{CO}_2$ systems. Therefore, $p\text{CO}_2$ in wet air (100 % saturation) in equilibrium with the surface seawater is calculated by

$$p\text{CO}_2(\text{sat}) = x\text{CO}_2(\text{dry}) \times (P_{\text{Licor}} - p\text{H}_2\text{O}) \quad (4)$$

where P_{Licor} is atmospheric pressure for the air and surface seawater samples (atm) and $p\text{H}_2\text{O}$ is the water vapor pressure (atm) at equilibrator temperature as defined by Weiss and Price (1980). $f\text{CO}_2$ in wet air (100 % saturation) in equilibrium with the surface seawater is calculated by

$$f\text{CO}_2(\text{sat}) = p\text{CO}_2(\text{sat}) \times e^{\left[\frac{P_{\text{Licor}} \times (B_{11} + 2\delta_{12})}{R \times T}\right]} \quad (5)$$

where the ideal gas constant $R = 82.0578 \text{ cm}^3 \text{ atm mol}^{-1} \text{ K}^{-1}$, T is SST (K) from the CTD, and the B_{11} virial coefficient and δ_{12} cross virial coefficient for CO_2 as defined by Weiss (1974). The raw CO_2 data, temperature, salinity, and pressures are included in all published MAPCO_2 data sets so other data users can recalculate $x\text{CO}_2$, $f\text{CO}_2$, and $p\text{CO}_2$. Additional parameters included with the $p\text{CO}_2$ mooring data set are listed and described in Table 1.

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autonomous operation. However, since we measure RH and temperature of the gas stream exiting the LI-820, we can calculate $x\text{CO}_2(\text{dry})$ using Eqs. (1)–(3). The RH measurements used to calculate $x\text{CO}_2(\text{dry})$ have separate precisions and accuracies that can propagate through Eqs. (1)–(3) (Table 2). The total estimated precision and accuracy of $x\text{CO}_2(\text{dry})$ is calculated by summing each variable's precision and accuracy using the root sum of squares method. As presented in Table 2, propagation of all the errors from the separate variables does not cause the precision of calculated $x\text{CO}_2(\text{dry})$ to differ from measured $x\text{CO}_2(\text{wet})$ and results in a small impact to the accuracy ($0.1 \mu\text{mol mol}^{-1}$).

In addition to the propagation of error, an estimate of in situ accuracy is key to determining the overall uncertainty of the MAPCO₂ system. The GLOBALVIEW-CO₂ data product maintained by NOAA ESRL can be used as one validation data set for comparison to the MAPCO₂ air $x\text{CO}_2$ (dry) measurements (GLOBALVIEW-CO₂, 2013). Figure 4 shows 3 hourly atmospheric MAPCO₂ measurements and bi-weekly atmospheric CO₂ values from the MBL layer of GLOBALVIEW-CO₂ at the latitude closest to each MAPCO₂ location. Atmospheric MAPCO₂ data presented here are in the finalized, processed form as described in Sect. 2.2. Both MBL and MAPCO₂ data capture seasonal variability and long-term trends, but as expected, high-frequency MAPCO₂ measurements show short-term variability typically deviating from the smoothed MBL data product by $< 5 \mu\text{mol mol}^{-1}$ (Fig. 4). The Mauna Loa atmospheric CO₂ record is also shown in Fig. 4c and provides a reference for illustrating the larger seasonal variability in the lower atmosphere directly influenced by the presence of the ocean's surface. For the time series longer than 2 years, growth rates of the 3 hourly MAPCO₂ and bi-weekly MBL atmospheric CO₂ are presented in Table 3. Atmospheric CO₂ growth rates observed by five of the seven mooring time series differ from the MBL data by $\leq 0.1 \mu\text{mol mol}^{-1} \text{ yr}^{-1}$, suggesting that the finalized MAPCO₂ observations are consistent with other atmospheric observations collected using different methods. Further work to smooth and deseasonalize the data, as in most growth rate analyses, could

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persists for the next 8 days. In another example shown in Fig. 6c, the 15 μatm difference between the MAPCO₂ and underway system observed on 10 November 2008 is similar to the daily variability observed at the buoy in the 4 days prior to arrival of the *Ka'imimoana* and could reflect true differences observed by the underway and MAPCO₂ systems located 1–7 km apart. These examples highlight the difficulty of separating environmental variability and instrument uncertainty in these types of comparison exercises.

In order to minimize environmental variability while maximizing sample size for descriptive statistics, we use discrete measurements made within 10 km and 1.5 h and averaged underway $p\text{CO}_2$ measurements made within 10 km and 10 min of the MAPCO₂ system measurements for the seawater $p\text{CO}_2$ comparison analysis. While underway and MAPCO₂ systems utilize similar methodology, discrete $p\text{CO}_2$ presented in Table 4 is calculated from measurements of dissolved inorganic carbon (DIC) and total alkalinity (TA) using the program CO2SYS developed by Lewis and Wallace (1998) with the constants of Lueker et al. (2000). Typical error in calculated $p\text{CO}_2$ using this method is 5%. Only finalized seawater MAPCO₂ data are used for the descriptive statistics presented in Table 4. Unlike the descriptive statistics for the MAPCO₂ air comparisons, the statistics that result from using MAPCO₂ seawater measurements pre-MBL offset are not statistically different than the finalized, post-MBL offset statistics presented in Table 4. This could be due to the large natural variability in seawater $p\text{CO}_2$ compared to atmospheric CO₂.

Agreement between discrete and mooring surface ocean $p\text{CO}_2$ measurements is within 1.3 μatm (mean Δ in Table 4; BTM example in Fig. 5). Although more discrete measurements have been made at these and other mooring locations, this comparison is based on discrete samples restricted to within 10 km and 1.5 h of the MAPCO₂ system measurements with $n > 5$. Even with these restrictions, it is likely that environmental variability is not completely removed and is reflected in the mean Δ standard deviations of 3.7–6.2 μatm (Table 4). The small sample sizes (≤ 10 at each site) also resulting from these restrictions create large uncertainty in mean Δ values with standard

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error and confidence levels exceeding mean Δ values. This analysis shows promising results with a close agreement between discrete and MAPCO₂ measurements; however, more discrete samples will need to be collected within 10 km and 1.5 h of MAPCO₂ system measurements in order to improve the statistical significance of the seawater $p\text{CO}_2$ comparison.

Sample sizes are larger ($13 \leq n \leq 76$) for the comparison between underway and MAPCO₂ measurements at the BTM, TAO125W, and TAO140W locations. While underway measurements exist at other equatorial Pacific mooring locations, comparisons within 10 km and 10 min are restricted to TAO125W and TAO140W due to the large gaps in $p\text{CO}_2$ mooring data, the infrequent mooring servicing ship visits to each site (\sim once every 1–1.5 years), and the necessity for the mooring servicing ship to leave for the next station before the MAPCO₂ system has gone through a few cycles and measurements have stabilized. Even with these challenges, there are 76 comparison samples at BTM during the two buoy deployments in 2006–2007 (Fig. 5). These measurements show a mean Δ of $1.8 \pm 4.8 \mu\text{atm}$ with a low confidence level values of 1.1, indicating strong statistical significance ($p < 0.05$) that the actual mean Δ is between 0.7 and $2.9 \mu\text{atm}$ (Table 4). Standard deviations of the difference between the BTM vs. discrete (5.6) and underway measurements (4.8) are similar, which may be reflective of the environmental variability in this region of the surface ocean. Mean Δ in the equatorial Pacific is higher ($-3.3 \pm 15.2 \mu\text{atm}$ at TAO125W and $2.1 \pm 8.3 \mu\text{atm}$ at TAO140W), but statistical significance of these values is low due to the lower sample sizes and higher environmental variability in this region (Fig. 3). The largest standard deviation in mean Δ of 15.2 is at TAO125W, which is the site that exhibits the largest natural variability (i.e., total range of $\sim 200 \mu\text{atm}$, Fig. 6a) in surface seawater $p\text{CO}_2$ of the open ocean mooring data sets compared in Table 4.

The MAPCO₂ system has also been involved in two independent ocean $p\text{CO}_2$ instrument inter-comparisons. During an Alliance for Coastal Technologies (ACT) demonstration project, the difference between the MAPCO₂ system and an underway $p\text{CO}_2$ system was $-9 \pm 8 \mu\text{atm}$ in coastal Washington, USA waters and $-3 \pm 9 \mu\text{atm}$ in coral

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of weeks (e.g., Fig. 4 in Sutton et al., 2014). Sustained, long-term mooring time series also provide the opportunity to identify and remove the short-term variability from the time series and investigate long-term trends. For example, in a synthesis of equatorial Pacific mooring data, Sutton et al. (2014) found that the uptake of anthropogenic CO_2 and an acceleration in equatorial upwelling since the shift in the Pacific Decadal Oscillation in 1998 has led to high rates of $p\text{CO}_2$ change of +2.3 to +3.3 $\mu\text{atm yr}^{-1}$ in this region. This decadal shift in CO_2 outgassing is consistent with underway $p\text{CO}_2$ observations made in this region since 1982 (Feely et al., 2014).

Mooring data from most of the deployments through 2010 listed in Table 5 are also included in the most recent version of SOCAT (Bakker et al., 2014). This SOCATv2.0 synthesis involves a standardized, second level quality control of 10.1 million surface seawater $f\text{CO}_2$ measurements from many different sources, including underway and mooring systems. SOCAT also produces a gridded surface ocean $f\text{CO}_2$ data product in a uniform format available at www.socat.info. Rödenbeck et al. (2013) compared the previous version of SOCAT (v1.5), which did not include mooring data, to some of the open ocean MAP CO_2 time series in Table 5. In a comparison between seawater $p\text{CO}_2$ data from the TAO170W MAP CO_2 and model estimates based on SOCATv1.5, Rödenbeck et al. (2013) find that the model estimates in the tropics are unrelated to or even opposite of the mooring observations. While the Rödenbeck et al. (2013) model captures high-frequency $p\text{CO}_2$ variations related to thermodynamic effects, it does not reproduce high $p\text{CO}_2$ upwelling events. We expect the recent mooring additions to SOCATv2.0 and the open ocean MAP CO_2 data set presented here to make a large impact on our efforts to model and understand the global carbon cycle in the coming years.

4 Conclusion

Mooring observations can play a critical role in improving our ability to model, understand, and describe the ocean carbon cycle on all time scales. In particular, time series

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Akl, J., Barbero, L., Bates, N. R., Boutin, J., Bozec, Y., Cai, W.-J., Castle, R. D., Chavez, F. P., Chen, L., Chierici, M., Currie, K., de Baar, H. J. W., Evans, W., Feely, R. A., Fransson, A., Gao, Z., Hales, B., Hardman-Mountford, N. J., Hoppema, M., Huang, W.-J., Hunt, C. W., Huss, B., Ichikawa, T., Johannessen, T., Jones, E. M., Jones, S. D., Jutterström, S., Kitidis, V., Körtzinger, A., Landschützer, P., Lauvset, S. K., Lefèvre, N., Manke, A. B., Mathis, J. T., Merlivat, L., Metzl, N., Murata, A., Newberger, T., Omar, A. M., Ono, T., Park, G.-H., Pater-
 5 son, K., Pierrot, D., Ríos, A. F., Sabine, C. L., Saito, S., Salisbury, J., Sarma, V. V. S. S., Schlitzer, R., Sieger, R., Skjelvan, I., Steinhoff, T., Sullivan, K. F., Sun, H., Sutton, A. J., Suzuki, T., Sweeney, C., Takahashi, T., Tjiputra, J., Tsurushima, N., van Heuven, S. M. A. C., Vandemark, D., Vlahos, P., Wallace, D. W. R., Wanninkhof, R., and Watson, A. J.: An up-
 10 date to the Surface Ocean CO_2 Atlas (SOCAT version 2), *Earth Syst. Sci. Data*, 6, 69–90, doi:10.5194/essd-6-69-2014, 2014.

Buck, A. L.: New equations for computing vapor pressure and enhancement factor, *J. Appl. Meteorol.*, 20, 1527–1532, 1981.

15 Feely, R. A., Wanninkhof, R., Milburn, H. B., Cosca, C. E., Stapp, M., and P. Murphy, P.: A new automated underway system for making high precision $p\text{CO}_2$ measurements onboard re- search ships, *Anal. Chim. Acta*, 377, 185–191, doi:10.1016/S0003-2670(98)00388-2, 1998.

Feely, R. A., Cosca, C. E., Sutton, A. J., Sabine, C. L., Wanninkhof, R., and Mathis, J. T.: Decadal changes of air–sea CO_2 fluxes in the Equatorial Pacific Ocean, *Geophys. Res. Lett.*, in preparation, 2014.

20 Friederich, G. E., Brewer, P. G., Herlien, R., and Chavez, F. P.: Measurement of sea sur- face partial pressure of CO_2 from a moored buoy, *Deep-Sea Res. Pt. I*, 42, 1175–1186, doi:10.1016/0967-0637(95)00044-7, 1995.

GLOBALVIEW-CO2 Cooperative Global Atmospheric Data Integration Project: Multi-laboratory compilation of synchronized and gap-filled atmospheric carbon dioxide records for the period 1979–2012 (obspack_co2_1_GLOBALVIEW-CO2_2013_v1.0.4_2013-12-23), Com-
 25 piled by NOAA Global Monitoring Division: Boulder, Colorado, USA, available at: doi:10.3334/OBSPACK/1002, 2013 (updated annually).

Hofmann, D. J., Butler, J. H., and Tans, P. P.: A new look at atmospheric carbon dioxide, *Atmos. Environ.*, 43, 2084–2086, doi:10.1016/j.atmosenv.2008.12.028, 2009.

30 Keeling, C. D., Bacastow, R. B., Bainbridge, A. E., Ekdahl, C. A., Guenther, P. R., Water- man, L. S., and Chin, J. F. S.: Atmospheric carbon dioxide variations at Mauna Loa Ob- servatory, Hawaii, *Tellus*, 28, 538–551, doi:10.1111/j.2153-3490.1976.tb00701.x, 1976.

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- Lewis, E. and Wallace, D. W. R.: Program Developed for CO_2 System Calculations, Carbon Dioxide Information Analysis Center, Oak Ridge National Laboratory, US Department of Energy, Oak Ridge, Tennessee, 1998.
- Lueker, T. J., Dickson, A. G., and Keeling, C. D.: Ocean $p\text{CO}_2$ calculated from dissolved inorganic carbon, alkalinity, and equations for K1 and K2: validation based on laboratory measurements of CO_2 in gas and seawater at equilibrium, *Mar. Chem.*, 70, 105–119, doi:10.1016/S0304-4203(00)00022-0, 2000.
- Pfeil, B., Olsen, A., Bakker, D. C. E., Hankin, S., Koyuk, H., Kozyr, A., Malczyk, J., Manke, A., Metzl, N., Sabine, C. L., Akl, J., Alin, S. R., Bates, N., Bellerby, R. G. J., Borges, A., Boutin, J., Brown, P. J., Cai, W.-J., Chavez, F. P., Chen, A., Cosca, C., Fassbender, A. J., Feely, R. A., González-Dávila, M., Goyet, C., Hales, B., Hardman-Mountford, N., Heinze, C., Hood, M., Hoppema, M., Hunt, C. W., Hydes, D., Ishii, M., Johannessen, T., Jones, S. D., Key, R. M., Körtzinger, A., Landschützer, P., Lauvset, S. K., Lefèvre, N., Lenton, A., Lourantou, A., Merlivat, L., Midorikawa, T., Mintrop, L., Miyazaki, C., Murata, A., Nakadate, A., Nakano, Y., Nakaoka, S., Nojiri, Y., Omar, A. M., Padin, X. A., Park, G.-H., Paterson, K., Perez, F. F., Pierrot, D., Poisson, A., Ríos, A. F., Santana-Casiano, J. M., Salisbury, J., Sarma, V. V. S. S., Schlitzer, R., Schneider, B., Schuster, U., Sieger, R., Skjelvan, I., Steinhoff, T., Suzuki, T., Takahashi, T., Tedesco, K., Telszewski, M., Thomas, H., Tilbrook, B., Tjiputra, J., Vandemark, D., Veness, T., Wanninkhof, R., Watson, A. J., Weiss, R., Wong, C. S., and Yoshikawa-Inoue, H.: A uniform, quality controlled Surface Ocean CO_2 Atlas (SOCAT), *Earth Syst. Sci. Data*, 5, 125–143, doi:10.5194/essd-5-125-2013, 2013.
- Pierrot, D., Neill, C., Sullivan, K., Castle, R., Wanninkhof, R., Lüger, H., Johannessen, T., Olsen, A., Feely, R. A., and Cosca, C. E.: Recommendations for autonomous underway $p\text{CO}_2$ measuring systems and data-reduction routines, *Deep-Sea Res. Pt. II*, 56, 512–522, doi:10.1016/j.dsr2.2008.12.005, 2009.
- Rödenbeck, C., Keeling, R. F., Bakker, D. C. E., Metzl, N., Olsen, A., Sabine, C., and Heimann, M.: Global surface-ocean $p\text{CO}_2$ and sea-air CO_2 flux variability from an observation-driven ocean mixed-layer scheme, *Ocean Sci.*, 9, 193–216, doi:10.5194/os-9-193-2013, 2013.
- Schar, D., Atkinson, M., Johengen, T., Pinchuk, A., Purcell, H., Robertson, C., Smith, G. J., and Tamburri, M.: Performance demonstration statement PMEL MAP CO_2 /Battelle Seology $p\text{CO}_2$ Monitoring System, UMCES Technical Report Series: Ref. No. [UMCES]CBL 10-092, Solomons, MD, USA 2010.

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- Sutton, A. J., Feely, R. A., Sabine, C. L., McPhaden, M. J., Takahashi, T., Chavez, F. P., Friederich, G. E., and Mathis, J. T.: Natural variability and anthropogenic change in equatorial Pacific surface ocean $p\text{CO}_2$ and pH, *Global Biogeochem. Cy.*, 28, GB4679, doi:10.1002/2013GB004679, 2014.
- 5 Takahashi, T.: Carbon dioxide in the atmosphere and in Atlantic Ocean water, *J. Geophys. Res.*, 66, 477–494, 1961.
- Takahashi, T., Sutherland, S. C., Wanninkhof, R., Sweeney, C., Feely, R. A., Chipman, D. W., Hales, B., Friederich, G., Chavez, F., Sabine, C., Watson, A., Bakker, D. C. E., Schuster, U., Metzl, N., Yoshikawa-Inoue, H., Ishii, M., Midorikawa, T., Nojiri, Y., Körtzinger, A., Steinhoff, T., Hoppema, M., Olafsson, J., Arnarson, T. S., Tilbrook, B., Johannessen, T., Olsen, A., Bellerby, R., Wong, C. S., Delille, B., Bates, N. R., and de Baar, H. J. W.: Climatological mean and decadal change in surface ocean $p\text{CO}_2$, and net sea–air CO_2 flux over the global oceans, *Deep-Sea Res. Pt. II*, 56, 554–577, doi:10.1016/j.dsr2.2008.12.009, 2009.
- 10 Thoning, K. W., Tans, P. P., and Komhyr, W. D.: Atmospheric carbon dioxide at Mauna Loa Observatory: 2. Analysis of the NOAA GMCC data, 1974–1985, *J. Geophys. Res.-Atmos.*, 94, 8549–8565, doi:10.1029/JD094iD06p08549, 1989.
- UNESCO: SOCAT Equatorial Pacific, North Pacific, and Indian Ocean Regional Workshop, Tokyo, Japan, 8–11 February 2010, IOC Workshop Report No. 229, 2010.
- Wanninkhof, R. and Thoning, K.: Measurement of fugacity of carbon dioxide in surface water and air using continuous sampling methods, *Mar. Chem.*, 44, 189–205, 1993.
- 20 Wanninkhof, R., Park, G.-H., Takahashi, T., Sweeney, C., Feely, R., Nojiri, Y., Gruber, N., Doney, S. C., McKinley, G. A., Lenton, A., Le Quéré, C., Heinze, C., Schwinger, J., Graven, H., and Khatiwala, S.: Global ocean carbon uptake: magnitude, variability and trends, *Biogeochemistry*, 10, 1983–2000, doi:10.5194/bg-10-1983-2013, 2013.
- 25 Weiss, R. F.: Carbon dioxide in water and seawater: the solubility of a non-ideal gas, *Mar. Chem.*, 2, 203–215, doi:10.1016/0304-4203(74)90015-2, 1974.
- Weiss, R. F. and Price, B. A.: Nitrous oxide solubility in water and seawater, *Mar. Chem.*, 8, 347–359, doi:10.1016/0304-4203(80)90024-9, 1980.
- Weiss, R. F., Jahnke, R. A., and Keeling, C. D.: Seasonal effects of temperature and salinity on the partial pressure of CO_2 in seawater, *Nature*, 300, 511–513, 1982.
- 30

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Table 1. Final data variable names and descriptions.

Variable name	Description	Units	Equation (if applicable)
Mooring	mooring name as shown in Fig. 3	character string	
Latitude	average latitude during deployment	decimal degrees	
Longitude	average longitude during deployment	decimal degrees	
Date	date of measurement in UTC	MM/DD/YYYY	
Time	time of measurement in UTC	HH:MM	
xCO ₂ _SW_wet	mole fraction of carbon dioxide in air in equilibrium with surface seawater at SST and humidity	$\mu\text{mol mol}^{-1}$	
xCO ₂ _SW_QF	primary flag associated with seawater xCO ₂ measurement	SOCAT standards ^a	
H ₂ O_SW	mole fraction of water in gas from equilibrator	$\mu\text{mol mol}^{-1}$	
xCO ₂ _Air_wet	mole fraction of carbon dioxide in air at ~ 1.5 m above the sea surface at sample humidity	$\mu\text{mol mol}^{-1}$	
xCO ₂ _SW_QF	primary flag associated with air xCO ₂ measurement	SOCAT standards ^a	
H ₂ O_Air	mol fraction of water in air	$\mu\text{mol mol}^{-1}$	
Licor_Atm_Pressure	Atmospheric pressure at ~ 1.5 m above the sea surface	hPa	
Licor_Temp	Licor temperature	°C	
Percent_O ₂ ^b	% oxygen in surface seawater divided by % oxygen in air at ~ 1.5 m above the sea surface	%	
SST ^c	Sea surface temperature	°C	
SSS ^c	Sea surface salinity		
xCO ₂ _SW_dry	mole fraction of carbon dioxide in dry air in equilibrium with surface seawater	$\mu\text{mol mol}^{-1}$	1
xCO ₂ _Air_dry	mole fraction of carbon dioxide in dry air at ~ 1.5 m above the sea surface	$\mu\text{mol mol}^{-1}$	1
fCO ₂ _SW_sat	fugacity of carbon dioxide in wet air (100 % humidity) in equilibrium with surface seawater	μatm	5
fCO ₂ _Air_sat	fugacity of carbon dioxide in wet air (100 % humidity) at ~ 1.5 m above the sea surface	μatm	5
dFCO ₂	fCO ₂ _SW_sat – fCO ₂ _Air_sat	μatm	
pCO ₂ _SW_sat ^d	partial pressure of carbon dioxide in wet air (100 % humidity) in equilibrium with surface seawater	μatm	4
pCO ₂ _Air_sat ^d	partial pressure of carbon dioxide in wet air (100 % humidity) at ~ 1.5 m above the sea surface	μatm	4
dpCO ₂ ^d	pCO ₂ _SW_sat – pCO ₂ _Air_sat	μatm	

Notes:

^a SOCAT flags used in this data set: 2 = acceptable measurement, 3 = questionable measurement, 4 = bad measurement (note: bad data values are reported in the final data file submitted to CDIAC prior to QC software upgrade in Jun 2013 but reported as –999 in files submitted after the upgrade).

^b Oxygen measured in the MAPCO₂ system is exposed to air and likely modified within the system prior to measurement. Rapid changes in oxygen are not properly captured using this method. This data should not be used as a quantitative measure of oxygen.

^c Usually measured by other academic partners at each site. See metadata for each deployment for details on SST and SSS measurements.

^d pCO₂ only presented in data sets submitted to CDIAC after Jun 2013 when QC software was upgraded to include this calculation. Data users of earlier data sets can calculate pCO₂ as defined in Eq. (4).

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Table 2. Sources of error for the calculation of $x\text{CO}_2(\text{dry})$ at atmospheric pressure = 101 kPa, $\text{RH}_{\text{sample}} = 75\%$, $\text{RH}_{\text{span}} = 30\%$, $\text{SST} = 25^\circ\text{C}$, $\text{SSS} = 35$, and $x\text{CO}_2(\text{wet}) = 375 \mu\text{mol mol}^{-1}$. Total estimated precision and accuracy is calculated using the root sum of squares method (RSS): $\text{RSS} = (\sum a^2)^{1/2}$.

Sources of error	Variable precision (\pm)	Effect on precision of final calculation (a)	Variable accuracy (\pm)	Effect on accuracy of final calculation (a)
VP_{Licor} calculation				
VP _{sat} (kPa)	0.019 ^a	0.009	0.057 ^a	0.026
RH _{sample}	0.1 ^b	negligible	3.0 ^b	0.1
RH _{span}	0.1 ^b	negligible	3.0 ^b	0.1
assumption that VP _{RH} = VP _{Licor}				0.052
Total estimated error: VP _{Licor}		0.009		0.153
$x\text{CO}_2(\text{dry})$ calculation				
$x\text{CO}_2(\text{wet})$ ($\mu\text{mol mol}^{-1}$)	0.7 ^c	0.7	1.5 ^c	1.5
P_{Licor} (kPa)	0.001 ^b	negligible	0.010 ^b	negligible
VP _{Licor} (kPa) (calculated above)	0.009	0.034	0.153	0.585
Total estimated error: $x\text{CO}_2(\text{dry})$		0.7		1.6

Notes:

^a Error calculated using manufacturer estimated error for T_{RH} of $\pm 0.1^\circ\text{C}$ precision and $\pm 0.3^\circ\text{C}$ accuracy (see Eq. 2).^b Error reported by manufacturer.^c Precision estimate based on standard deviation of the high-frequency raw data (~ 57 repeated measurements over 1 min) in the field; accuracy estimate based on pre-deployment testing in the laboratory.

Negligible indicates value < significant digits of variable.

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Table 3. Growth rate of GLOBALVIEW-CO₂ MBL and MAPCO₂ atmospheric $x\text{CO}_2$ time series (GLOBALVIEW-CO₂, 2013). For mooring time series locations see Fig. 3.

Time series > 2 years	Growth rate ($\mu\text{mol mol}^{-1} \text{yr}^{-1}$)	
	MBL	MAPCO ₂
Papa	3.1	3.1
KEO	1.7	1.1
MOSEAN/WHOTS	2.0	1.6
TAO125W	1.9	2.0
TAO140W	1.9	1.9
TAO170W	1.9	1.9
Stratus	1.8	1.9

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Table 4. Descriptive statistics of Δ (MAPCO₂ measurement – comparison measurement). The closest MAPCO₂ value in time and space is compared to: biweekly GLOBALVIEW-CO2 MBL values from the latitude nearest to average buoy location, single discrete measurements made within 10 km and 1.5 h, and averaged underway $p\text{CO}_2$ measurements made within 10 km and 10 min of the MAPCO₂ system measurement. Standard error is the standard error of the mean and confidence level values illustrate the 95 % probability that the actual population mean = sample mean \pm confidence level.

	<i>n</i>	Mean	Standard Error	Standard Deviation	Confidence Level (95 %)
MAPCO ₂ air $x\text{CO}_2$ (dry) comparison to MBL air ($\mu\text{mol mol}^{-1}$)					
pre-MBL ^a offset (estimate of MAPCO ₂ system in situ accuracy)	1823	−1.5	0.1	2.4	0.1
post-MBL ^a offset (estimate of finalized MAPCO ₂ data accuracy)	1823	−0.3	< 0.1	1.7	0.1
MAPCO ₂ seawater $p\text{CO}_2$ comparison to calculated $p\text{CO}_2$ (μatm) from discrete DIC, TA					
WHOTS vs. HOTS ^b	7	0.1	1.4	3.7	3.4
BTM vs. BATS ^c	9	1.3	1.9	5.6	4.3
Papa vs. Station P ^d	10	−0.4	2.0	6.2	4.5
MAPCO ₂ seawater $p\text{CO}_2$ comparison to underway $p\text{CO}_2$ (μatm)					
BTM vs. <i>Atlantic Explorer</i> ^e	76	1.8	0.5	4.8	1.1
TAO125W vs. <i>Ka'imimoana</i> ^f	16	−3.3	3.8	15.2	8.1
TAO140W vs. <i>Ka'imimoana</i> ^f	13	2.1	2.3	8.3	5.0

Notes on data sources and archives:

^a GLOBALVIEW-CO2 Marine Boundary Layer (MBL) data source: NOAA Earth System Research Laboratory, www.esrl.noaa.gov/gmd/ccgg/globalview/co2/co2_intro.html (GLOBALVIEW-CO2, 2013).

^b Hawaii Ocean Time-Series (HOTS) data source: University of Hawaii, hahana.soest.hawaii.edu/hot.

^c Bermuda Atlantic Time-series Study (BATS) data source: Bermuda Institute of Ocean Sciences, bats.bios.edu.

^d Station P data source: University of Washington and NOAA PMEL.

^e Atlantic Explorer data source: Bermuda Institute of Ocean Sciences, cdiac.ornl.gov/oceans/CARINA/.

^f *Ka'imimoana* data source: NOAA PMEL, cdiac.ornl.gov/oceans/VOS_Program/kaimimoana.html.

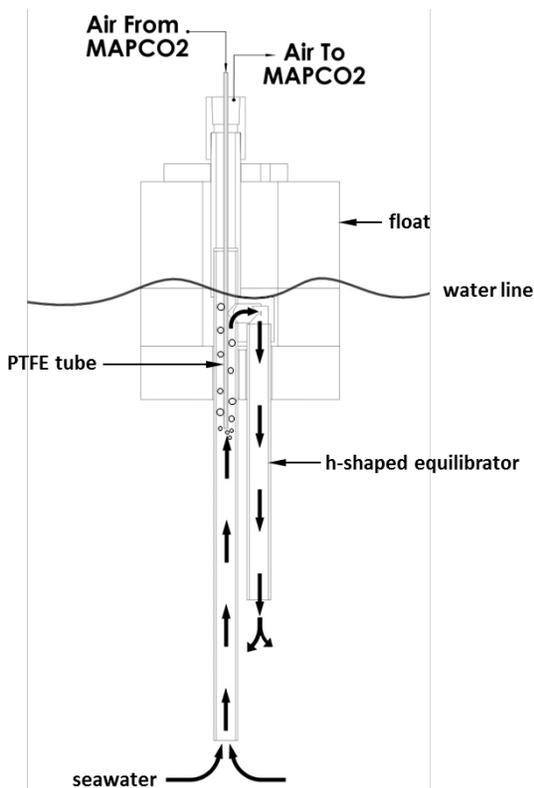


Figure 2. Schematic diagram of the floating air-water equilibrator assembly in the MAPCO₂ system during the seawater equilibration cycle. Air is pumped from the MAPCO₂ through a PTFE tube and bubbled into the equilibrator. As the bubbles rise through the water, the air comes into equilibrium with the dissolved gases in the surface seawater. The rising air bubbles in the equilibrator also create circulation by pushing water up and over the horizontal leg of the h-shaped equilibrator and out the short leg of the equilibrator. Image is not to scale.

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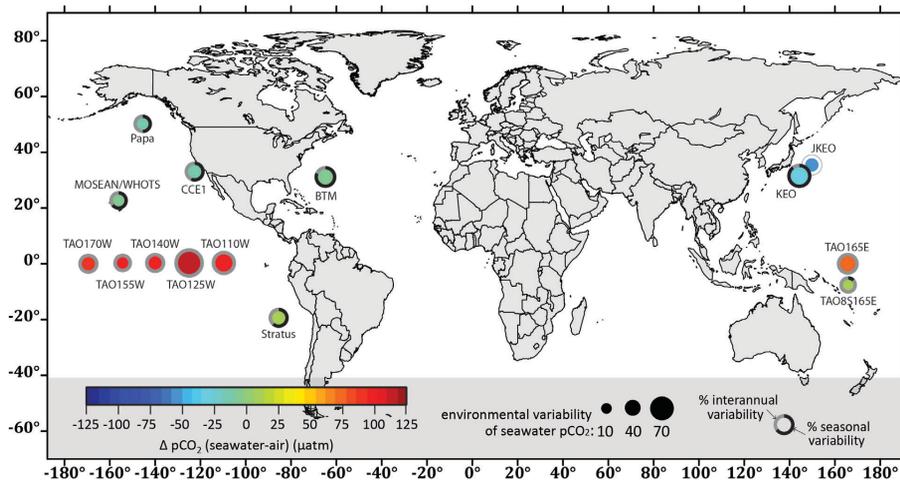


Figure 3. Location of open ocean moorings in the MAPCO₂ data set. Inner circle color illustrates the mean $\Delta p\text{CO}_2$ of the finalized data at that location. Inner circle size is relative to the environmental variability in the time series defined here as the standard deviation of seawater $p\text{CO}_2$ values. The outer ring shows the proportion of environmental variability in seawater $p\text{CO}_2$ due to the seasonal cycle (black) and interannual variability (gray). Seasonal variability is defined as the mean seasonal peak amplitude and interannual variability is the mean Δ of annual mean values. Seasonal and interannual variability cannot be quantified at JKEO with a time series of < 1 year and is represented here by an outer ring with no color.

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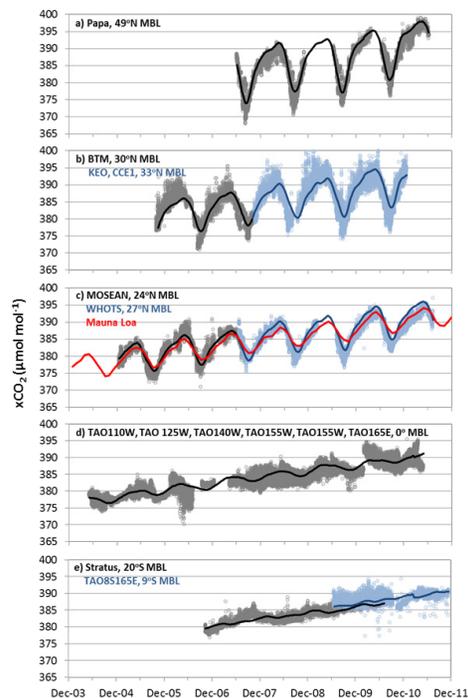


Figure 4. MAPCO₂ and GLOBALVIEW-CO₂ MBL atmospheric $x\text{CO}_2$ ($\mu\text{mol mol}^{-1}$) presented by latitude: **(a)** Papa MAPCO₂ (gray points) and MBL at 49° N (black line); **(b)** BTM MAPCO₂ (gray points), MBL at 30° N (black line), KEO and CCE1 MAPCO₂ (blue points), and MBL at 33° N (blue line); **(c)** MOSEAN MAPCO₂ (gray points), MBL at 24° N (black line), WHOTS MAPCO₂ (blue points), MBL at 27° N (blue line), and Mauna Loa Observatory atmospheric $x\text{CO}_2$ (red line); **(d)** six equatorial MAPCO₂ buoys (gray points) and MBL at 0° (black line); and **(e)** Stratus MAPCO₂ (gray points), MBL at 20° S (black line), TAO8S165E MAPCO₂ (blue points), and MBL at 9° S (blue line). MBL data from GLOBALVIEW-CO₂ (2013). Mauna Loa observatory monthly mean data from Pieter Tans, NOAA/ESRL (www.esrl.noaa.gov/gmd/ccgg/trends/) and Ralph Keeling, Scripps Institution of Oceanography (scrippsco2.ucsd.edu/).

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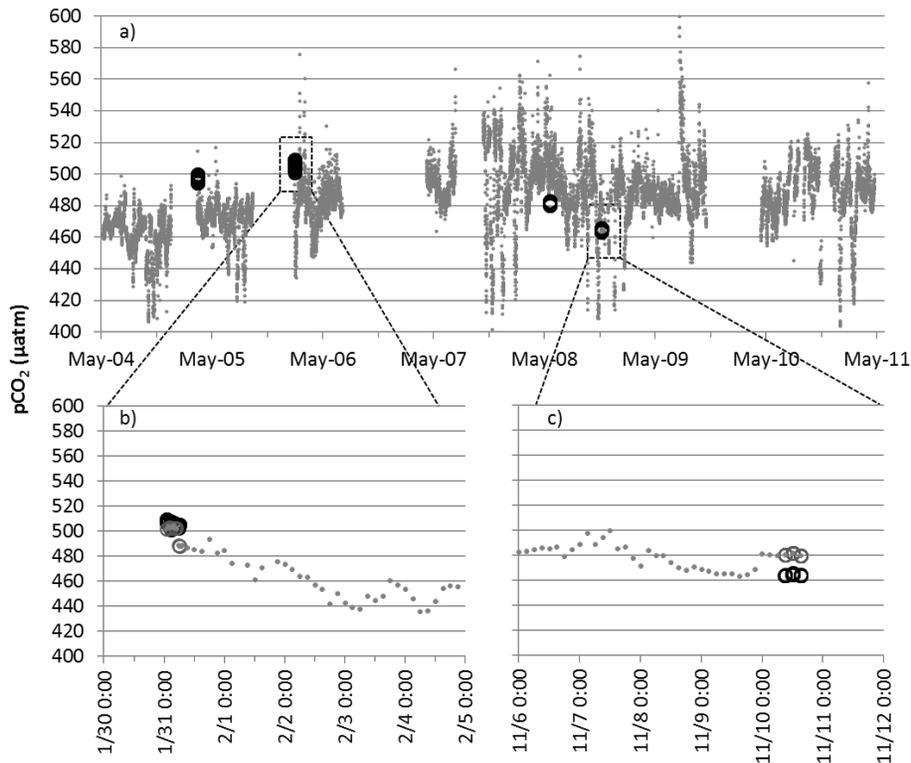


Figure 6. (a) TAO125W surface seawater MAPCO₂ observations (gray points) for the entire time series at this location with average R/V *Ka'imimoana* underway $p\text{CO}_2$ data within 10 km and 10 min of the MAPCO₂ measurements (black open circles). Two examples of comparison data over one week time series are shown in panels (b) and (c) with MAPCO₂ measurements corresponding to the average underway observations illustrated in gray open circles. Selection boxes in (a) are not to scale of actual axes in (b) and (c) panels. *Ka'imimoana* data from NOAA PMEL, cdiac.ornl.gov/oceans/VOS_Program/kaimimoana.html.